

# DNA, RNA, PROTEINS

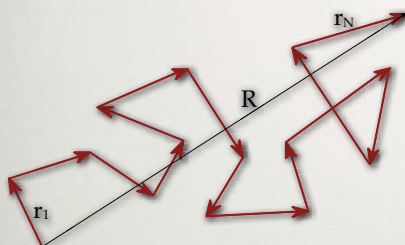
MIKLÓS KELLERMAYER

## BIOPHYSICS OF MACROMOLECULES

- **Space**  
Size, shape, local and global structure
- **Time**  
Fluctuations, structural change, folding
- **Interactions**  
Internal and external interactions, bonds, bond energies  
Mechanics, elasticity

## SHAPE OF THE POLYMER CHAIN RESEMBLES RANDOM WALK

Brown movement: random walk



“Square-root law”:  $\langle R^2 \rangle = Nl^2 = Ll$

$R$  = end-to-end distance

$r_i$  = elementary vector

$N$  = Number of elementary vectors

$l = |\vec{r}_i|$  = correlation length (“persistence length”, describes bending rigidity)

$Nl = L$  = contour length

In case of Brown-movement  $R$ =displacement,  
 $N$ =number of elementary steps,  $L$ =total path length,  
és  $l$ =mean free path length.

Tendency for entropy maximization results  
in chain *elasticity*

**Entropic\* elasticity:**

Thermal fluctuations of the polymer chain



Configurational entropy (orientational disorder of  
elementary vectors) increases.



The chain shortens.



\*Entropy: disorder

## BIOPOLYMER ELASTICITY IS RELATED TO GLOBAL SHAPE

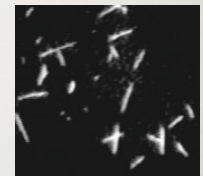
$l$  = persistence length: measure of bending rigidity  
 $L$  = contour length

**Rigid chain**

$l \gg L$



Microtubule

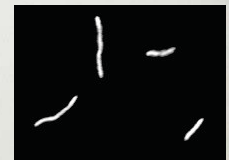


**Semiflexible chain**

$l \sim L$



Actin filament

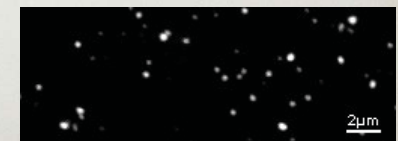


**Flexible chain**

$l \ll L$

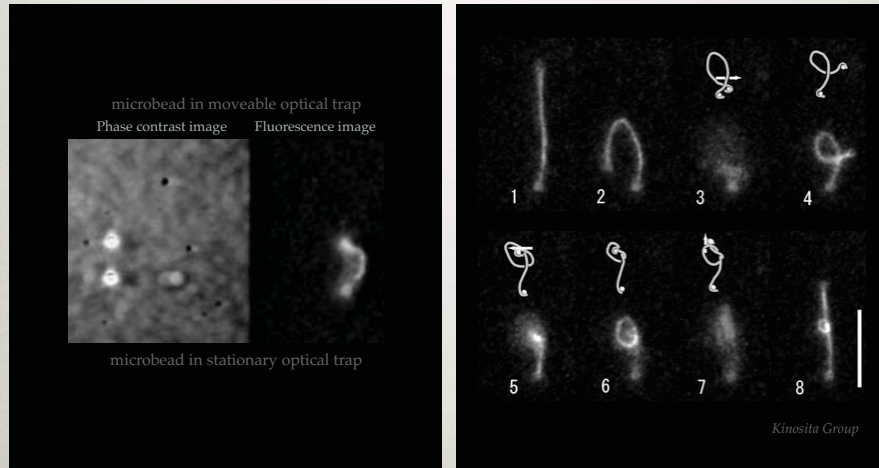


DNA



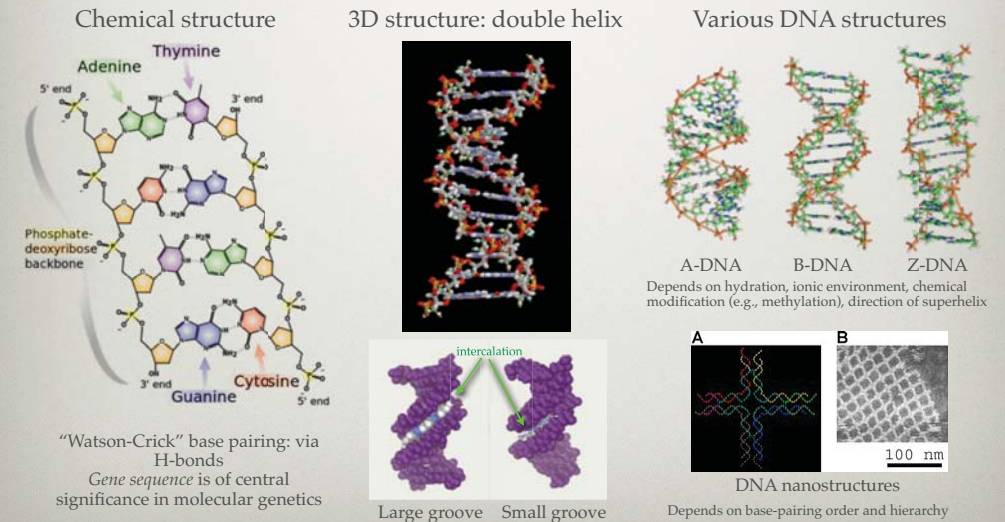
# VISUALIZATION OF BIOPOLYMER ELASTICITY

Tying a knot on a single DNA molecule



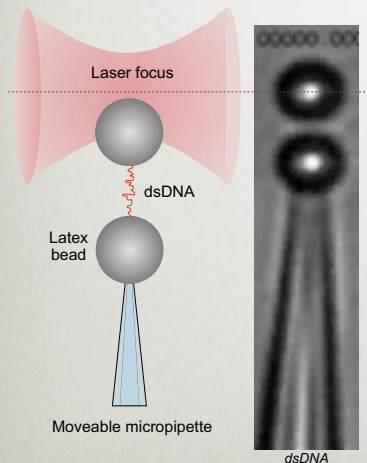
# 1. DNA: DEOXYRIBONUCLEIC ACID

*Function:* molecule of biological information storage

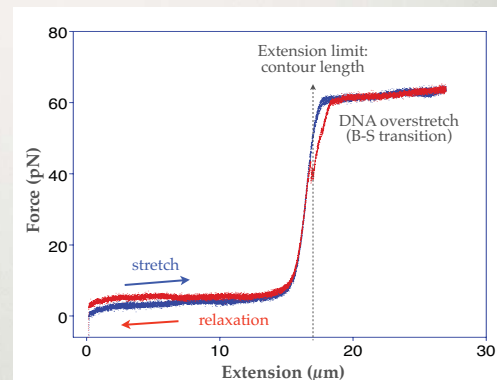


# THE DNA MOLECULE IS ELASTIC!

Force measurement: with optical tweezers

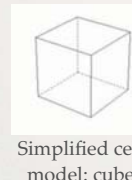
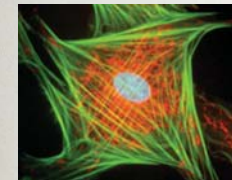


Force versus extension curve of a single dsDNA molecule



Persistence length of dsDNA: ~50 nm  
Overstretch transition at ~65 pN

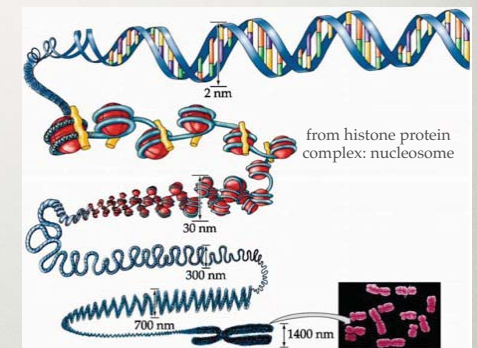
# HOW MUCH DNA IN A CELL?



Simplified cell model: cube

*Solution:* DNA needs to be packed

Chromosome condensation



	Cell: 20 μm edge cube	Analogue - Lecture hall: 20 m edge cube
DNA thickness	2 nm	2 mm
Full length of human DNA	~2 m	~2000 km (!!!)
Persistence length of dsDNA	~50 nm	~50 cm
End-to-end distance (R)	~350 μm (!)	~350 m (!)
Volume of fully compacted DNA	~2 x 2 x 2 μm <sup>3</sup>	~2 x 2 x 2 m <sup>3</sup> (= 8 m <sup>3</sup> )

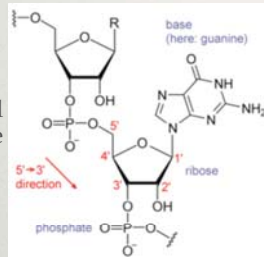
- Condensins play a role in high-order DNA packaging
- DNA chain: complex linear path with roadblocks!



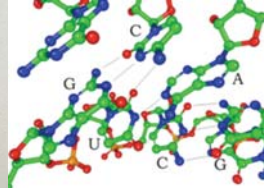
## 2. RNA: RIBONUCLEIC ACID

**Function:** information transfer (transcription), structural element (e.g., ribosome), regulation (turning gene expression on and off)

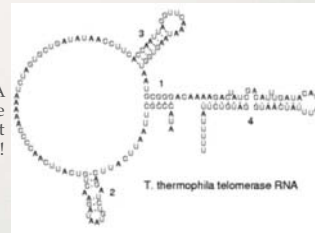
Chemical structure



"Watson-Crick" base pairing



The RNA molecule is not paired!

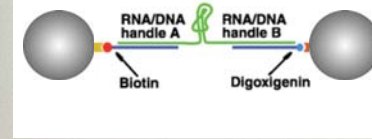


Secondary and tertiary structural elements

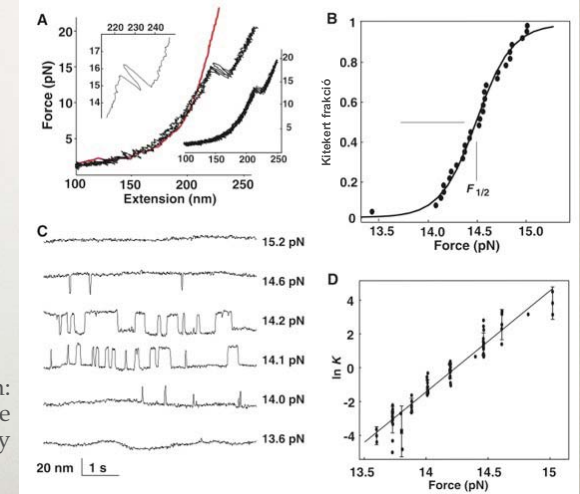


## RNA STRUCTURE CAN BE PERTURBED WITH MECHANICAL FORCE

Stretching with optical tweezers



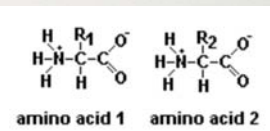
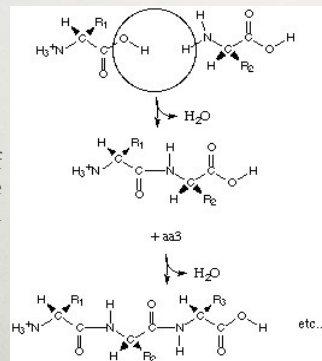
Unfolding of RNA hairpin: near reversible process - the RNA hairpin refolds rapidly



## 3. PROTEINS: BIOPOLYMERS INTERCONNECTED WITH PEPTIDE BONDS

**Function:** most important molecules of the cell.  
Highly diverse functions - structure, chemical catalysis energy transduction, motoric functions, etc.

Formation of the peptide bond



Condensation reaction followed by the release of water

## PROTEIN STRUCTURE

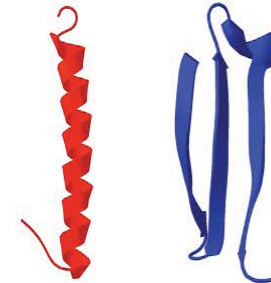
Primary

Amino acid sequence

Determines spatial structure as well.

Secondary

$\alpha$ -helix  
 $\beta$ -sheet  
 $\beta$ -turn ( $\beta$ -hairpin)

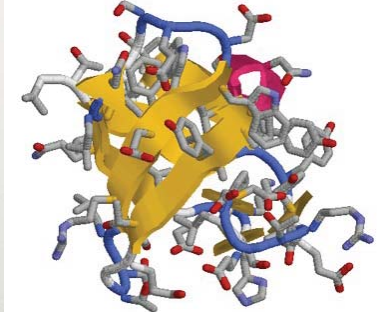


$\alpha$ -helix:  
• right handed  
• 3.4 residue/turn  
• H-bridges

$\beta$ -sheet:  
• parallel or antiparallel  
• H-bridges between distant residues

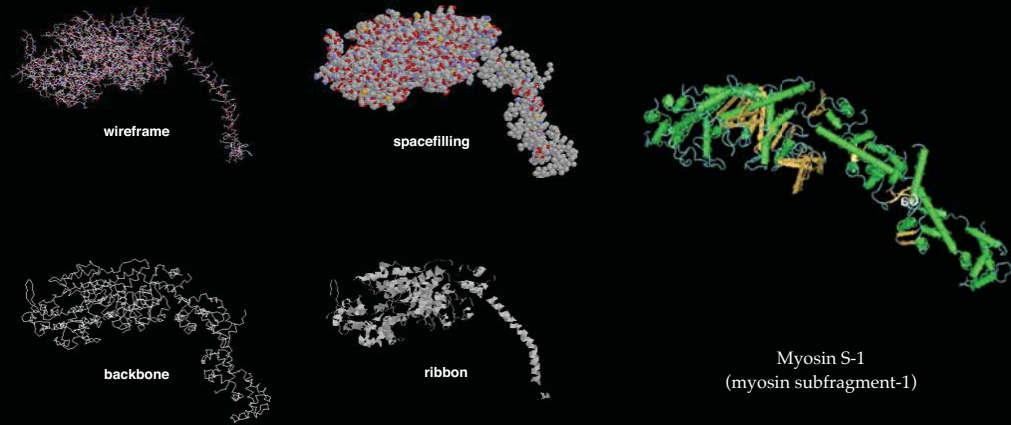
Tertiary

3D structure of single-chain protein



\*Quaternary structure: binding of independent subunits into a complex

## DISPLAY OF PROTEIN STRUCTURE



## BONDS HOLDING PROTEIN STRUCTURE TOGETHER

- Weak (secondary) bonds
1. **Hydrogen bond**: proton sharing between proton-donor side chains.
  2. **Electrostatic interaction** (salt bridge): between oppositely charged residues.
  3. **van der Waals bond**: weak interaction between atoms (molecules) with closed electron shells.
  4. **Hydrophobe-hydrophobe interaction**: between hydrophobic residues (in the interior of the molecule).
- Covalent bond
5. **Disulfide bridge**: between cysteine side chains; connects distant parts of the protein chain.

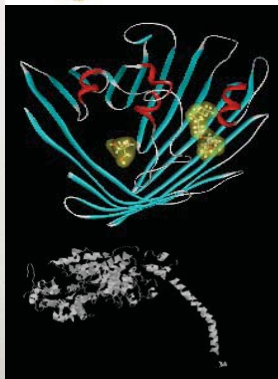
## PROTEIN STRUCTURE CLASSES

### 1. All alpha



calmodulin

### 2. All beta

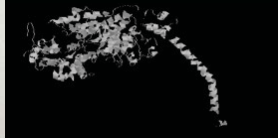


porin

### (3. Alpha-beta)

### 4. Multidomain

Domain:  
folding subunit



myosin

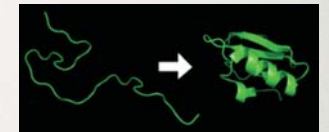
Although there are as many sequences as proteins, the spatial structures are classified into a surprisingly small number of classes!

## HOW IS THE THREE-DIMENSIONAL STRUCTURE ACQUIRED?



Christian Anfinsen  
(1916-1995)

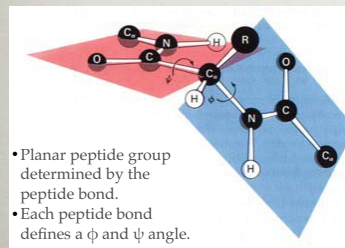
**Anfinsen**: proteins fold spontaneously (sequence determines structure)



Unfolded state

Native state (N)  
Lowest energy

**Levinthal's paradox** (Cyrus Levinthal, 1969):  
Are all available conformations explored?

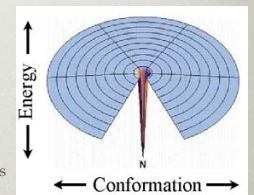


Number of possible conformations (degrees of freedom):  $i^n$

$i$  = number of possible angular positions of a given  $\phi$  or  $\psi$  angle  
 $n$  = total number of  $\phi$  and  $\psi$  angles

Example: in a peptide composed of 100 residues the number of possible  $\phi$  or  $\psi$  angles is 2.  
 $n=198m$ .

Number of possible conformations:  $2^{198}(!!!)$

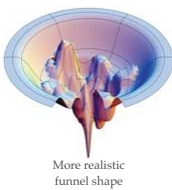
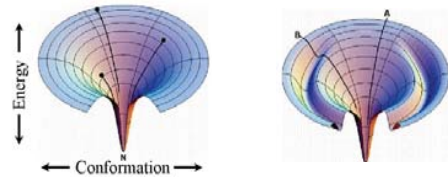


What is the probability that a billiards ball will find the hole merely via random motion?



## PROTEIN FOLDING IS GUIDED BY THE SHAPE OF ITS CONFORMATIONAL SPACE

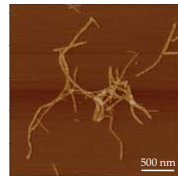
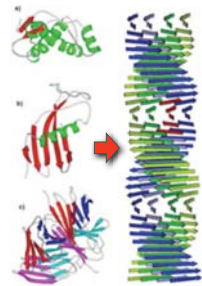
### Shape of conformational space: "Folding funnel"



- Proteins "slide down" the wall of the funnel.
- Folding funnel shape can be complex (determination of the shape is usually very difficult).
- A protein may get stuck at intermediate states (pathology).
- In the living cell chaperones assist folding.

### Pathology

- Protein "folding diseases"
- Alzheimer's disease
- Parkinson's disease
- II-type diabetes
- Familial amyloidotic neuropathy



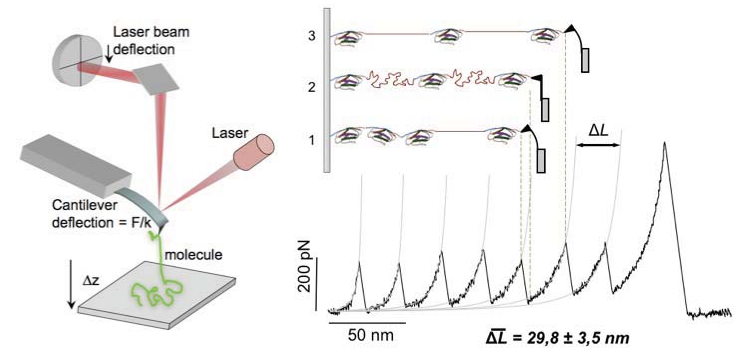
$\beta$ -fibrils:  
undissolved precipitate  
cross- $\beta$  structure

## METHODS OF PROTEIN UNFOLDING (DENATURATION)

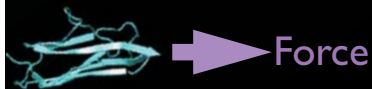
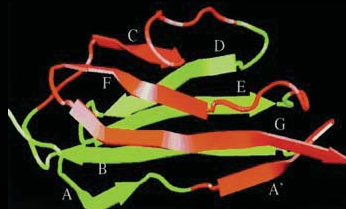
- Heat
- Chemical agent
- Mechanical force

Break secondary chemical bonds  
Disrupt secondary and tertiary structure

### Mechanical unfolding of a single protein with atomic force microscope



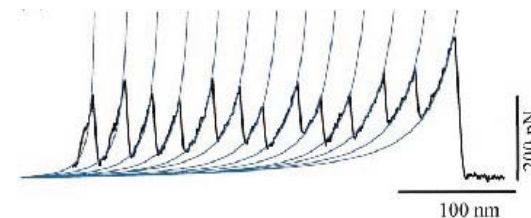
## Titin's Ig domains are mechanically stable



## BASIS OF MECHANICAL STABILITY: PARALLEL COUPLING OF H-BONDS

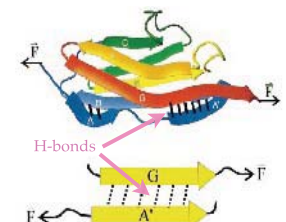
Mechanical stability provided by shear pattern of H-bond patch

### Force spectrum



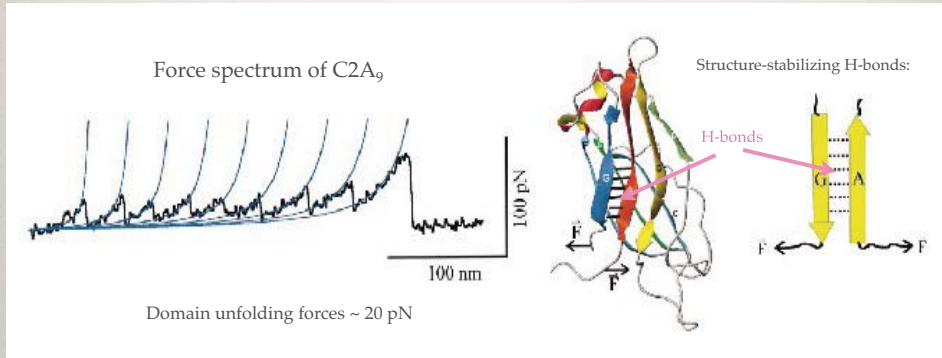
Domain unfolding force > 200 pN

### Structure-stabilizing H-bonds:



## LOW MECHANICAL STABILITY: H-BONDS ARE COUPLED IN SERIES

Low mechanical stability due to zipper pattern of H-bond patch

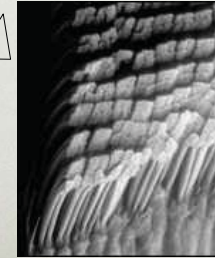


## LOGIC OF MECHANICAL STABILIZATION IN MACROSCOPIC SYSTEMS

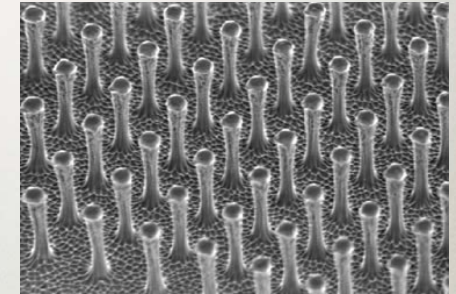
Principle of parallel mechanical coupling



Gecko foot stickiness:  
Bristles (setae)  
coupled in parallel



Application:



Artificial gecko foot