

Mass Spectrometry

Mass

Inertial Mass

$F = m \cdot a$
force mass acceleration
e.g. particles
accelerated in
a force field

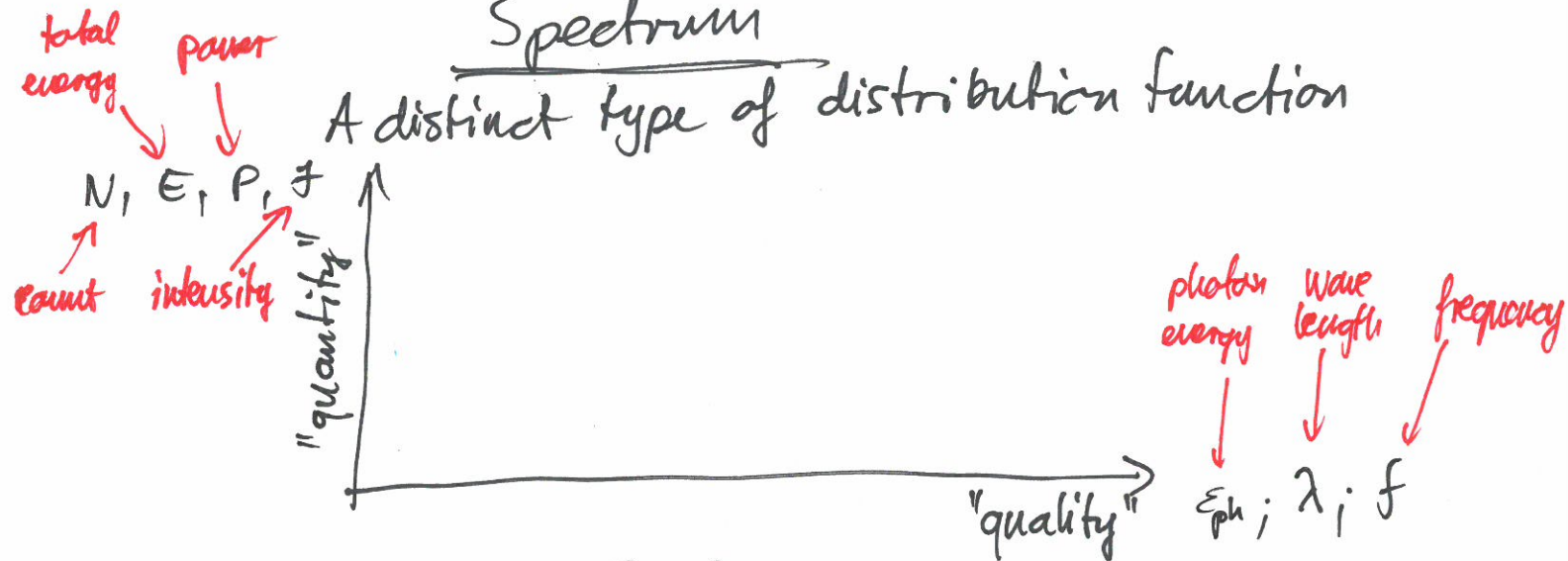
↓
mass spectrometry
microscopic technique
(down to femtomoles)

Gravitational Mass

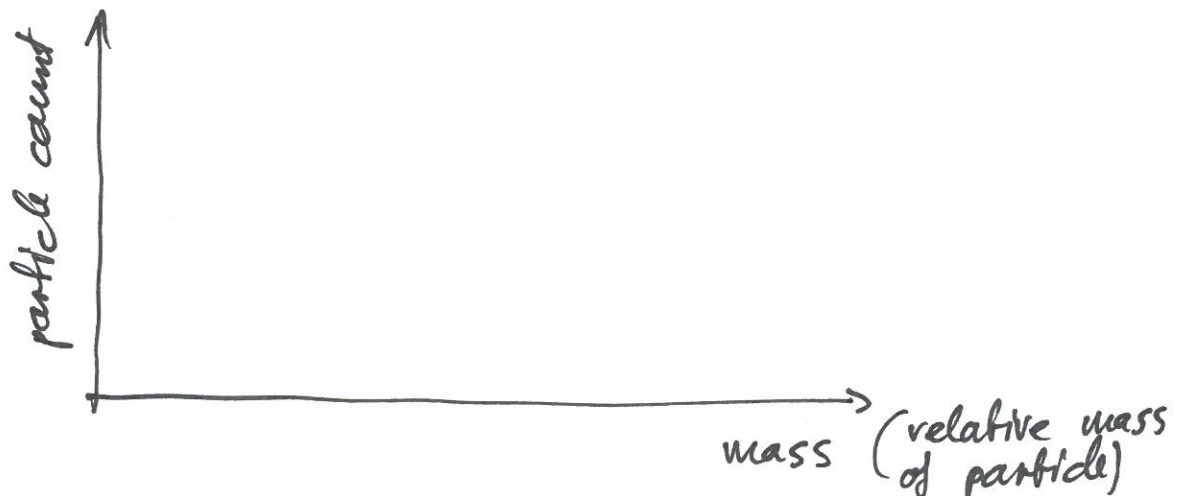
• → ← •
 $F_w = g \cdot m$
Weight gravitational
(force) acceleration
"weight measurement"
↓
gravimetry
macroscopic technique
(down to μg)

Spectrum

A distinct type of distribution function



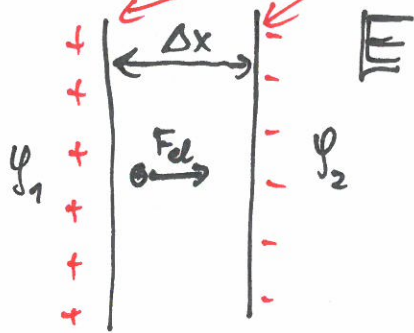
Mass Spectrum



How can a charged particle be accelerated

Electric field

capacitor plates



$$F_e = E \cdot q$$

electric force

electric field strength

charge of particle

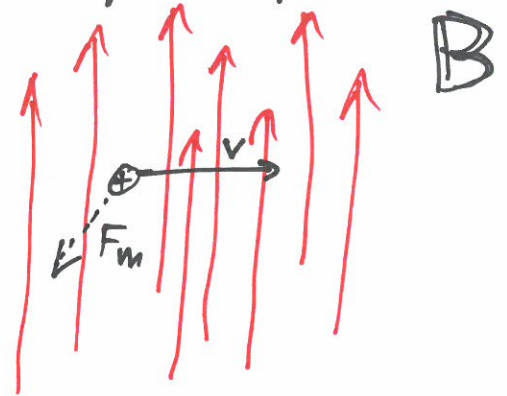
$$E = \frac{\Delta \phi}{\Delta x}$$

electric potential difference
" Voltage

distance between plates

As a result of electric field, the magnitude of velocity will increase
 $|v|$ increases

Magnetic field



$$F_m = q \cdot v \times B$$

magnetic force

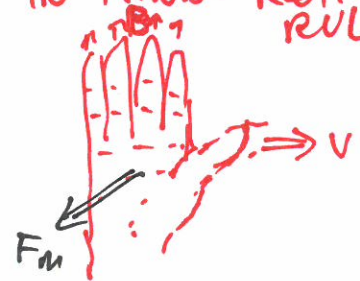
charge

speed

magnetic field strength

vectorial product

KEEP IN MIND: RIGHT HAND RULE

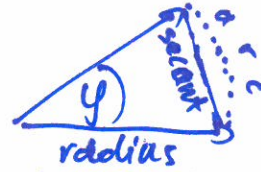
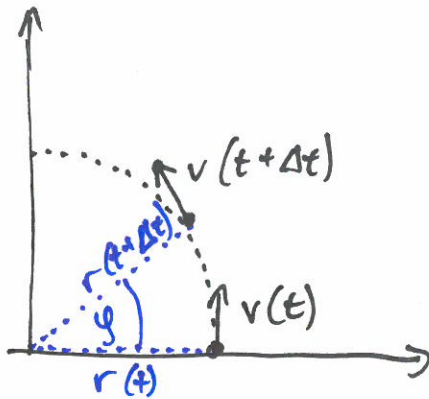


As a result of the magnetic field, the moving charged particle will be forced onto a curvilinear path

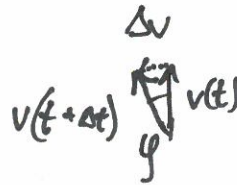
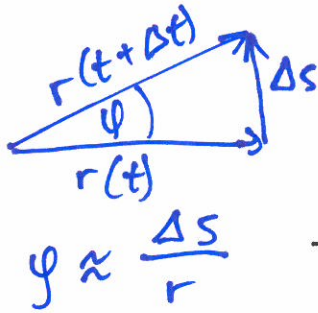
Circular motion & centripetal acceleration

definition of angle:

$$\varphi = \frac{\text{arc}}{\text{radius}} \approx \frac{\text{secant}}{\text{radius}}$$



for small angles, the difference between arc and ~~arc~~ secant is negligible



$$\Delta v = v(t + \Delta t) - v(t)$$

$$\varphi \approx \frac{\Delta v}{v}$$

Definition of speed:

$$v = \frac{\Delta s}{\Delta t}$$

$$v \cdot \Delta t = \Delta s$$

$$\frac{\Delta s}{r} = \frac{\Delta v}{v}$$

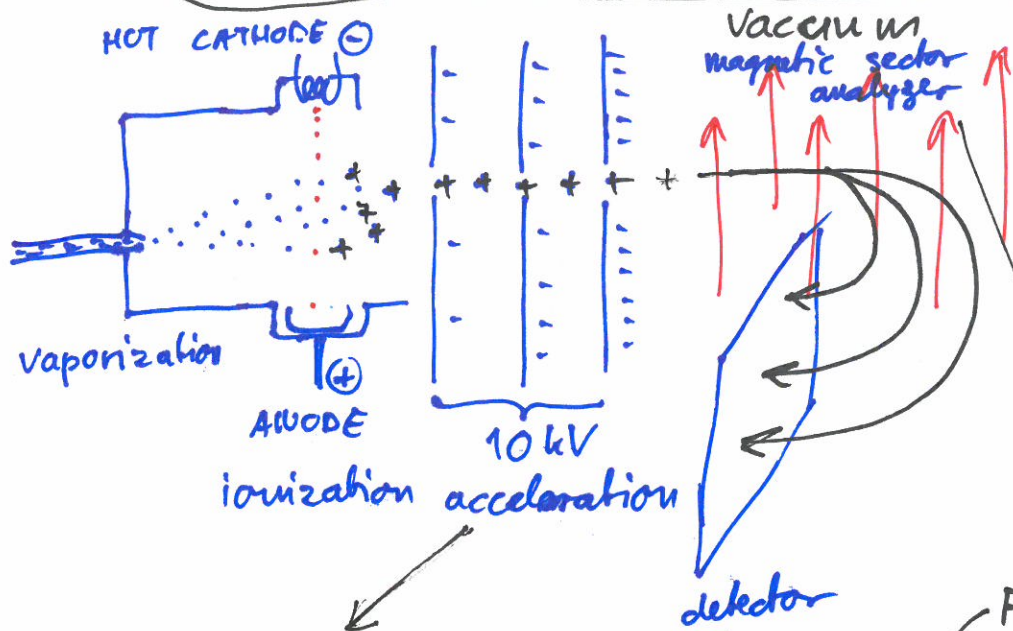
$$\frac{v \cdot \Delta t}{r} = \frac{\Delta v}{v}$$

centripetal acceleration $a_{cp} = \frac{\Delta v}{\Delta t} = \frac{v^2}{r}$

centripetal force $F_{cp} = a_{cp} \cdot m$

Basic Mass Spectrometer

Sample Inlet \Rightarrow vaporization atomization ionization fragmentation \Rightarrow mass analyzer \Rightarrow detector



$B = 5000 \text{ G} = 0.5 \text{ T}$
 the SI unit of B is tesla (T) the former cgs unit is gauss (G)
 $1 \text{ T} = 10^4 \text{ G}$

$$E_{\text{kin}} = \frac{1}{2}mv^2 \quad E_{\text{el}} = q \cdot U$$

$$qU = \frac{1}{2}mv^2$$

$$v = \sqrt{\frac{2qU}{m}}$$

$$F_{\text{cp}} = a_{\text{cp}} \cdot m = \frac{v^2}{r} \cdot m$$

$$F_{\text{m}} = q \cdot v \cdot B$$

$$\frac{v^2}{r} \cdot m = q \cdot v \cdot B$$

$$r = \frac{v \cdot m}{q \cdot B}$$

$$r = \frac{\sqrt{\frac{2qU}{m}} \cdot m}{q \cdot B} = \frac{1}{B} \cdot \sqrt{\frac{2U \cdot m}{q}}$$

e.g.

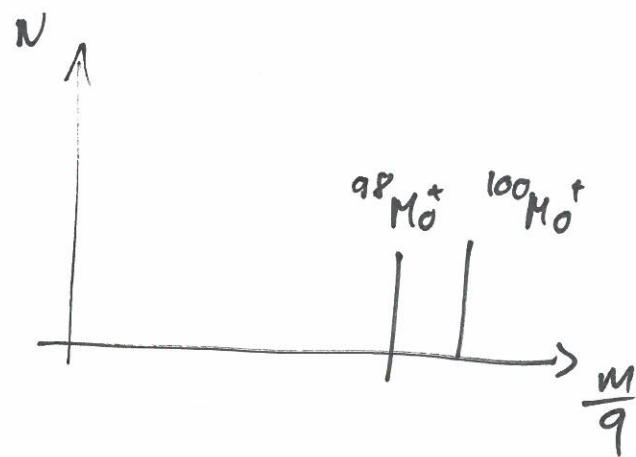
$^{98}\text{Mo}^+$	$1.6333 \times 10^{-25} \text{ kg}$
$^{100}\text{Mo}^+$	$1.6867 \times 10^{-25} \text{ kg}$

$$q = +1.6 \times 10^{-19} \text{ C}$$

$$r(^{98}\text{Mo}^+) = 0.28577 \text{ m}$$

$$r(^{100}\text{Mo}^+) = 0.28868 \text{ m}$$

$$\Delta r = 2.9 \times 10^{-3} \text{ m} = 2.9 \text{ mm}$$



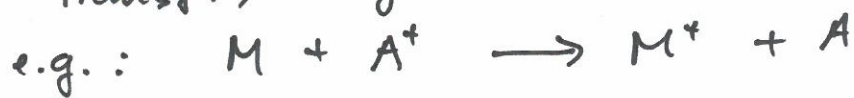
Types of Mass Spectrometers

1.) Ion sources (different ways of ionization)

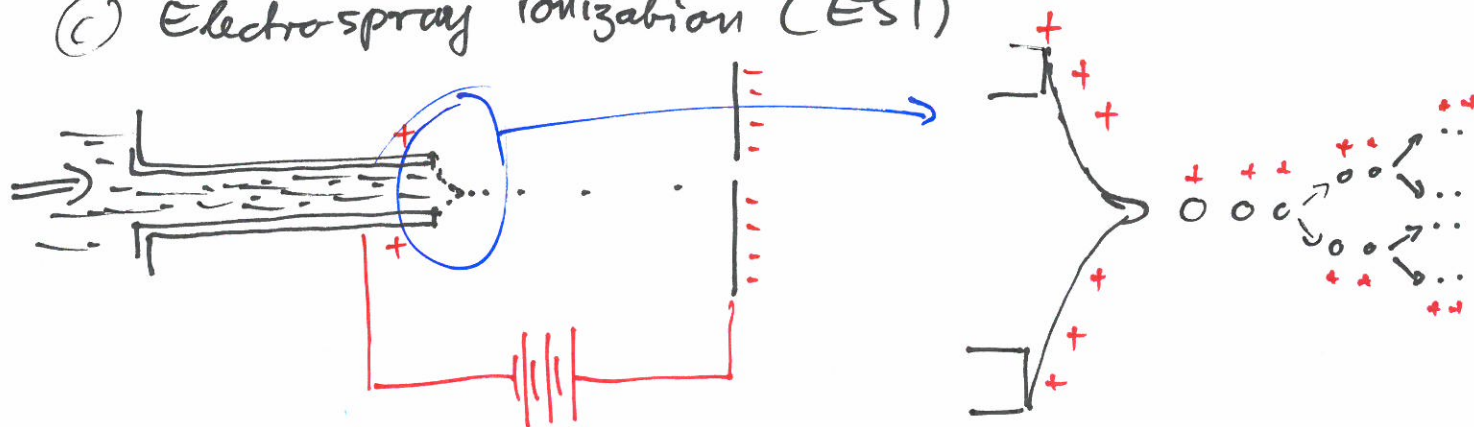
(A) Electron Ionization (EI) "hard" ionization (high degree of fragmentation) SEE ABOVE

(B) Chemical Ionization (CI)

- an inert substance (e.g. NH_3 , CH_4) is brought into the ionization space (A)
- this gets ionized $A + e^- \rightarrow A^+ + 2e^-$
- the sample (analyte) is introduced into the sample space (M)
- the A^+ transfers charge to the analyte

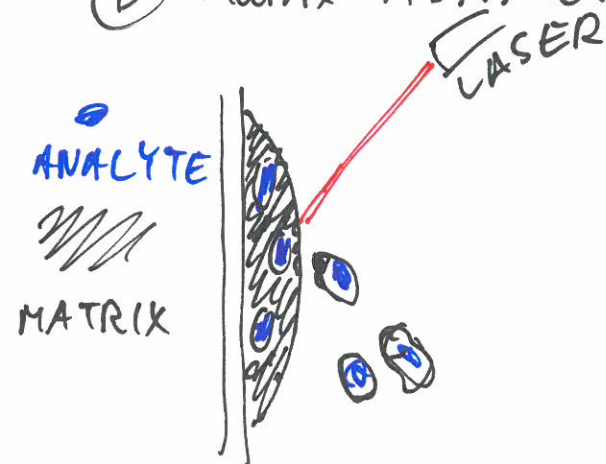


(C) Electrospray Ionization (ESI)



- the droplet becomes smaller due to evaporation
- the surface charge density will increase
- the droplet becomes unstable \rightarrow explodes

(D) Matrix Assisted Laser Desorption Ionization (MALDI)



Very mild ionization technique

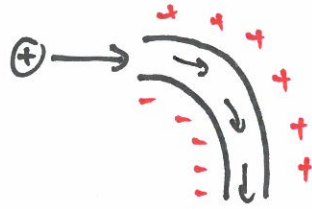
2) Mass analyzers

(A) Magnetic sector



$$F_{cp} = F_m$$

(B) Electric sector : bent capacitor

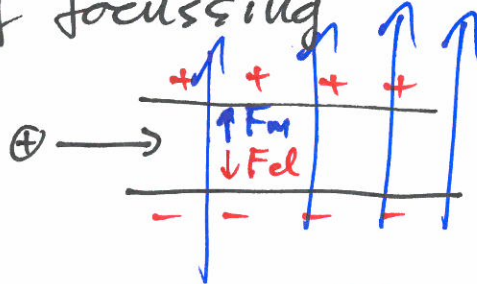


$$F_{cp} = F_{el}$$

$$\frac{m \cdot v^2}{r} = E \cdot q$$

$$r = \frac{v^2}{E} \cdot \frac{m}{q}$$

- however, speed is a variable here
- we need to select a single speed
velocity focussing



it will go through :

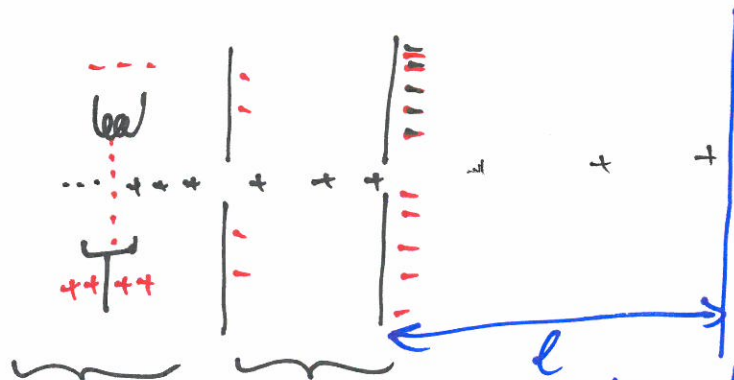
$$F_{el} = F_m$$

$$q \cdot U = q \cdot v \cdot B$$

$$U = v \cdot B$$

$$v = \frac{U}{B}$$

(C) Time of flight (TOF)



ionization accel.

analysis = time measurement

$$v = \frac{l}{t} \Rightarrow t = \frac{l}{v} = \frac{l}{\sqrt{\frac{2 \cdot U \cdot q}{m}}}$$

[-6-]