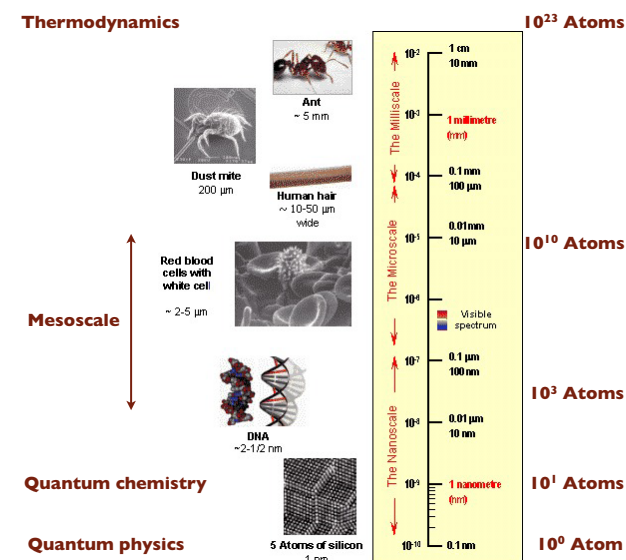


# Biomolecular structure

Miklós Kellermayer

## Why structure...?



## Outline

- Diffraction, interference
- X-ray diffraction, X-ray crystallography
- Diffraction-limited microscopies
- Beating the diffraction limit
- Polarization; CD spectroscopy
- Mass spectrometry

## We access structure via interaction with electromagnetic waves

### Properties of electromagnetic waves

**Amplitude (A)** - Intensity  $\sim A^2$

**Oscillation time (period):** duration of a single oscillation ("T").

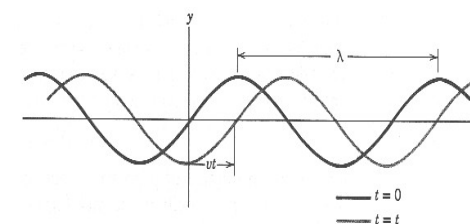
**Frequency:** inverse of period (f).

The wave propagates with a given **velocity** ("phase velocity", "v" or "c")

Distance between points of identical phase: "**wavelength**" (λ)

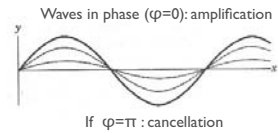
**Phase difference (φ):** difference (in time or space) between identical points of waves

$$\lambda = cT = \frac{c}{f}$$

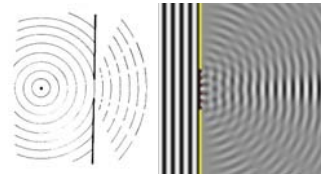


# Diffraction and interference

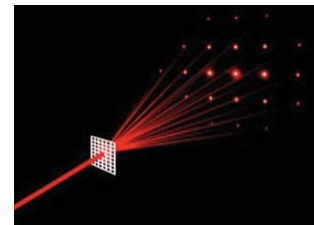
**Interaction of waves:**  
constructive or desctructive  
interference (amplification versus  
cancellation)



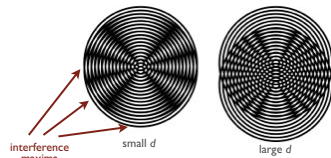
Diffraction through slits with a distance  
comparable to wavelength (=pointlike slits  
separated by distance  $d$ , where  $d \sim \lambda$ )



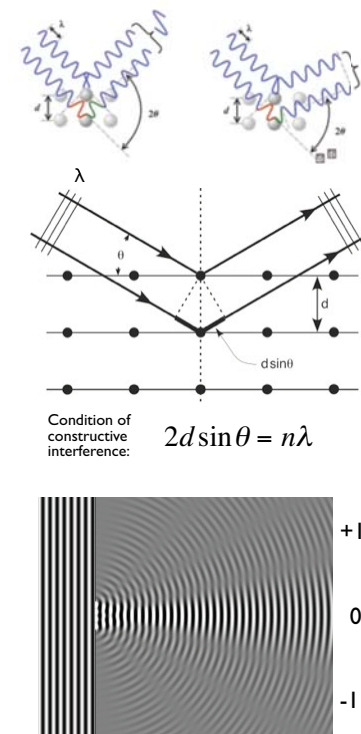
Interference pattern of a 2-dimensional grating



**Interference pattern** depends on distance ( $d$ )  
separating the wave sources

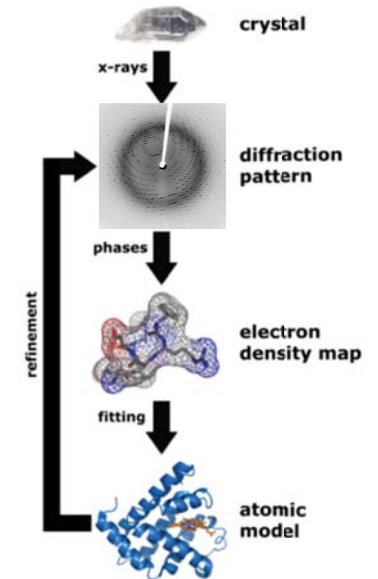


# X-ray crystallography

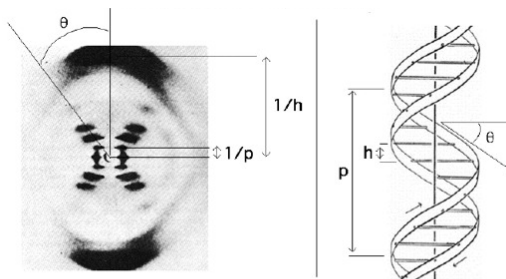


Condition of  
constructive  
interference:

$$2d \sin \theta = n\lambda$$

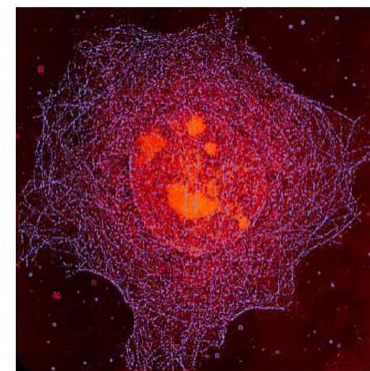


**Solving DNA  
structure with X-ray  
crystallography**



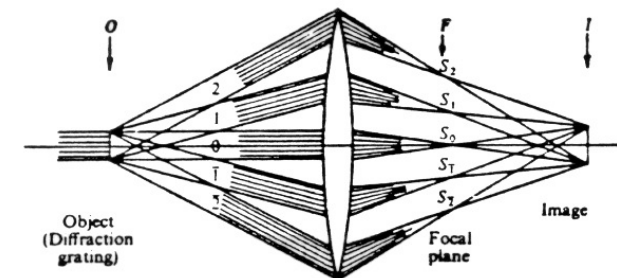
$\theta$  - tilt of helix (angle from perpendicular to long axis)  
 $h = 3.4 \text{ \AA}$  (Distance between bases)  
 $p = 34 \text{ \AA}$  (Distance for one complete turn of helix;  
Repeat unit of the helix)

**Image formation with  
X-ray**



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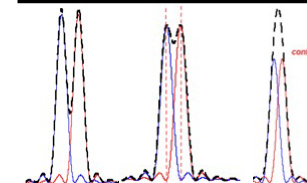
# Imaging with waves



Rayleigh Criterion : Two light sources must be  
separated by at least the diameter of first dark band.



Due to diffraction:  
image of a point  
object is an Airy disk



Smallest resolved  
distance (Abbé):

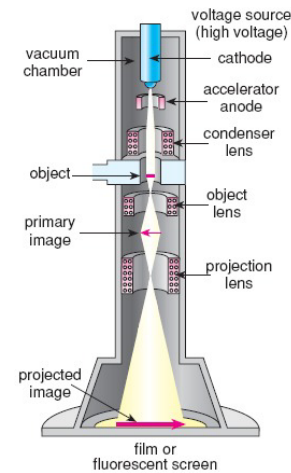
$$d = \frac{0.61\lambda}{n \sin \alpha}$$

# How to beat the Abbé formula?

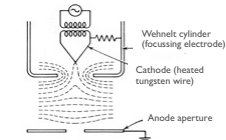
- Improve equation parameters (decrease wavelength, increase numerical aperture)
- Convert resolution problem into position-determination problem
- Use non-diffraction-limited imaging

# Image with electron waves: the electron microscope

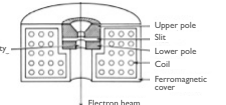
## Transmission electron microscope (TEM)



Ray source: electron gun



Focussing: diverting the electron with magnetic lens



$$F = eBV_e \sin \alpha$$

$F$ =force on the electron;  $e$ =electron's charge;  $B$ =magnetic field;  
 $V_e$ =electron's speed;  $\alpha$ =angle between the optical axis and the direction of the magnetic field

Resolution: 
$$d = \frac{\lambda}{\alpha}$$

$d$ =smallest resolved distance  
 $\lambda$ =de Broglie wavelength  
 $\alpha$ =angle between the optical axis and the direction of the magnetic field

Based on the de Broglie wavelength the theoretical resolution is:  $d \sim 0.005 \text{ nm}$  ( $\approx 5 \text{ pm}$ ).

# Resolution and contrast in the electron microscope

## A. Theoretical resolution:

Modified Abbe-formula (for small  $\alpha$  angles)

Based on electron velocity (100000 km/s),  $d=0.005 \text{ nm}$

$$d = \frac{\lambda}{\alpha}$$

## B. Real resolution: limited by small NA, $\sim 0.1 \text{ nm}$ .

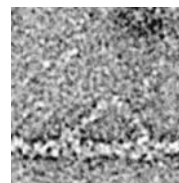
Because of small NA, depth of focus is large (several  $\mu\text{m}$ ).

## C. Practical resolution in biological samples:

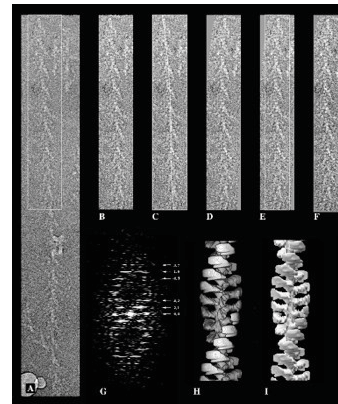
1/10 of section thickness.

## D. Contrast mechanism: electron diffraction

Contrast enhancement: by electron dense dyes



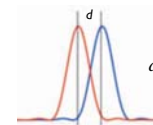
Cryo-electron microscopy, particle-analysis image reconstruction  
The Muscle Group, Leeds 2000



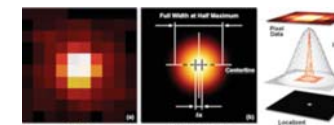
# Super-resolution microscopy

Resolution problem is converted into position-determination problem

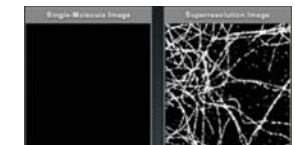
## Resolution problem (Abbé)



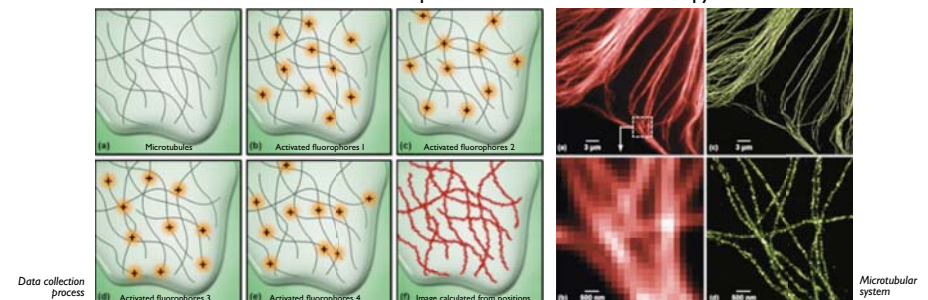
## Position determination problem (precision depends on photon count)



## "Stochastic" data collection, single fluorophores



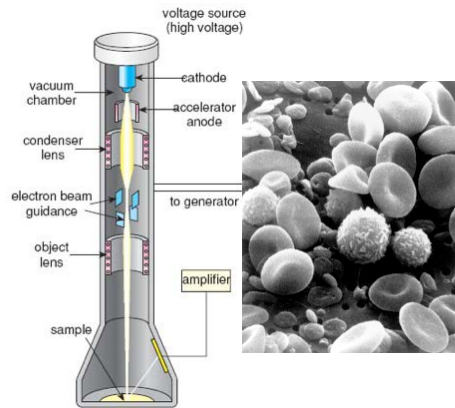
## STORM: "stochastic optical reconstruction microscopy"





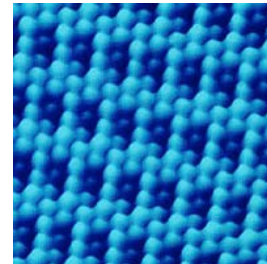
# Depart from diffraction-limited imaging

## Scanning electron microscope (SEM)



## Scanning-probe microscopies (SPM, AFM)

Oxygen atoms on the surface of rhodium crystal



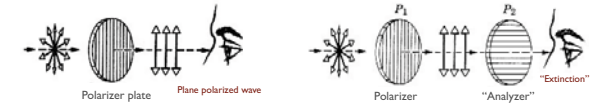
Unit of nanoworld:  
1 nanometer

# Polarization

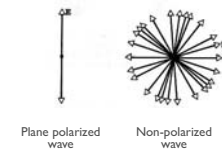
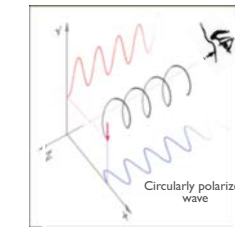
- **Polarization:** oscillation is oriented in some preferred direction
- **Birefringence** is related to polarization: anisotropic propagation velocity
- Only transverse waves can be polarized.



Polarization of  
Electromagnetic waves

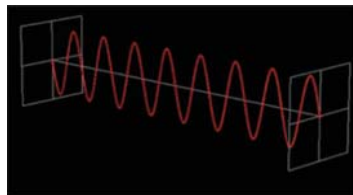


Head-on view of  
polarization plane:

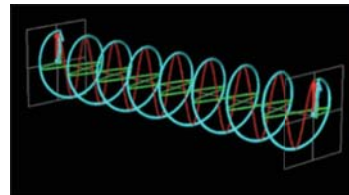


# Interaction of polarized light with matter

Linearly (plane) polarized light

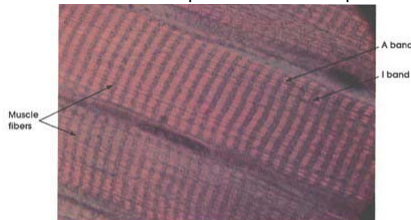


Circularly polarized light

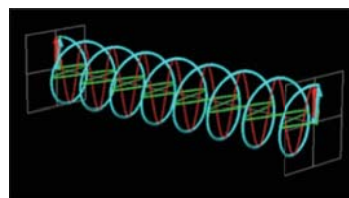


Rightward  
rotating (D)

Muscle fiber in polarization microscope



Optical birefringence: refractive index (~speed of light) depends on polarization plane of linearly polarized light



Leftward  
rotating (L)

Circular birefringence: refractive index (~speed of light) depends on rotational direction of circularly polarized light

# Circular dichroism (CD) spectroscopy

**Principle: wavelength-dependent differential absorption of L/D circularly polarized light**

- Absorption of circularly polarized light depends on rotation direction (L/D) AND
- Absorption of circularly polarized light depends on frequency (wavelength)

**Dichroism ("two colors"):**

- 1.) Certain wavelengths of light either pass through or are reflected from a material surface
- 2.) Light in different polarization states travelling through a material are absorbed by different amounts.

**Chiral molecules display strong circular dichroism**

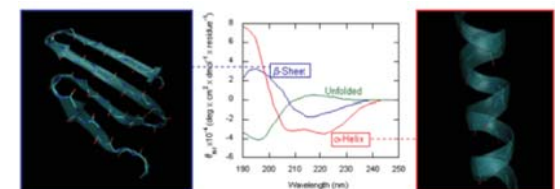


**CD spectrum:**

x-axis: wavelength or frequency (mostly wavelength); y-axis: difference in molar extinction coefficients of L and D circular polarized light (or a quantity proportional to this coefficient)

$$\Delta A(\lambda) = A(\lambda)_{LCPL} - A(\lambda)_{RCPL}$$

$\lambda$ : wavelength

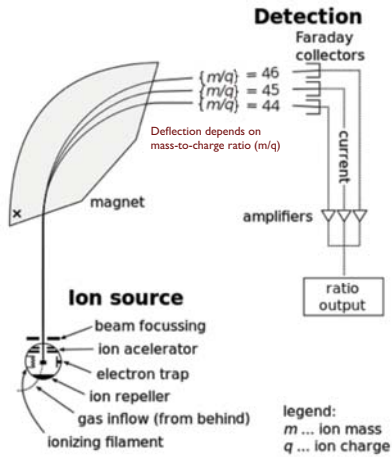


# Mass spectrometry

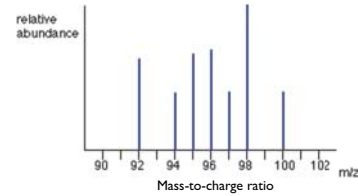
**Mass spectrometry (MS):** analytical technique producing spectra of the masses of the atoms or molecules in a sample. The spectra are used to determine the elemental or isotopic signature, thereby elucidating the chemical structures of molecules.

## Steps:

1. Ionization
2. Acceleration
3. Deflection
4. Detection



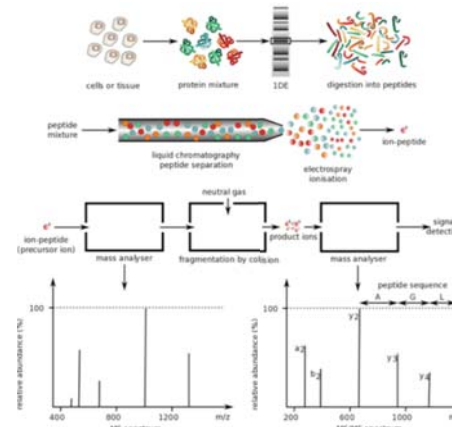
## Result: "Stick" diagram



Spectrum is compared with structure database

# Mass spectrometry applications

## Protein analysis (proteomics)



## Real-time tissue analysis ("onco-knife")

