

Umstellung von Formeln

1. Stelle die folgenden Formeln um.

- a) $F = ma$ $a =$
 b) $pV = NRT$ $T =$
 c) $qU = h \cdot \frac{c}{\lambda_{\min}}$ $\lambda_{\min} =$
 d) $E_{\text{pot}} = mg\Delta h$ $\Delta h =$
 e) $F_e = k_e \cdot \frac{q_1 \cdot q_2}{r^2}$ $q_2 =$
 f) $P_{\text{total}} = c_{\text{Rtg}} \cdot U^2 I Z$ $I =$
 g) $\frac{1}{2}mv^2 = \frac{3}{2}kT$ $T =$
 h) $\frac{\Delta V}{\Delta t} = -\frac{\pi \cdot r^4}{8 \cdot \eta} \cdot \frac{\Delta p}{\Delta l}$ $\Delta p =$
 $\eta =$
-

2. Stelle die folgenden Formeln um.

- a) $M = \sigma \cdot T^4$ $T =$
 b) $E_{\text{kin}} = \frac{1}{2}mv^2$ $v =$
 c) $\Delta E = \sigma \cdot (T_1^4 - T_2^4) \cdot A \cdot t$ $T_1 =$
 $T_2 =$
 d) $P_{\text{total}} = c_{\text{Rtg}} \cdot U^2 I Z$ $U =$
 e) $\frac{1}{2}mv^2 = \frac{3}{2}kT$ $v =$
 f) $mg\Delta h = \frac{1}{2}mv^2$ $v =$
 g) $F_e = k_e \cdot \frac{q_1 \cdot q_2}{r^2}$ $r =$
 h) $\frac{\Delta V}{\Delta t} = -\frac{\pi \cdot r^4}{8 \cdot \eta} \cdot \frac{\Delta p}{\Delta l}$ $r =$
 i) $f_0 = \frac{1}{2\pi} \cdot \sqrt{\frac{D}{m}}$ $D =$
 $m =$
 j) $f_0 = \frac{1}{2\pi} \cdot \sqrt{\frac{g}{l}}$ $l =$

k) $f_0 = \frac{1}{2\pi} \cdot \sqrt{\frac{1}{LC}}$ $L =$
 $C =$

3. Stelle die folgenden Formeln um.

a) $\delta = \frac{\lambda_v}{n \cdot \sin(\omega)}$ $\sin(\omega) =$

$\omega =$

b) $\frac{\sin(\alpha_1)}{\sin(\alpha_2)} = \frac{c_1}{c_2}$ $\sin(\alpha_2) =$

$\alpha_2 =$

c) $y = y_{\max} \cdot \sin(2\pi f \cdot t + \varphi_0)$ $f =$

4. Stelle die folgenden Formeln um.

a) $J = J_0 \cdot e^{-\mu x}$ $x =$

$\mu =$

b) $J = J_0 \cdot e^{-\frac{\ln(2)}{D}x}$ $x =$

$D =$

c) $U = U_0 \cdot e^{-t/(RC)}$ $t =$

$C =$

d) $n_i = n_0 \cdot e^{-\frac{Mg\Delta h}{RT}}$ $\Delta h =$

$T =$

e) $\Lambda = \Lambda_0 \cdot 2^{-\frac{t}{T}}$ $t =$

$T =$

f) $U = U_{\max} \cdot (1 - e^{-t/(RC)})$ $t =$

$R =$

5. Stelle die folgenden Formeln um.

a) $\text{pH} = -\log \left[\frac{\text{H}^+}{1\text{M}} \right]$ $[\text{H}^+] =$

b) $E = \log \left(\frac{J_0}{J} \right)$ $J =$

c) $S = k_B \cdot \ln(\Omega)$ $\Omega =$

d) $n_{dB} = 10 \cdot \log \left(\frac{J}{J_0} \right)$ $J =$

$$e) \quad \text{pH} = \text{p}K + \log \left(\frac{[\text{Basis}]}{[\text{Säure}]} \right) \quad [\text{Basis}] =$$

$$[\text{Säure}] =$$

$$f) \quad \text{pH} = \text{p}K + \log \left(\frac{[\text{Basis}]}{c - [\text{Basis}]} \right) \quad [\text{Basis}] =$$

$$g) \quad E = E^0 + \frac{RT}{z_e F} \cdot \ln \left(\frac{c_{\text{ox}}}{c_{\text{red}}} \right) \quad c_{\text{ox}} =$$

$$c_{\text{red}} =$$

$$h) \quad E = E^0 + \frac{0,059 \text{ V}}{z_e} \cdot \log \left(\frac{c_{\text{ox}}}{c_{\text{red}}} \right) \quad c_{\text{ox}} =$$

$$c_{\text{red}} =$$

Lösungen

1.

a) $F = ma$ $a = \frac{F}{m}$

b) $pV = NRT$ $T = \frac{pV}{NR}$

c) $qU = h \cdot \frac{c}{\lambda_{\min}}$ $\lambda_{\min} = \frac{hc}{qU}$

d) $E_{\text{pot}} = mg\Delta h$ $\Delta h = \frac{E_{\text{pot}}}{mg}$

e) $F_e = k_e \cdot \frac{q_1 \cdot q_2}{r^2}$ $q_2 = \frac{F_e \cdot r^2}{k_e \cdot q_1}$

f) $P_{\text{total}} = c_{\text{Rtg}} \cdot U^2 I Z$ $I = \frac{P_{\text{total}}}{c_{\text{Rtg}} \cdot U^2 \cdot Z}$

g) $\frac{1}{2}mv^2 = \frac{3}{2}kT$ $T = \frac{mv^2}{3k}$

h) $\frac{\Delta V}{\Delta t} = -\frac{\pi \cdot r^4}{8 \cdot \eta} \cdot \frac{\Delta p}{\Delta l}$ $\Delta p = -\frac{\Delta V \cdot 8 \cdot \eta \cdot \Delta l}{\Delta t \cdot \pi \cdot r^4}$

$$\eta = -\frac{\pi \cdot r^4 \cdot \Delta p \cdot \Delta t}{8 \cdot \Delta l \cdot \Delta V}$$

2.

a) $M = \sigma \cdot T^4$ $T = \sqrt[4]{\frac{M}{\sigma}}$

b) $E_{\text{kin}} = \frac{1}{2}mv^2$ $v = \sqrt{\frac{2 \cdot E_{\text{kin}}}{m}}$

c) $\Delta E = \sigma \cdot (T_1^4 - T_2^4) \cdot A \cdot t$ $T_1 = \sqrt[4]{\frac{\Delta E}{\sigma \cdot A \cdot t} + T_2^4}$

$$T_2 = \sqrt[4]{T_1^4 - \frac{\Delta E}{\sigma \cdot A \cdot t}}$$

d) $P_{\text{total}} = c_{\text{Rtg}} \cdot U^2 I Z$ $U = \sqrt{\frac{P_{\text{total}}}{c_{\text{Rtg}} \cdot I \cdot Z}}$

e) $\frac{1}{2}mv^2 = \frac{3}{2}kT$ $v = \sqrt{\frac{3kT}{m}}$

f) $mg\Delta h = \frac{1}{2}mv^2$ $v = \sqrt{2g\Delta h}$

g) $F_e = k_e \cdot \frac{q_1 \cdot q_2}{r^2}$ $r = \sqrt{\frac{k_e \cdot q_1 \cdot q_2}{F_e}}$

$$h) \quad \frac{\Delta V}{\Delta t} = -\frac{\pi \cdot r^4}{8 \cdot \eta} \cdot \frac{\Delta p}{\Delta l} \quad r = \sqrt[4]{-\frac{\Delta V \cdot 8 \cdot \eta \cdot \Delta l}{\Delta t \cdot \pi \cdot \Delta p}}$$

$$i) \quad f_0 = \frac{1}{2\pi} \cdot \sqrt{\frac{D}{m}} \quad D = (f_0 \cdot 2\pi)^2 \cdot m$$

$$m = \frac{D}{(f_0 \cdot 2\pi)^2}$$

$$j) \quad f_0 = \frac{1}{2\pi} \cdot \sqrt{\frac{g}{l}} \quad l = \frac{g}{(f_0 \cdot 2\pi)^2}$$

$$k) \quad f_0 = \frac{1}{2\pi} \cdot \sqrt{\frac{1}{LC}} \quad L = \frac{1}{C \cdot (f_0 \cdot 2\pi)^2}$$

$$C = \frac{1}{L \cdot (f_0 \cdot 2\pi)^2}$$

3.

$$a) \quad \delta = \frac{\lambda_v}{n \cdot \sin(\omega)} \quad \sin(\omega) = \frac{\lambda_v}{n \cdot \delta}$$

$$\omega = \arcsin\left(\frac{\lambda_v}{n \cdot \delta}\right) \quad (\arcsin(\cdot) \text{ oder } \sin^{-1}(\cdot))$$

$$b) \quad \frac{\sin(\alpha_1)}{\sin(\alpha_2)} = \frac{c_1}{c_2} \quad \sin(\alpha_2) = \frac{\sin(\alpha_1) \cdot c_2}{c_1}$$

$$\alpha_2 = \arcsin\left(\frac{\sin(\alpha_1) \cdot c_2}{c_1}\right)$$

$$c) \quad y = y_{\max} \cdot \sin(2\pi f \cdot t + \varphi_0) \quad f = \frac{\arcsin\left(\frac{y}{y_{\max}}\right) - \varphi_0}{2\pi \cdot t}$$

4.

$$a) \quad J = J_0 \cdot e^{-\mu x} \quad x = -\frac{\ln\left(\frac{J}{J_0}\right)}{\mu} = \frac{\ln\left(\frac{J_0}{J}\right)}{\mu} = \frac{\log\left(\frac{J_0}{J}\right)}{\mu \cdot \log(e)}$$

$$\mu = \frac{\ln\left(\frac{J_0}{J}\right)}{x}$$

$$b) \quad J = J_0 \cdot e^{-\frac{\ln(2)}{D}x} \quad x = \frac{D \cdot \ln\left(\frac{J_0}{J}\right)}{\ln(2)} = D \cdot \log_2\left(\frac{J_0}{J}\right)$$

$$D = \frac{x \cdot \ln(2)}{\ln\left(\frac{J_0}{J}\right)} = \frac{x}{\log_2\left(\frac{J_0}{J}\right)}$$

$$c) \quad U = U_0 \cdot e^{-t/(RC)} \quad t = RC \cdot \ln\left(\frac{U_0}{U}\right)$$

$$C = \frac{t}{R \cdot \ln\left(\frac{U_0}{U}\right)}$$

$$d) \quad n_i = n_0 \cdot e^{-\frac{Mg\Delta h}{RT}} \quad \Delta h = \frac{RT}{Mg} \cdot \ln \left(\frac{n_0}{n_i} \right)$$

$$T = \frac{Mg\Delta h}{R \cdot \ln \left(\frac{n_0}{n_i} \right)}$$

$$e) \quad \Lambda = \Lambda_0 \cdot 2^{-\frac{t}{T}} \quad t = T \cdot \log_2 \left(\frac{\Lambda_0}{\Lambda} \right) = \frac{T \cdot \ln \left(\frac{\Lambda_0}{\Lambda} \right)}{\ln(2)}$$

$$T = \frac{t}{\log_2 \left(\frac{\Lambda_0}{\Lambda} \right)} = \frac{t \cdot \ln(2)}{\ln \left(\frac{\Lambda_0}{\Lambda} \right)}$$

$$f) \quad U = U_{\max} \cdot (1 - e^{-t/(RC)}) \quad t = -RC \cdot \ln \left(1 - \frac{U}{U_{\max}} \right)$$

$$R = -\frac{t}{c \cdot \ln \left(1 - \frac{U}{U_{\max}} \right)}$$

5.

$$a) \quad \text{pH} = -\log \left[\frac{[\text{H}^+]}{1 \text{M}} \right] \quad [\text{H}^+] = (10^{-\text{pH}}) \text{M}$$

$$b) \quad E = \log \left(\frac{J_0}{J} \right) \quad J = J_0 \cdot 10^{-E}$$

$$c) \quad S = k_B \cdot \ln(\Omega) \quad \Omega = e^{\frac{S}{k_B}}$$

$$d) \quad n_{\text{dB}} = 10 \cdot \log \left(\frac{J}{J_0} \right) \quad J = J_0 \cdot 10^{\frac{n_{\text{dB}}}{10}}$$

$$e) \quad \text{pH} = \text{p}K + \log \left(\frac{[\text{Basis}]}{[\text{Säure}]} \right) \quad [\text{Basis}] = [\text{Säure}] \cdot 10^{(\text{pH} - \text{p}K)}$$

$$[\text{Säure}] = \frac{[\text{Basis}]}{10^{(\text{pH} - \text{p}K)}}$$

$$f) \quad \text{pH} = \text{p}K + \log \left(\frac{[\text{Basis}]}{c - [\text{Basis}]} \right) \quad [\text{Basis}] = \frac{c \cdot 10^{(\text{pH} - \text{p}K)}}{1 + 10^{(\text{pH} - \text{p}K)}}$$

$$g) \quad E = E^0 + \frac{RT}{z_e F} \cdot \ln \left(\frac{c_{\text{ox}}}{c_{\text{red}}} \right) \quad c_{\text{ox}} = c_{\text{red}} \cdot e^{\frac{(E - E^0) \cdot z_e F}{RT}}$$

$$c_{\text{red}} = \frac{c_{\text{ox}}}{e^{\frac{(E - E^0) \cdot z_e F}{RT}}}$$

$$h) \quad E = E^0 + \frac{0,059 \text{ V}}{z_e} \cdot \log \left(\frac{c_{\text{ox}}}{c_{\text{red}}} \right) \quad c_{\text{ox}} = c_{\text{red}} \cdot 10^{\frac{(E - E^0) \cdot z_e}{0,059 \text{ V}}}$$

$$c_{\text{red}} = \frac{c_{\text{ox}}}{10^{\frac{(E - E^0) \cdot z_e}{0,059 \text{ V}}}}$$