

Theoretical questions – final

- 1 Types of radiations.
- 2 Dependence of irradiance on distance from the source.
- 3 Fundamentals of geometric optics.
- 4 Radiometric quantities.
- 5 Attenuation law.
- 6 Fermat's principle.
- 7 Law of refraction.
- 8 Law of reflection.
- 9 Total internal reflection and its applications.
- 10 Image formation on a curved surface.
- 11 Principal light rays.
- 12 Lens combinations.
- 13 Refractive power.
- 14 Lens equation.
- 15 Image formation by the light microscope.
- 16 Rules of image formation.
- 17 Concepts of magnification and angular magnification.
- 18 Magnification in the light microscope.
- 19 Oscillations.
- 20 Diffraction on an optical grating.
- 21 Polarization of light.
- 22 Types of waves.
- 23 Limit of resolution of the light microscope.
- 24 Phase contrast microscope.
- 25 Huygens-Fresnel principle.
- 26 Polarization microscope.
- 27 Wave interference.
- 28 Wave diffraction.
- 29 Interpretation of the color of light.
- 30 Wave nature of light.
- 31 Dual nature of light.
- 32 Matter waves.
- 33 The electromagnetic spectrum.
- 34 The photoelectric effect.
- 35 The electron microscope.
- 36 Photon energy, the eV scale.
- 37 Interpretation of momentum of light: optical tweezers.
- 38 Models of the atom (Dalton, Thomson).
- 39 Wave nature of the electron.
- 40 The bound electron, quantum numbers.
- 41 Bohr's atomic model.
- 42 Heisenberg's uncertainty principle.
- 43 Physical foundations of the periodic table.
- 44 Franck-Hertz experiment.
- 45 Potential energy of interatomic interactions.
- 46 Electronegativity.
- 47 Scanning probe microscopy.
- 48 Primary and secondary bonds.
- 49 Resolving power of the atomic force microscope.
- 50 The Ideal gas.
- 51 Maxwell-Boltzmann velocity distribution.
- 52 Applications of the Boltzmann-distribution I. : Nernst equation.
- 53 The real gas.
- 54 State equation of real gases.
- 55 Applications of the Boltzmann-distribution II.: equilibrium and rate of chemical reactions. (The Arrhenius plot).
- 56 Macrostate and microstate in thermodynamics.
- 57 Boltzmann distribution.

- 58 Boltzmann's definition of entropy.
- 59 Kinetic gas theory.
- 60 Pressure of ideal gases.
- 61 Applications of the Boltzmann-distribution III.: barometric formula.
- 62 Applications of the Boltzmann-distribution IV.: electric conductivity of semiconductors.
- 63 The crystalline state (unit cell, crystal defects).
- 64 Optical properties of crystalline materials.
- 65 Thermotropic liquid crystals.
- 66 Energy levels of electrical insulators.
- 67 The function of the semiconductor diode.
- 68 Lyotropic liquid crystals.
- 69 Energy levels of electrical conductors.
- 70 The liquid state.
- 71 Electro- and thermo-optical phenomena in liquid crystals.
- 72 Energy levels of intrinsic semiconductors.
- 73 Types of doped semiconductors.
- 74 Light scattering (Rayleigh and Mie).
- 75 The Lambert-Beer law.
- 76 Properties of the absorption spectrum.
- 77 Turbidimetry and nephelometry.
- 78 Dynamic light scattering.
- 79 Measurement of the absorption spectrum.
- 80 Energy levels of atoms and molecules: the Jablonski diagram.
- 81 Thermal radiation.
- 82 Planck's radiation law.
- 83 Light sources based on thermal radiation.
- 84 Absolute black body.
- 85 Emission spectrum of the absolute black body.
- 86 Medical applications of thermal radiation.
- 87 Kirchhoff's law.
- 88 The Stefan-Boltzmann law.
- 89 Wien's displacement law.
- 90 Luminescence: excitation and relaxation.
- 91 Kasha's rule.
- 92 The fluorescence spectrometer.
- 93 Fluorescence.
- 94 Luminescence spectra.
- 95 FRET.
- 96 Phosphorescence.
- 97 Stokes-shift.
- 98 FRAP.
- 99 Notable transitions of luminescence: vibrational relaxation, intersystem crossing.
- 100 Quantum yield of luminescence.
- 101 Fluorescence microscopy.
- 102 Luminescence lifetime.
- 103 Laser: induced emission.
- 104 Laser: the optical resonator.
- 105 Types of lasers.
- 106 Laser: population inversion.
- 107 Properties of laser light.
- 108 Applications of lasers.
- 109 Structure of the atomic nucleus.
- 110 Alpha decay.
- 111 Energy spectra of alpha, beta and gamma radiations.
- 112 Stability of the atomic nucleus.
- 113 Beta negative decay.
- 114 Production of isotopes.
- 115 Isotopes.
- 116 Beta positive decay.

- 117 Types of radioactive decay.
- 118 Gamma decay.
- 119 Activity.
- 120 Interaction of alpha radiation with matter.
- 121 Interaction of gamma radiation with matter I: photoeffect.
- 122 Differential and integral forms of the decay law.
- 123 Interaction of beta negative radiation with matter.
- 124 Interaction of gamma radiation with matter II: Compton-scatter.
- 125 Half-life and average lifetime of an isotope.
- 126 Interaction of beta positive radiation with matter.
- 127 Interaction of gamma radiation with matter III: pair production.
- 128 Neutron radiation, proton radiation, the Bragg-peak.
- 129 Scintillation counter I.: the scintillation crystal.
- 130 The gas ionization chamber.
- 131 Thermoluminescent dosimetry.
- 132 Scintillation counter II.: the photomultiplier tube.
- 133 The Geiger-Müller counter.
- 134 Semiconductor detectors in dosimetry.
- 135 Physical, chemical and biological phases of radiation effects.
- 136 The absorbed dose.
- 137 Converting exposure in air to absorbed dose in tissue.
- 138 The stochastic radiation effect.
- 139 The exposure.
- 140 Weighting factors in dosimetry.
- 141 The deterministic radiation effect.
- 142 The equivalent dose.
- 143 ALARA-principle.
- 144 The direct and indirect effects of ionizing radiations.
- 145 The effective dose.
- 146 Typical dose values and dose limits.
- 147 The dose rate.
- 148 Information obtained by isotope diagnostics.
- 149 Principles of selecting the isotope for diagnostics according to half-life.
- 150 Parts and function of Tc-generator.
- 151 Cost-benefit principle in isotope diagnostics.
- 152 Principles of selecting the isotope for diagnostics according to radiation type and energy.
- 153 Definition of the radiopharmaceutical.
- 154 Parts and function of gamma-camera.
- 155 Determination of the biological half-life of an organ.
- 156 Relative depth dose.
- 157 Scintigraphy.
- 158 SPECT.
- 159 Teletherapy, geometric viewpoints.
- 160 Interpretation of a typical isotope accumulation curve.
- 161 Parts and working principle of PET.
- 162 Role of collimators in radiation therapy, gamma-knife.
- 163 Multimodal imaging: PET/CT and SPECT/MRI.
- 164 Principles of brachytherapy.
- 165 Classification and comparison of signals.
- 166 Typical frequency and amplitude ranges of biological signals.
- 167 Feedback amplifiers.
- 168 Fourier-theorem for periodic and aperiodic signals.
- 169 Parts and function of filtering circuits.
- 170 Digitalization of analog signals.
- 171 Shannon-Nyquist theorem.
- 172 Processing of pulse signals.
- 173 Typical diagnostic wavelength and photon energy range of x-ray.
- 174 Power and efficiency of the x-ray tube.
- 175 The Duane-Hunt-law.

- 176 Structure and function of the x-ray tube.
- 177 Spectrum of Bremsstrahlung.
- 178 Production of characteristic x-rays.
- 179 Mechanisms and energy dependence of x-ray absorption.
- 180 X-ray contrast media.
- 181 CAT-scan: Principles, generations.
- 182 The x-ray summation image.
- 183 X-ray image amplifier.
- 184 CAT-scan: image reconstruction.
- 185 DSA.
- 186 Hounsfield unit, windowing in CAT scan.
- 187 Production of high-energy x-rays.
- 188 Volumetric flow rate, stationary flow.
- 189 Bernoulli's law, plasma skimming.
- 190 Stokes' drag law.
- 191 Laminar and turbulent flow.
- 192 Real fluids: Newton's law of friction.
- 193 Hagen-Poiseuille-law, flow resistance.
- 194 Continuity equation.
- 195 Reynolds-number, critical velocity.
- 196 Determinants of blood viscosity.
- 197 Basics of diffusion: Concepts, thermal motion.
- 198 Fick's I. law.
- 199 Thermodiffusion.
- 200 Brownian motion. Random walk.
- 201 The diffusion coefficient. Einstein-Stokes-equation.
- 202 Heat transport, Fourier's law.
- 203 Physical quantities used for describing the transport of matter.
- 204 Gas exchange between blood and alveoli.
- 205 Osmosis, osmotic pressure, osmolarity.
- 206 Fick's II. law.
- 207 Fundamentals of thermodynamics I.: types of systems, the human body as a thermodynamic system.
- 208 Fundamentals of thermodynamics II.: change of internal energy.
- 209 The I. law of thermodynamics and its applications for biological systems.
- 210 Fundamentals of thermodynamics III.: types of energies, internal energy and its components.
- 211 Extensive and intensive quantities and their relations.
- 212 Entropy and its connections with order, thermal and configurational entropy.
- 213 The II. law of thermodynamics, direction of spontaneous processes.
- 214 The III. law of thermodynamics.
- 215 Direction of processes in isolated, isothermal, and isothermal-isobaric systems.
- 216 Isobaric, isothermal, isothermal-isobaric systems.
- 217 Equilibrium conditions of different thermodynamic systems.
- 218 Thermodynamic potentials.
- 219 Matter transport through the cell membrane.
- 220 The transport model and the Goldman-Hodgkin-Katz-equation.
- 221 Changes in the membrane potential as the function of time.
- 222 Resting transmembrane potential.
- 223 Electric model of the membrane.
- 224 Changes in the membrane potential as the function of space.
- 225 Diffusion of ions across the membrane, permeability.
- 226 The Donnan-equilibrium.
- 227 Properties of the action potential.
- 228 Propagation of the action potential, refractory period and its role.
- 229 Electric signals measured on the body surface for diagnostic purposes.
- 230 Electrochemical potential.
- 231 Ion currents during action potential.
- 232 Sound as a wave.
- 233 Acoustic impedance, reflection of sound, reflectivity.

- 234 Imaging modes in sonography.
- 235 Generation and detection of ultrasound.
- 236 The Doppler-effect, the Doppler-shift.
- 237 Effects of ultrasound, therapeutic applications.
- 238 Absorption of ultrasound.
- 239 The pulse-echo principle.
- 240 Propagation of ultrasound in air and in the body.
- 241 Structure and properties of water.
- 242 Structure of biopolymers.
- 243 Structure and elasticity of DNA.
- 244 Anomalous behavior of water.
- 245 Structural hierarchy of proteins.
- 246 Phase diagram of water.
- 247 Biopolymer elasticity.
- 248 Protein-stabilizing interactions.
- 249 Protein folding.
- 250 Steps of sensory signal transduction.
- 251 Photoreceptors of the retina.
- 252 Biophysics of hearing I.: the outer ear.
- 253 Information coding by the receptor potential.
- 254 Reaction steps of light sensation.
- 255 Biophysics of hearing II.: the middle ear.
- 256 Information coding by the action potential.
- 257 Basis of color sensing.
- 258 Biophysics of hearing III.: Békésy's hearing model.
- 259 Stevens' Law.
- 260 Weber-Fechner law.
- 261 Sensory adaptation.
- 262 Biophysics of hearing IV.: signal transduction in hair cells.
- 263 Signal amplification by hair cells.
- 264 The phon scale.
- 265 The sone scale.
- 266 Biomechanics I.: stress-strain diagram and its ranges.
- 267 Biomechanics IV.: Laplace-Frank-equation.
- 268 Viscoelasticity I.: mechanical model
- 269 Biomechanics II.: Hooke's law, Young-modulus.
- 270 Biomechanical characteristics of bone and enamel.
- 271 Viscoelasticity II.: stress-relaxation, energy dissipation.
- 272 Biomechanics of elastic arteries, distensibility.
- 273 Structure and types of motor proteins.
- 274 Muscle biophysics I.: twitch, summation, tetanus.
- 275 The sliding filament model of muscle contraction.
- 276 Processivity, typical force range and working distance of motor proteins.
- 277 Muscle biophysics II.: isometric and isotonic contraction.
- 278 The cross-bridge cycle of skeletal muscle myosin.
- 279 Muscle biophysics III.: work and power. Force-velocity curve.
- 280 Bragg-diffraction of X-rays.
- 281 Time of flight principle in mass spectrometry.
- 282 Determination of molecular structure by x-ray crystallography.
- 283 Ionization methods in mass spectrometry: electrospray, MALDI.
- 284 Mass spectrometry in medicine: proteomics, diagnostics, oncoknife.
- 285 Stern-Gerlach-experiment.
- 286 Macroscopic magnetization in MRI: spin-spin relaxation.
- 287 Spatial encoding in MRI.
- 288 Zeeman-effect.
- 289 Macroscopic magnetization in MRI: spin-lattice relaxation.
- 290 MRI contrast methods: proton density, T1 and T2 weighting.
- 291 Larmor-precession and nuclear magnetic resonance.
- 292 Differences between NMR and ESR spectroscopies.
- 293 Chemical shift.

- 294 Circulatory biophysics: function of the blood vessel system.
- 295 Pressure relations in the arterial system.
- 296 The cardiac cycle.
- 297 Changes in pressure in the circulatory system.
- 298 Auxiliary factors of circulation: the windkessel effect.
- 299 Pressure-volume relation of the heart.
- 300 Changes in the total cross section of vessels in the circulatory system.
- 301 Electrical description of heart function.
- 302 Work of the heart.
- 303 Changes in the flow velocity in the blood vessel system.
- 304 Respiratory biophysics I.: partial pressure, Henry's law.
- 305 Respiratory cycle.
- 306 Biophysics of physical examination I.: Inspection.
- 307 Box model of the human respiratory system.
- 308 Respiratory volumes and capacities.
- 309 Biophysics of physical examination II.: palpation.
- 310 Conductive and gas-exchange parts of the human respiratory system.
- 311 Biomechanics of respiration (compliance, obstructive and restrictive pathologies).
- 312 Biophysics of physical examination III.: percussion
- 313 Respiratory work.
- 314 Biophysics of physical examination IV.: auscultation.

Practice questions – final

- 1 How does refractive power of a lens change if its radius of curvature decreases?
- 2 How does refractive power of a lens change if its radius of curvature increases?
- 3 What is radius of curvature in case of a lens?

- 4 How does refractive power of a lens change if its index of refraction increases?
- 5 Calculate the refractive power of a lens with a focal distance of 25 cm.
- 6 Calculate the refractive power of a lens with a focal distance of 20 cm.
- 7 Calculate the refractive power of a lens with a focal distance of 17 mm.
- 8 Characterize the image of an object placed within the focal distance of a converging lens.
- 9 Characterize the image of an object placed between the single and the double focal distance of a converging lens.
- 10 Characterize the image of an object placed outside the double focal distance of a converging lens.
- 11 What kind of image is formed by a compound light microscope?
- 12 What is the total magnification of a light microscope if the objective magnification is 100x and the ocular magnification is 20x?
- 13 Describe the steps of the eyepiece scale calibration process.
- 14 What prisms are present in the Abbe-refractometer?
- 15 What sample can be measured with the Abbe-refractometer?
- 16 What is the role of Amici prism?
- 17 What is optical dispersion?
- 18 Factors influencing the value of index of refraction.
- 19 Formation of Snell circle.
- 20 How do you determine concentration by refractometry?
- 21 What is the refractive index of distilled water?
- 22 Definition of absorption spectrum.
- 23 What information can you obtain from an absorption spectrum?
- 24 How do you determine concentration by absorption photometry?
- 25 Define optical density (absorbance).
- 26 Define transmittance.
- 27 How much light is transmitted by a sample with an absorbance of 1?
- 28 Which sample transmits more light: OD=1 or OD=3? By how much?
- 29 How does the absorption spectrum change if the sample concentration is doubled?
- 30 How does the absorption spectrum change if the sample concentration is halved?

- 31 What is the absorption maximum characteristic of?
- 32 What is the function of the monochromator?
- 33 Define optical activity based on the refractive index.
- 34 Define Biot-law.
- 35 Describe the linearly polarized light.
- 36 Describe the circularly polarized light.
- 37 What light source is used for polarimetry and why?
- 38 How does optical rotation angle change if the sample tube length decreases?
- 39 How does optical rotation angle change if the sample concentration increases?
- 40 What is a chiral molecule? Provide an example.
- 41 Factors influencing specific optical rotation.
- 42 How do you determine concentration by polarimetry?
- 43 Refractive media of the eye. Image formation of the eye.
- 44 What is the refractive power of the unaccommodated human eye?
- 45 Which refractive surface contributes the most to the refractive power of the human eye?
- 46 How does the refractive power of human eye change during accommodation?
- 47 Describe the process of focal accommodation of the human eye.
- 48 How do you calculate the accommodation power of human eye?
- 49 How would you measure the position and diameter of the blind spot?
- 50 What is myopia and how do you correct it?
- 51 What is hyperopia and how do you correct it?
- 52 What is presbyopia and how do you correct it?
- 53 What is visual acuity and how do you measure it?
- 54 How did we measure the visual acuity?
- 55 Describe the reduced eye model.
- 56 Factors influencing the visual acuity.
- 57 Spatial distribution of photoreceptors on the retina.
- 58 What is the visual acuity of a patient with a limiting angle of view of 2'?
- 59 Parts of the scintillation counter.
- 60 Sources of noise in the scintillation counter.
- 61 How do you reduce external noise in scintillation counting?
- 62 How do you reduce internal noise in scintillation counting?
- 63 Define the integral discriminator.
- 64 Define the signal-to-noise ratio.
- 65 How to find the optimal ID setting of the scintillation counter?
- 66 How many electrons arrive at the PMT anode for every photoelectron if the number of the dynodes is 8 and the multiplication factor is 2.
- 67 Define the mass attenuation coefficient.
- 68 Define the surface density.
- 69 Define the attenuation coefficient.
- 70 Define the half-value layer thickness.
- 71 Define the tenth-value layer thickness.
- 72 Explain the energy dependence of mass attenuation coefficient in case of lead (graph in formula collection).
- 73 Compare the linear attenuation and mass attenuation coefficients for water and steam.
- 74 What fraction of intensity is transmitted through an absorber with a thickness twice its half-value layer thickness ($x=2D$).
- 75 What fraction of intensity is transmitted through an absorber with a thickness three times its half-value layer thickness ($x=3D$).
- 76 Harmonic oscillation (definition, equation, graph.)
- 77 Damped free oscillation.
- 78 Driven oscillation, resonance.
- 79 Resonance curve.
- 80 How does the resonance frequency change if the oscillating mass is doubled?
- 81 How does the resonance frequency change if the spring constant is doubled?
- 82 Define the eigenfrequency.
- 83 How do you determine the spring constant of a cantilever?

- 84 Definition and components of impedance.
- 85 Electric model of the skin.
- 86 Definition and unit of capacitive reactance.
- 87 Specific resistance of the skin.
- 88 Specific capacitance of the skin.
- 89 What component of skin impedance dominates in case of DC versus high frequency AC?
- 90 What is the difference between the measuring and the auxiliary electrodes in skin impedance measurement?
- 91 Definition and unit of capacitance.
- 92 Describe Ohm's law.
- 93 Alternating current and RMS voltage.
- 94 Gain and gain level of the amplifier.
- 95 Compare voltage and power gain.
- 96 What is the gain level if the voltage gain equals 1000?
- 97 What is the gain level if the voltage gain equals 1?
- 98 What is the power gain if the gain level is 3 dB?
- 99 Frequency response curve of the amplifier.
- 100 How do you determine the transfer band of an amplifier?
- 101 How does the bandwidth of an amplifier change with negative feedback?
- 102 Advantages and disadvantages of using negative feedback in an amplifier.
- 103 Voltage divider circuit.
- 104 Coulter principle.
- 105 Parts and functions of the Coulter-counter.
- 106 How does the voltage pulse amplitude depend on particle size in the Coulter-counter?
- 107 How do you separate the red-blood-cell versus white-blood-cell signals in a Coulter-counter?
- 108 How do you separate the red-blood-cell versus platelet signals in a Coulter-counter?
- 109 What is the role of differential discriminator in Coulter-counting?
- 110 Why is dilution of blood necessary in Coulter-counting?
- 111 What solution is used for blood dilution in Coulter-counting?
- 112 How does the matter flow density (flux) change if the concentration gradient is doubled?
- 113 Name the parameters influencing the value of diffusion coefficient.
- 114 What diffuses faster: a potassium ion or a virus particle?
- 115 How does the average distance travelled by a diffusing particle depend on time?
- 116 On what length scale diffusion is an effective transport process?
- 117 How does the the minimum wavelength of the x-ray spectrum change with increasing anode voltage?
- 118 How does the the minimum wavelength of the x-ray spectrum change with increasing anode current?
- 119 What is the maximum x-ray photon energy at 50 kV anode voltage?
- 120 What elements are suitable as x-ray-tube anode material?
- 121 What are the x-ray spectral lines characteristic of?
- 122 Why do we need to cool the x-ray-tube anode?
- 123 How does x-ray absorbance depend on the atomic number of absorber?
- 124 Which part of the x-ray spectrum is attenuated by filtering?
- 125 Which is a better x-ray absorber: Al or Ag?
- 126 Which attenuation mechanism dominates in x-ray diagnostics?
- 127 What are the parts of a gamma-radiation pulse amplitude spectrum?
- 128 Effect of activity on the pulse amplitude spectrum of a gamma-radiating isotope.
- 129 Effect of anode voltage on the pulse amplitude spectrum of a gamma-radiating isotope.
- 130 Compare the pulse amplitude spectra of two different gamma-radiating isotopes.
- 131 How can you determine the gamma energy of a radioactive isotope with a scintillation counter?
- 132 What type of discriminator is used for acquiring the pulse amplitude spectrum?

- 133 Define the human hearing range (threshold of hearing, threshold of pain, frequency limits).
- 134 Which one is louder: 50 Hz, 120 dB vs. 1 kHz, 110 dB sound (formula collection, isophone curves) ?
- 135 Which one is louder: 30 Hz, 90 dB vs. 1 kHz, 70 phon sound (formula collection, isophone curves) ?
- 136 How much louder is a 80 dB versus a 70 dB sound at 1000 Hz?
- 137 Does a greater dB value always correspond to a louder sound?
- 138 Does a greater phon value always correspond to a louder sound?
- 139 Does a greater sone value always correspond to a louder sound?
- 140 Definition and interpretation of the audiogram.
- 141 Define hearing loss and overhearing.
- 142 Describe a pulse signal.
- 143 The monostable multivibrator and its applications.
- 144 The bistable multivibrator and its applications.
- 145 The astable multivibrator and its applications.
- 146 Special functions of a pacemaker.
- 147 Parameters of a pacemaker pulse: period, amplitude, duty ratio, energy.
- 148 Explain the ECG curve.
- 149 Compare the depolarization and repolarization processes of skeletal and cardiac muscles.
- 150 Types of ECG leads I.: bipolar leads.
- 151 Types of ECG leads II.: unipolar chest leads.
- 152 Types of ECG leads III.: (semi)unipolar limb leads.
- 153 Calculate the value of R(III) if R(I)= 0.2 mV and R(II)= 1 mV in the standard ECG leads.
- 154 Einthoven-triangle, integral vector.
- 155 Differential amplifier of ECG.
- 156 What is the voltage amplitude of a 12-mm-high R(I) signal if vertical sensitivity is 1 mV/cm?
- 157 What is the duration of a 2-mm-wide QRS complex if the horizontal scale is 25 mm/s?
- 158 Definition of x-ray density and its significance in CAT-scan.
- 159 Compare the x-ray absorption of bone and muscle tissue.
- 160 Compare the x-ray absorption of lung and muscle tissue.
- 161 What is the x-ray density of a voxel that absorbs 90% of the incident x-ray.
- 162 What is the advantage of using x-ray density (D) in computer tomography?
- 163 How do we resolve 3D structure in CT scanning?