

LIGHT: WAVE AND PARTICLE

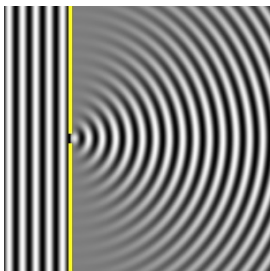
MIKLÓS KELLERMAYER

Biophysics of light

- Light as wave. Wave phenomena.
- Electromagnetic waves, spectrum.
- Light as particle. Photoelectric effect.
- Dual nature of light.
- Matter waves. Electron as wave.
- Applications

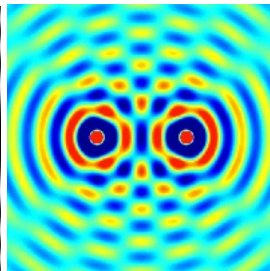
Light as wave:
“wave phenomena” are displayed

Diffraction



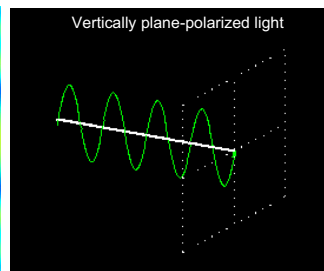
Light “bends” into unexpected areas

Interference



Spatial and temporal pattern of high- and low-amplitude regions

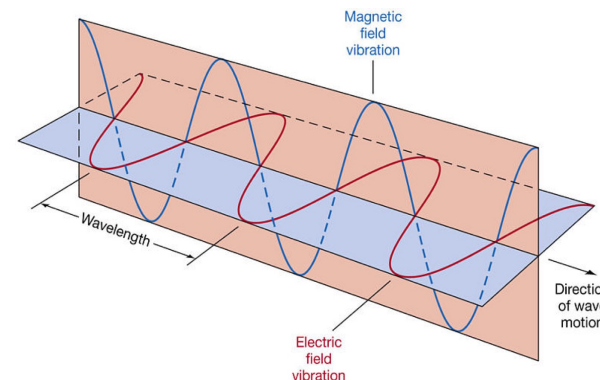
Polarization



Oscillation occurs in a preferred direction

What kind of wave is light? **Electromagnetic wave**

Electromagnetic disturbance propagating in space.
No elastic medium is required for its propagation.



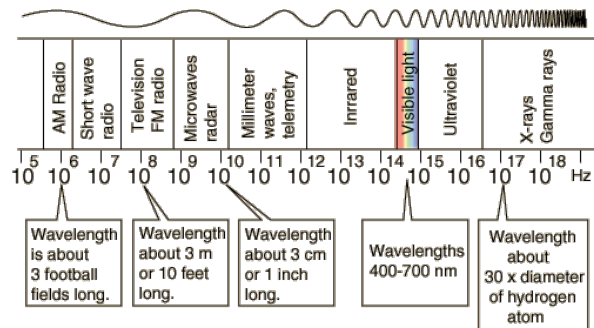
James Clerk Maxwell
(1831-1879)

Propagation velocity:

$$c = \lambda f$$

$$c_{\text{vacuum}} = 2.99792458 \times 10^8 \text{ ms}^{-1}$$

The electromagnetic spectrum



N.B.: 1) "spectrum" = function (intensity of EM radiation as a function of energy)
 2) "electromagnetic spectrum" = types of radiation as a function of energy

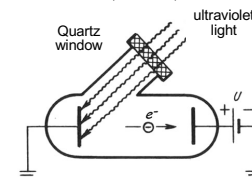
What happens if an object is illuminated with light?

Photoelectric effect: The experiment

Hallwachs-effect:
 Upon UV illumination, negative charges leave the metal surface



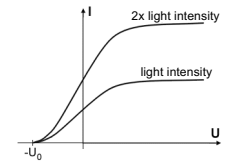
Wilhelm Hallwachs (1859-1922)



Measurements, observations



Philipp Lenard/ Lenard-Fülöp (1862-1947)



- Electron emission: instantaneous upon illumination
- Electron emission only in high-frequency (e.g., blue, UV) light
- No electron emission in low-frequency (e.g., red) light
- Photoelectric current: depends on light intensity
- Photoelectric current: does NOT depend on light color

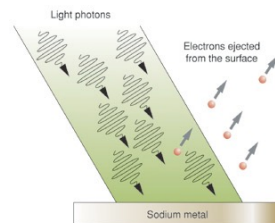
Photoelectric effect

The explanation

1905: "Annus mirabilis"
 • photoelectric effect
 • diffusion
 • special relativity



Albert Einstein (1879-1955)



$$E_{kin} = hf - W_{ex}$$

E_{kin} = kinetic energy of escaped electron
 h = Planck's constant ($6.62 \cdot 10^{-34}$ Js)
 f = frequency of light
 hf = light energy = light quantum, "photon"
 W_{ex} = work necessary for the escape of the electron from the atom

Photon:

- travels with the speed of light (c) in vacuum
- does not exist at rest
- has momentum
- has no rest mass

Light is *at once* wave and particle!



Grote Kerk church, The Hague



Christiaan Huygens (1629-1695)



Sir Isaac Newton (1643-1727)



Westminster abbey

Wave

During propagation

Manifestations:

- Diffraction
- Interference
- Polarization

Particle

During interactions

Manifestations:

- Photoelectric effect
- Refraction
- Excitation, Ionization
- Compton scatter
- Pair production

If light can be a particle, then can a particle be a wave?

Matter waves: The electron as a wave

Einstein: mass-energy equivalence
 $E = mc^2$

Planck: law of radiation
 $E = hf$

Maxwell: speed of light
 $c = \lambda f$



Louis-Victor-Pierre-Raymond, 7th duc de Broglie (1892-1987)

$$mc^2 = h \cdot \frac{c}{\lambda}$$

Momentum of particle (or photon!):

$$P = \frac{h}{\lambda}$$

Wavelength of particle ("de Broglie wavelength"):

$$\lambda = \frac{h}{mv}$$

Why don't we experience the wave nature of macroscopic particles (e.g., bullet)?

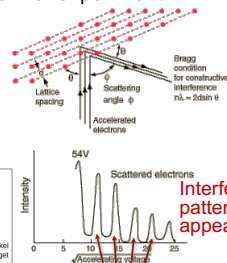
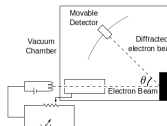


Bullet: for $m=1$ g and $v=1$ km s^{-1} ,
 $\lambda = 6 \times 10^{-34}$ m!!

Davisson-Germer experiment



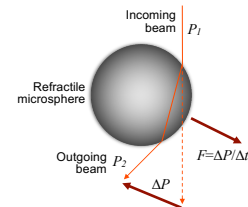
Clinton Joseph Davisson (1881-1958)



The electron is thus a wave!

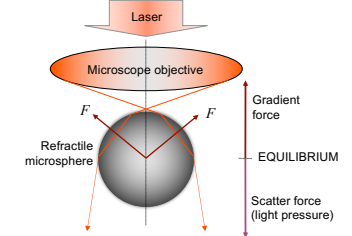
We can now better understand optical tweezers (applications I)

Refraction is accompanied by photonic momentum change (ΔP):

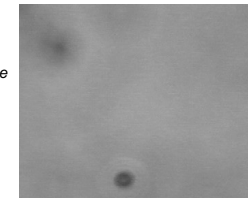


In the **optical trap** a momentum change occurs between the photons and the trapped particle:

Refractive particles may be **captured** with photonic forces:

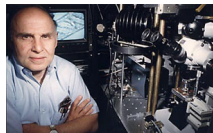


3 μ m latex (polystyrene) microspheres in the optical trap



N.B.: the terms optical tweezers, optical trap, laser tweezers, laser trap are synonymous

Milestones of optical tweezers



Arthur Ashkin (Nobel-prize 2018)

1970: Arthur Ashkin: optical tweezers

1991: J.Spudich, T.Yanagida, J.Molloy, single-molecule mechanics

1994: T.Yanagida, single ATP turnover on myosin

1994: K.Svoboda, S. Block, single kinesin mechanics

1996: C.Bustamante, D.Bensimon, DNA molecule stretch



C. Bustamante

1997: S. Chu, W.D. Phillips, C. Cohen-Tanoudji (Nobel-prize): atom cooling with laser

1997: M.Kellermayer, M.Rief, L.Tskhovrebova, protein molecule stretch

2000: Galajda P., Ormos O., microfabrication with optical tweezers, optically driven microscopic engines

2001: J.Liphardt, C.Bustamante, RNA molecule stretch

2002: Holographic optical tweezers (spatial light modulator, SLM)

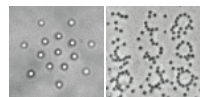
2008: Bustamante, Tinoco: ribosome mechanics



J. Molloy



Microfabricated propeller



Simultaneous manipulation of multiple particles with holographic optical tweezers



J. Spudich



J. Finer

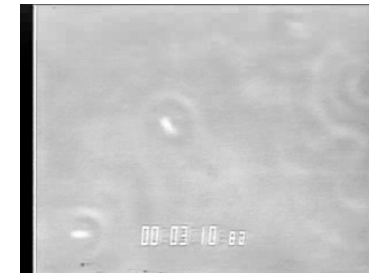


S. Chu, W.D. Phillips, C. Cohen-Tanoudji, Nobel-prize 1997



S. Block

Capturing cells with optical tweezers



Trapping of bacterial cells

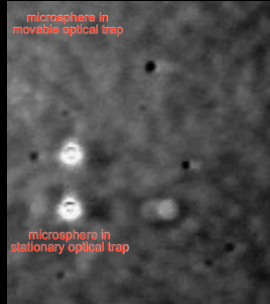
The optical trap is 3D handle without a shaft:
A knot can be tied on a molecular filament without releasing its ends

Actin filament

DNA

Phase contrast image

Fluorescence image

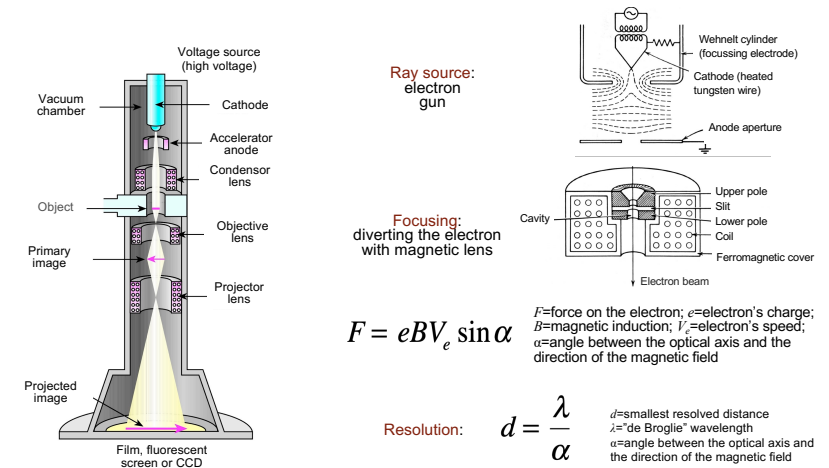


Fluorescence image

Arai et al. Nature 399, 446, 1999.

Applications II.

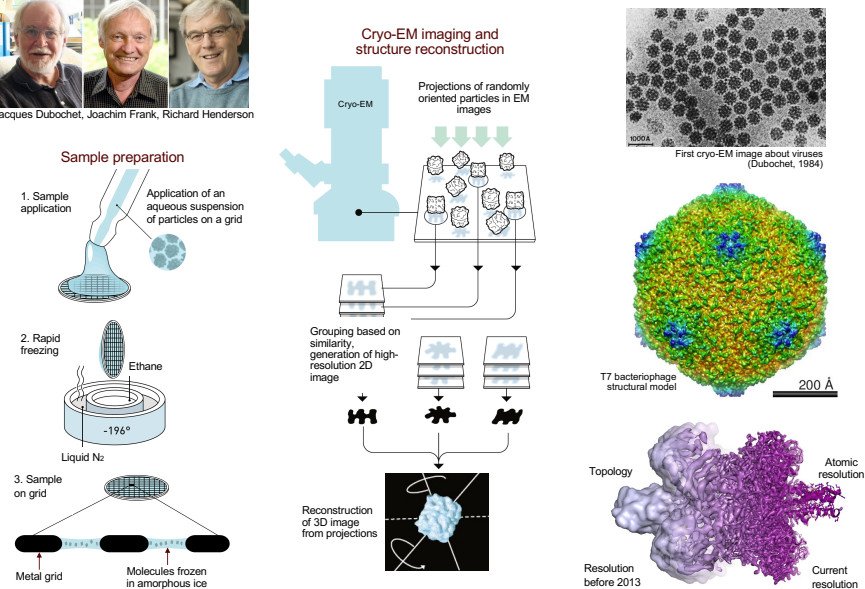
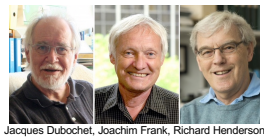
Matter waves: Electron microscope



Transmission electron microscope (TEM)

Based on the de Broglie wavelength the theoretical resolution is: $d \sim 0,005 \text{ nm}$ ($\approx 5 \text{ pm}$).

Cryo-electron microscopy (Nobel-prize 2017)



Applications III.

Photoelectric effect: photodetection, photocell, CCD, etc, etc.....

Light detection, image recording, CCD camera

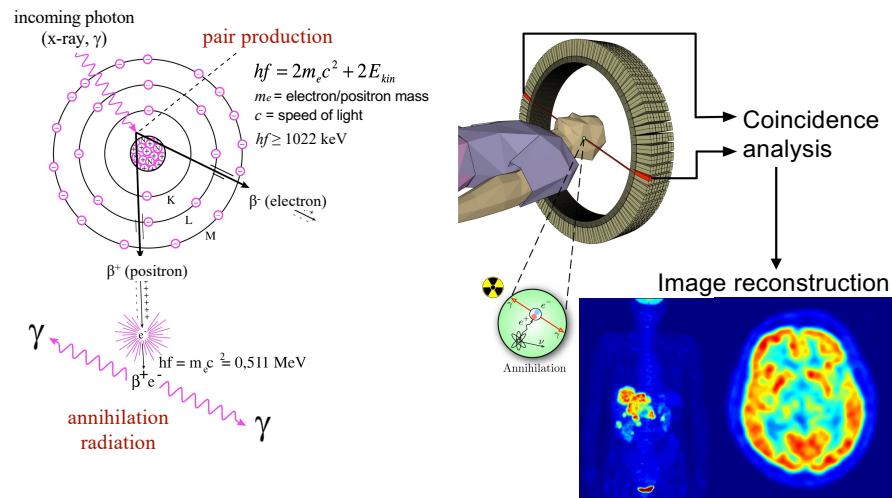
Harvesting and transformation of light energy

Light amplification, intensification

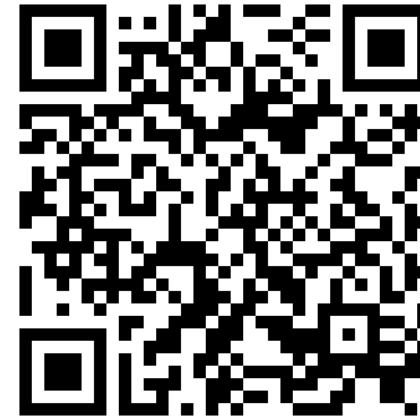


Applications IV.

Pair production – annihilation radiation: Positron Emission Tomography (PET)



Feedback



<https://feedback.semmelweis.hu/feedback/index.php?feedback-qr=B6LHN4TC6ESMLI1Q>