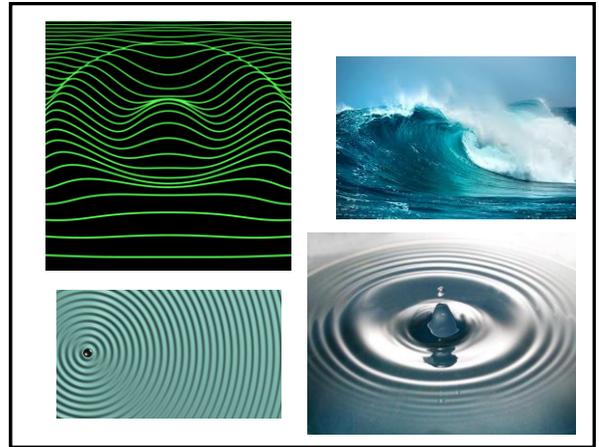


Wave optics

Waves are everywhere, not just in optics!

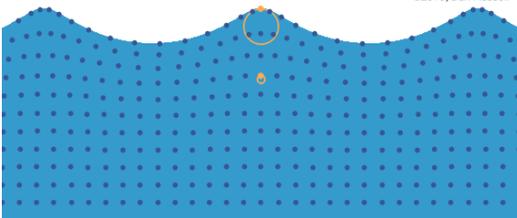
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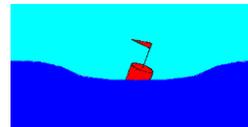


Waves are seen most often in water:

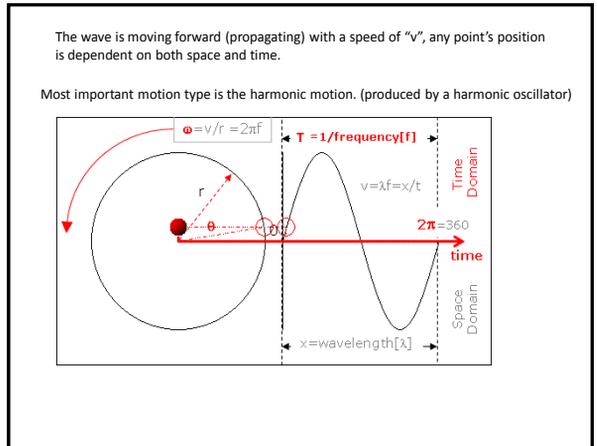
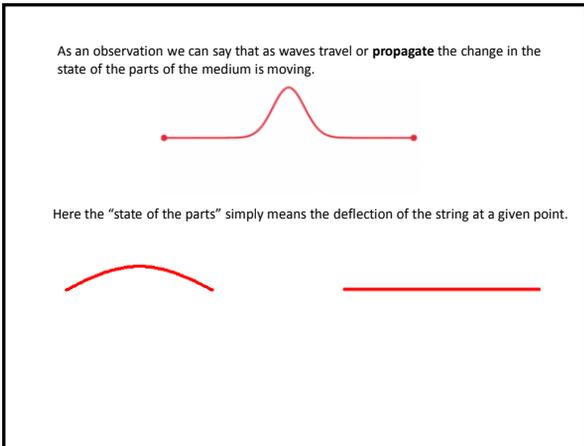
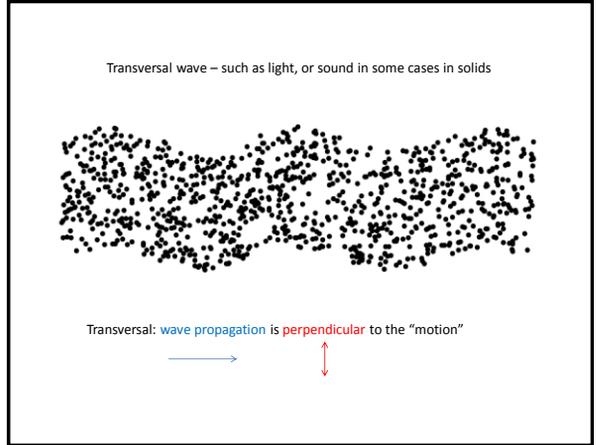
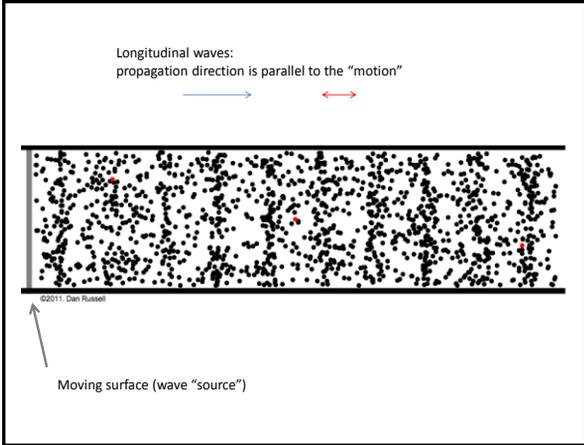
©2016, Dan Russell



Waves can be described by the wave equation, which relates the motion of individual parts of the medium to the observed wave.



It is important to note, that as the waves propagate, the parts of the medium (here the water molecules) stay "in place", which means there is no net transport of material.



The wave equation is a bit complicated:

$$\frac{\partial^2 u}{\partial t^2} = c^2 \frac{\partial^2 u}{\partial x^2}$$

We take the change of any property (here "u") in time (du/dt) and also in space (du/dx), but we need to take the change of the change (d²u/dx²), and these are linked by the propagation speed (or other named phase velocity) (here as "c").

A simple solution for u(x,t) is:

$$u(x,t) = A \cdot \sin(k \cdot x + \omega \cdot t + \phi)$$

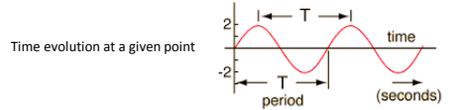
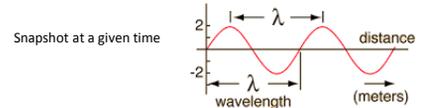
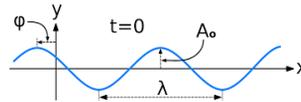
where

A is the amplitude of the wave, k is the wavenumber, and ω is the angular frequency

$\omega = 2\pi f$, where $f = 1/T$ [Hz], while T is the period time.

$\omega = c \cdot k$ defines the wavenumber, which can be written as $k = 2\pi/\lambda$. here λ is the wavelength.

Graphical representation of the solution



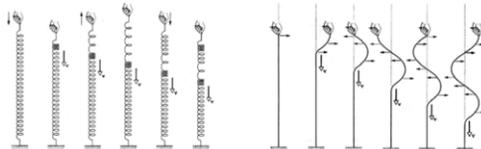
Different types of waves

- According to **source**:
 1. Mechanical: elastic deformation propagating through elastic medium
 2. Electromagnetic: electric disturbance propagating through space (vacuum)
- According to **propagation dimension**:
 1. One-dimensional (rope)
 2. Surface waves (pond)
 3. Spatial waves (sound)

- According to **relative direction of oscillation and propagation**:

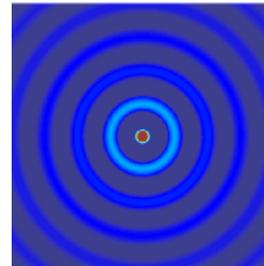
1. Longitudinal

2. Transverse

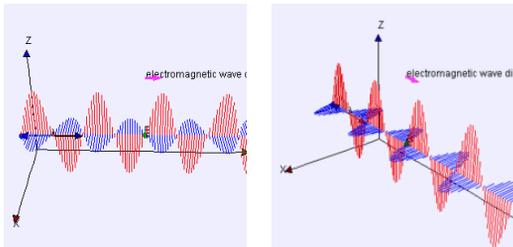


Point source: radiating in all directions along a sphere.

This is a transverse wave example, longitudinal is also possible.



Since we have interference, we must assume light is a wave.
 If so, then we have a wave equation for it. Since it is electro-magnetic, we have two oscillating quantities: electric field strength (**E**) and magnetic field strength (**B**).



For EM waves we have two equations, and the wave can travel to x,y,z directions, so the equations are a bit even more complicated:

$$\frac{\partial^2 \mathbf{E}}{\partial t^2} - c_0^2 \cdot \nabla^2 \mathbf{E} = 0$$

$$\frac{\partial^2 \mathbf{B}}{\partial t^2} - c_0^2 \cdot \nabla^2 \mathbf{B} = 0$$

Here the ∇ means the $d^2/d...^2$ in all directions

The solution is again a sine or cosine wave:

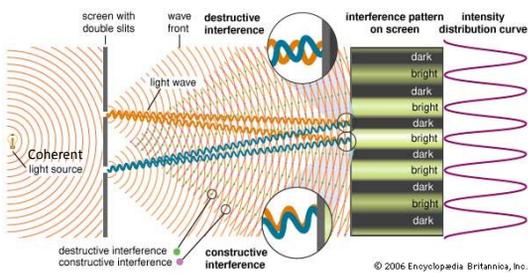
$$\mathbf{E}(\mathbf{r}, t) = \mathbf{E}_0 \cos(\omega t - \mathbf{k} \cdot \mathbf{r} + \phi_0)$$

$$\mathbf{B}(\mathbf{r}, t) = \mathbf{B}_0 \cos(\omega t - \mathbf{k} \cdot \mathbf{r} + \phi_0)$$

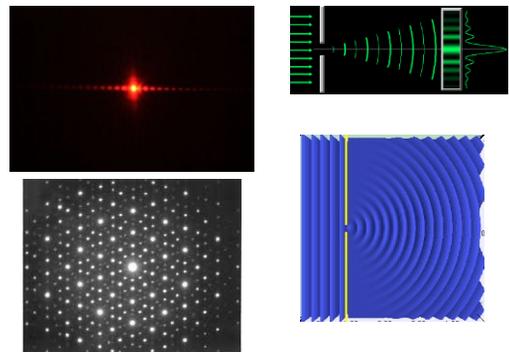
At any point of the observation we have to add all of the incoming sine waves, and that gives the net value of **E** and **B**.

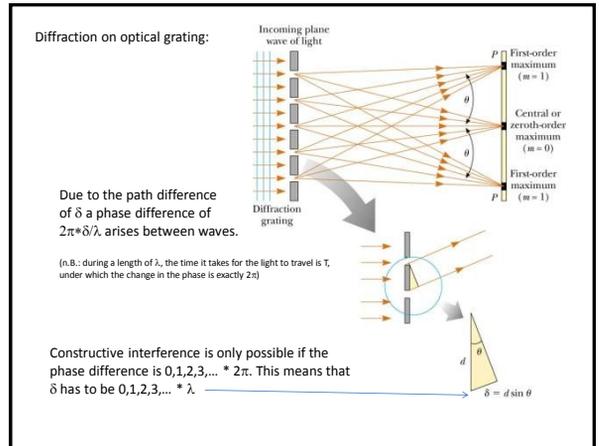
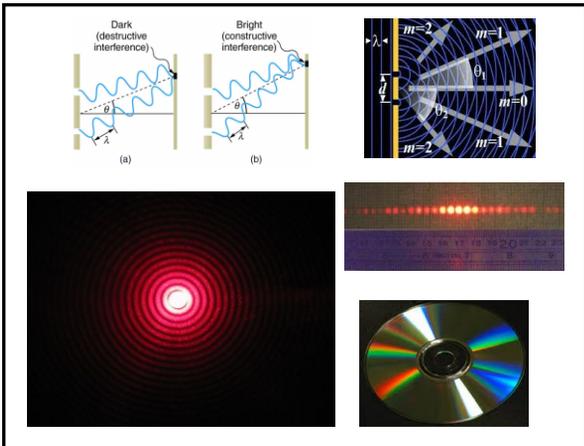
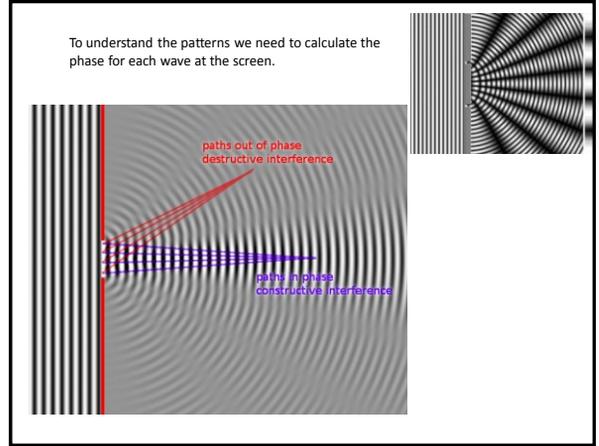
Remember: incoherent waves add up to practically 0, while coherent ones can add up from 0 to a maximum, depending on the phase difference.

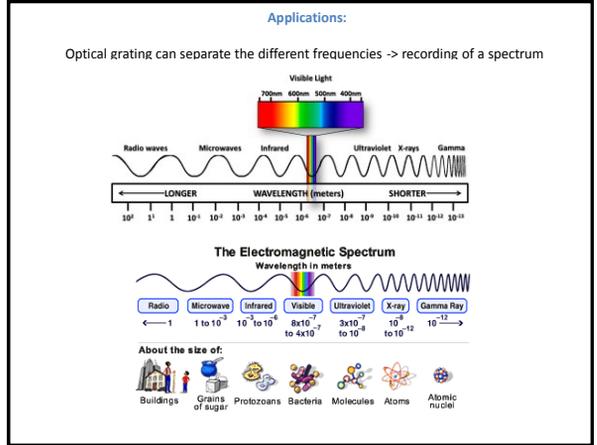
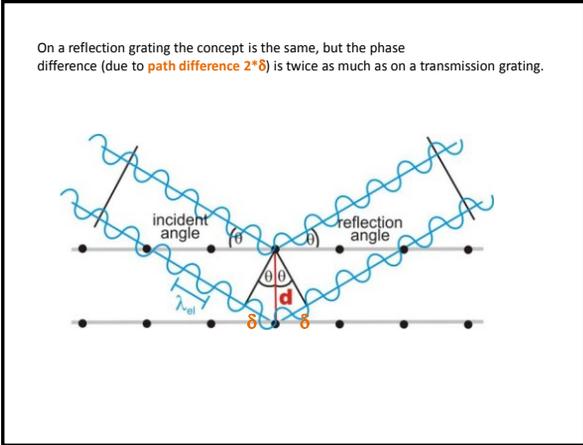
The explanation of the periodic profile : interference of waves, Huygens principle.



Diffraction and interference patterns with coherent light







Resolving power of a microscope is limited by the wave's diffraction.

Abbe's principle:
There is **only** an image formation in the microscope if at least the first order diffracted waves enter the objective lens.

„Central Beugung“
(that is only parallel, incident light, no condenser applied)

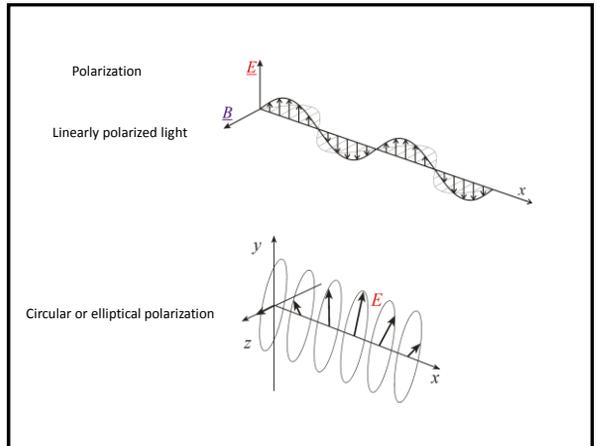
The Airy Disk and Point-Spread Function

(i) Airy Disk
The Airy Disk and Point-Spread Function

(ii) Point Source
Point-Spread Function

$$d = \frac{\lambda}{2 \sin \alpha}$$

Ernst Abbe (1840-1905)



Summation of two waves with different linear polarization

Summation if there is a phase difference, now 90 degrees

The amplitudes are the same, so the resulting electrical field vector rotates in time. The direction of the rotation depends only on the phase difference (+ or - 90 deg.)

We can even sum two circularly polarized lights:

Application: polarimetry.
Chiral molecules turn the polarization direction when light passes through a solution, it can be used to determine the concentration of the solution. (see lab practice of polarimetry)

For more animations, go to: <http://cddemo.szilab.org/>

Polarization microscopy

Colors: due to the dependence of specific rotation on the wavelength of light.

Polarized Light Microscope Optical Configuration Schematic Diagram

- Eye
- Eyepiece (Ocular)
- Analyzer
- Compensator
- Objective
- Specimen
- Condenser
- Polarizer
- Mirror
- Lamp
- Brinford Lens
- Slits
- A1
- S1

Phase contrast microscopy

The phase object is transparent, but the refractive index is different, so the speed of light is different, which builds up a phase difference during the way through the object. This is not observable to the human eye, but can be made visible with the phase-contrast microscope, which creates intensity changes from the phase changes.

It creates the contrast (amplitude change) by coherently re-interfering a reference, or surround beam (S), with a diffracted beam (D) from the specimen. The S beam is shown in yellow, and the D beam is shown in blue.

Diffraction is also present in the human eye:

Because of diffraction: image of a point object is an Airy disk

Rayleigh criterion: objects may be resolved if their corresponding Airy disks do not overlap

Smallest resolved distance has a limit (Abbe equation):

$$d = \frac{0.61 \lambda}{n \sin \alpha}$$

λ = wavelength
 n = refractive index of medium
 α = angle between axis and outermost ray

For a detailed explanation, and more figures: see the "optics of the eye" lab manual!

There are some experiments, observations which still can not be explained alone by the wave theory

Photoelectric effect
 Quantum dots
 Fluorescence
 Lasers
 Black body radiation

-> light is a wave AND a particle (this leads us to quantum physics)

The photoelectric effect:

If we shine light onto some material (often metals) then we might observe the release of an electron. This however **does NOT depend on the intensity of the light, but instead solely on the frequency!**

$E_{\text{photon}} = h\nu$

700 nm 1.77 eV
 550 nm 2.25 eV
 400 nm 3.1 eV
 no electrons
 Photoelectron - 2.0 eV needed to eject electron
 $v_{\text{max}} = 6.22 \times 10^5 \text{ m/s}$
 $v_{\text{max}} = 2.96 \times 10^5 \text{ m/s}$

From simple waves, since energy is flowing in a wave, after a given time the electron should have gathered the required energy to split free of the atom. If we increase the intensity, so the time required for this would become shorter, and we would see more electrons coming out of the surface in a given time-range. Instead, we see absolutely no electrons independently of the intensity at some wavelength, but some electrons at a shorter wavelength. This contradicts the wave theory.

Solution (Einstein): The energy flow in light is not a smooth constant flow, but instead energy is distributed in "packets" of $\epsilon = h \cdot f$, where h is the Planck's constant, and f is the frequency.

This theory says, that while light often "behaves" as a wave, it also has some type of particle nature, which is manifested in the existence of a finite energy, which is closely related to the frequency of the wave. This energy is called the "photon energy", and the term "photon" reflects the particle-type of behavior of light.

The particle-type is also manifested in the impulse of the photon, although it has no resting mass.

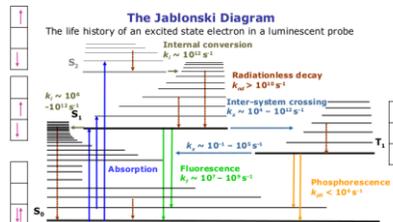
$$E = \hbar\omega = h\nu = \frac{hc}{\lambda}$$

$$p = \hbar k,$$

Where $h = h/2\pi$

$$p = \hbar k = \frac{h\nu}{c} = \frac{h}{\lambda}$$

Fluorescence, phosphorescence, lasers also need the photon picture, if we want to understand what is happening. (this will come later)



• Fluorescence is observed if $k_f \gg k_i + k_{isc}$

- The time a molecule spends in the excited state is determined by the sum of the kinetic constants of all deexcitation processes