

# Medical Biophysics I.

Interactions of light with matter.

Light scattering, light absorption. Basic principles of absorption spectrometry.

(Help to understand and learn exam question 12. and to measurement „Light absorption“.)

8<sup>th</sup> Lecture, 25<sup>th</sup> October 2016

# 12<sup>th</sup> exam question

How the light can interact with atoms and molecules?

*Basic concepts:* Light scattering, Rayleigh-scattering, Mie-scattering, light absorption.

*Phenomena:* Blue sky, white and grey clouds, transparency.

*Physical quantities:* Light intensity (J), electric dipole momentum (p), absorbance (A).

*Laws, relations and the way leading to them:* Attenuation of light intensity passing through medium. Dependence of scattered light intensity on the wavelength. Lambert–Beer-law.

*Applications:* Measurement of static light scattering and absorption, absorption spectrometry, determination of concentration. Dark field microscope.

## Repetition I. - Light

What is light? – visible electromagnetic wave

Wave: quantity(ies) changing periodically in SPACE and TIME

Electromagnetic wave:

electric field (E) and magnetic flux density („field“ (B)) -||-

$$\vec{F} = q \cdot \vec{E}$$

$$\vec{F}_L = q \cdot \vec{v} \times \vec{B}$$

Electric field is defined as the electric force per unit charge

Magnetic fields are produced by electric currents (moving charges), which can be macroscopic currents in wires, or microscopic currents associated with electrons in atomic orbits

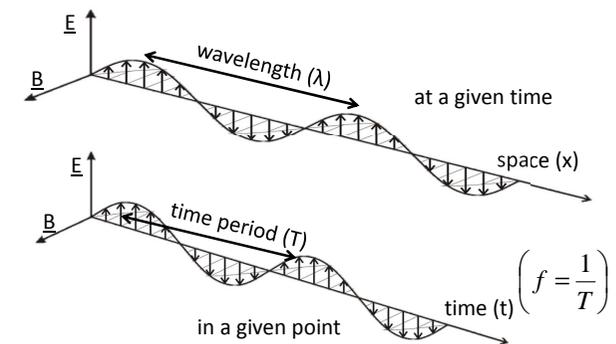
## Repetition I. - Light

$$E = A_E \cdot \sin\left(\frac{2 \cdot \pi}{\lambda} \cdot x + \varphi_0\right)$$

$$B = A_B \cdot \sin\left(\frac{2 \cdot \pi}{\lambda} \cdot x + \varphi_0\right)$$

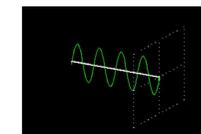
$$E = A_E \cdot \sin\left(\frac{2 \cdot \pi}{T} \cdot t + \varphi_0\right)$$

$$B = A_B \cdot \sin\left(\frac{2 \cdot \pi}{T} \cdot t + \varphi_0\right)$$

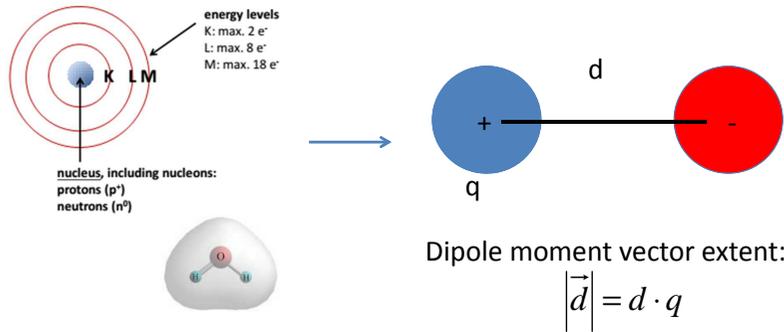


TIME and SPACE:  $c = \lambda \cdot f$

$$\varepsilon = h \cdot f \quad E = \sum N \cdot \varepsilon_\lambda \quad J = \frac{E}{A \cdot t}$$



## Repetition II. Matter: Atoms and Molecules

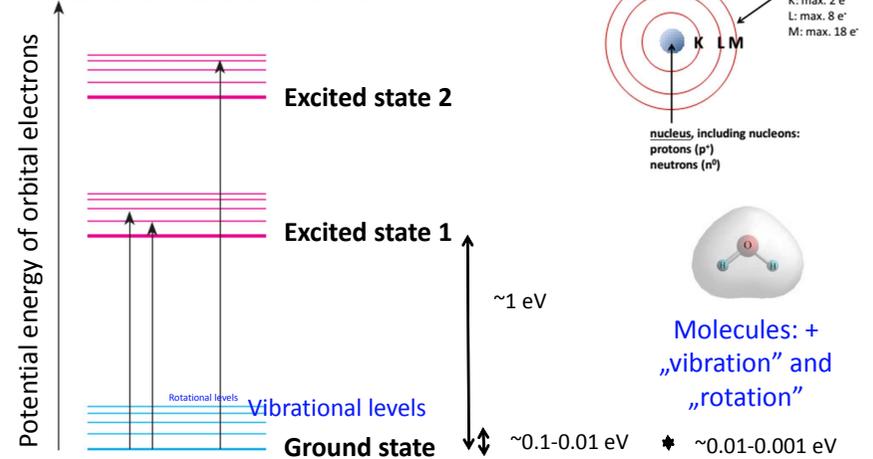


Polarizability ( $\alpha$ ): extent of a dipole that is generated by a unit of electric force-field

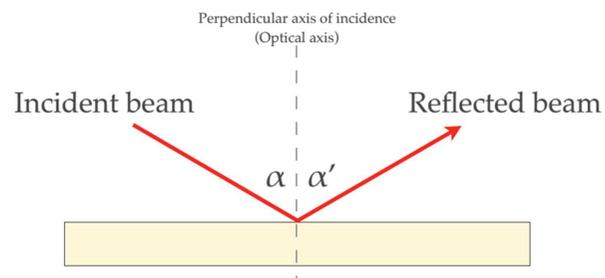
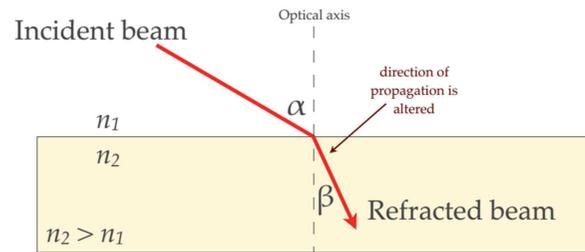
$$\vec{d} = \alpha \cdot \vec{E}$$

## Repetition II. Atoms and Molecules

Electronic and vibrational energy levels of atoms and molecules



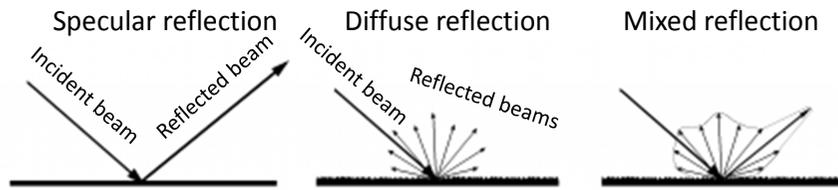
## InteractionS – repetition



## InteractionS – repetition

Experiment 1.

# InteractionS – Reflection



Without ( $< \lambda$ ) irregularity



With irregularity



# InteractionS – Reflection

Coefficient of reflection

$$\rho_\lambda = \frac{J_{\text{reflected}}(\lambda)}{J_{\text{incident}}(\lambda)}$$

depends on:  
wavelength  
incident angle  
material (refractive indices ...)

$$\rho_\lambda = \frac{J_{\text{reflected}}(\lambda)}{J_{\text{incident}}(\lambda)} = \left( \frac{n_2 - n_1}{n_2 + n_1} \right)^2$$

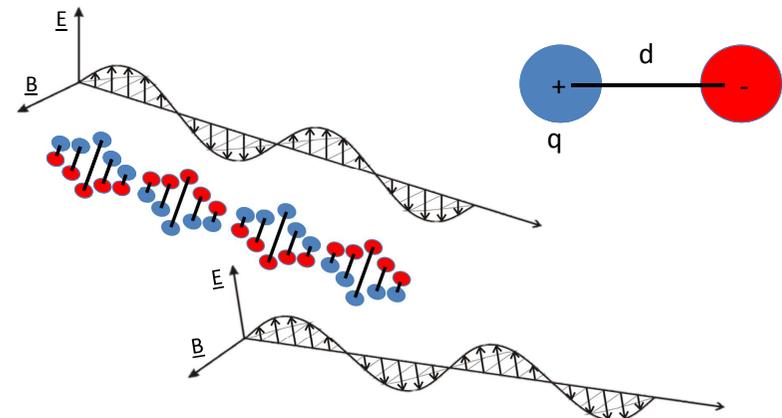
equation is true **IF** incident angle is  $0^\circ$ ,  
material is transparent (see later)

# InteractionS

Experiment 2.

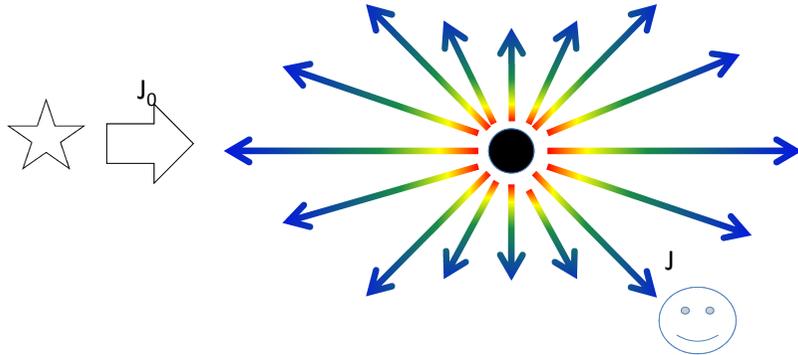
# InteractionS – Scattering

1. Elastic scattering ( $f$  does NOT change)



## InteractionS – Rayleigh Scattering

Small, individual particles – individual oscillations without interference

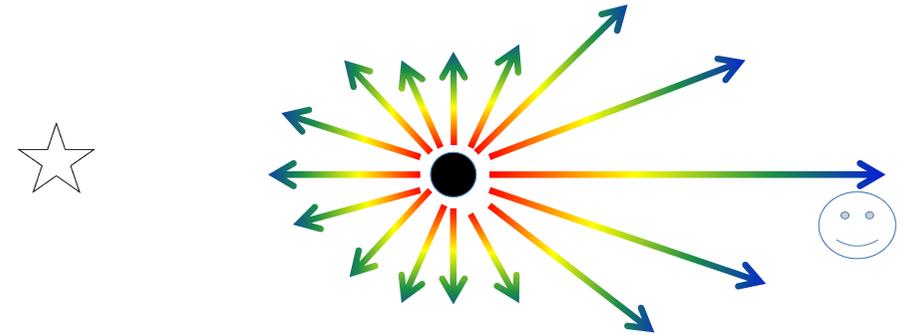


$$J_{\lambda} = J_0 \cdot \frac{8 \cdot \pi^4 \cdot N \cdot \alpha^2}{\lambda^4 R^2} \cdot (1 + \cos^2(\beta))$$

N: number of scatterers  
 $\alpha$ : polarizability  
 $\lambda$ : wavelength  
 R: distance from scatterer  
 $\beta$ : observing angle

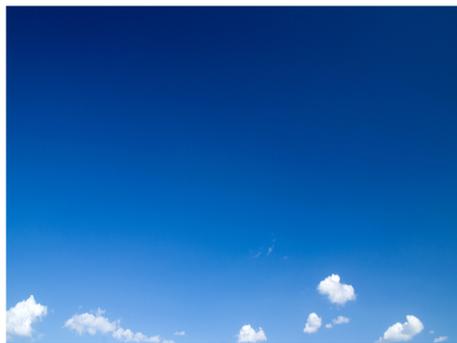
## InteractionS – Mie Scattering

Large particles, group of atoms, molecules (size >  $\lambda/10$ )  
 – constructive interference



wavelength independent  
 large forward scattering  
 Size dependent:  
 Size increasing: intensity increasing  
 Size increasing further (>  $\lambda \cdot 10$ ): intensity decreasing

## InteractionS – Rayleigh Scattering



Blue sky  
 Red sunset

## InteractionS - Mie scattering



Glare, white cloud

Gray cloud

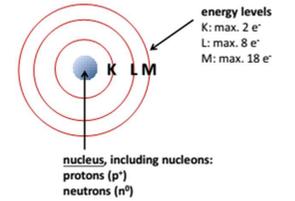
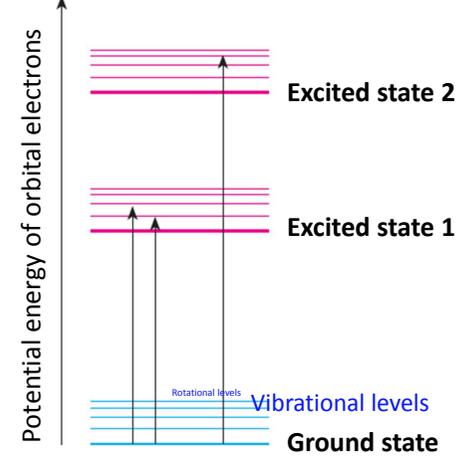
# Interactions

Experiment 3.

# Interactions - Absorption

**Resonance:** induced oscillation – energy levels

Electronic and vibrational energy levels of atoms and molecules



Molecules: + „vibration” and „rotation”

# Interactions - Scattering

## 2. Inelastic scattering ( $f$ changes)

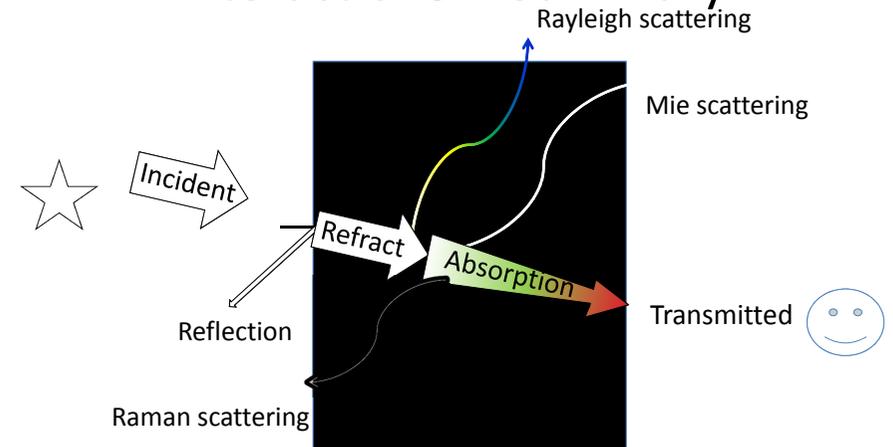
– energy absorption between vibrational levels + re-emission  
short life time in the excited state ( $\sim$ fs)

Raman scattering

Stokes: scattered  $\epsilon$  reduced

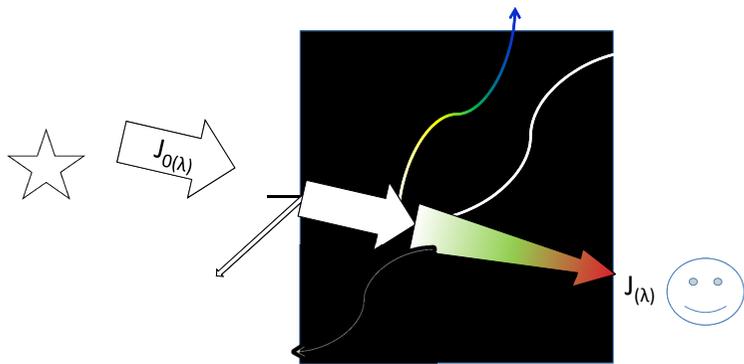
Anti-Stokes: scattered  $\epsilon$  increased

# Interactions - Summary



Which one? – Probabilities!  
depends on: wavelength  
incident angle  
material

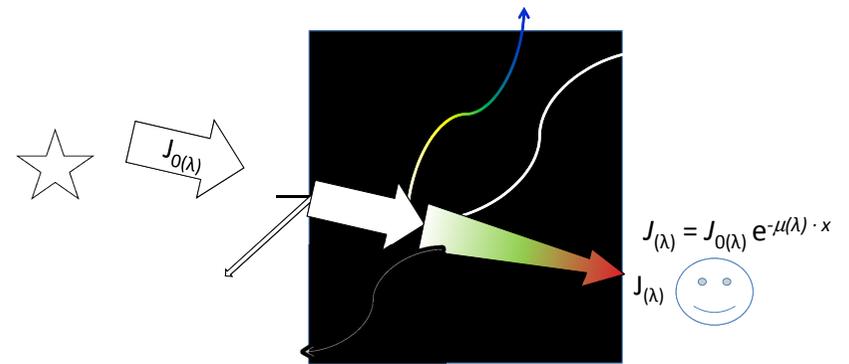
## Intensity attenuation - Decay law



$$J(\lambda) = J_{0(\lambda)} e^{-\mu(\lambda) \cdot x}$$

J: intensity  
 $\mu$ : attenuation coefficient  
 x: layer thickness

## Transmittance, Optical density

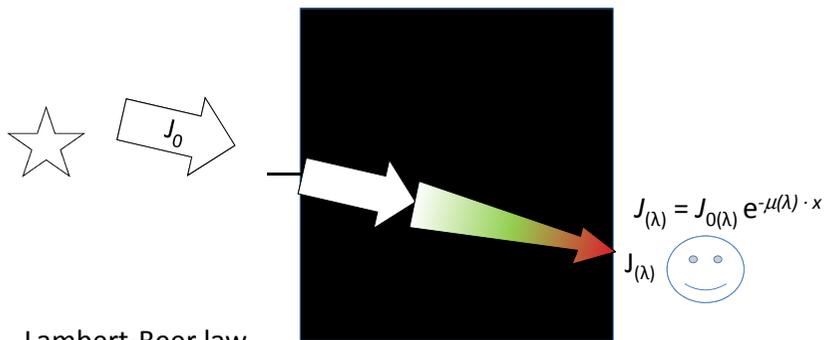


$$T = \frac{J(\lambda)}{J_{0(\lambda)}} (\%)$$

T: transmittance  
 OD: optical density

$$OD = \log \frac{J_{0(\lambda)}}{J(\lambda)} = \log e \cdot \mu(\lambda) \cdot x$$

## Absorption



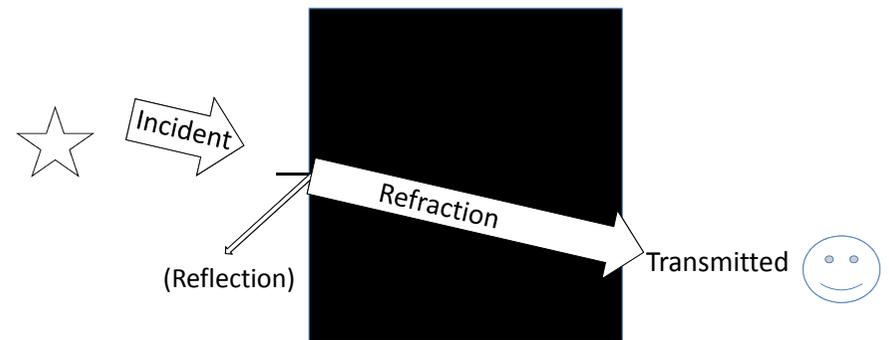
Lambert-Beer law

$$A = \log \frac{J_{0(\lambda)}}{J(\lambda)} = \log e \cdot \mu(\lambda) \cdot x = \epsilon_{(\lambda)} \cdot c \cdot x$$

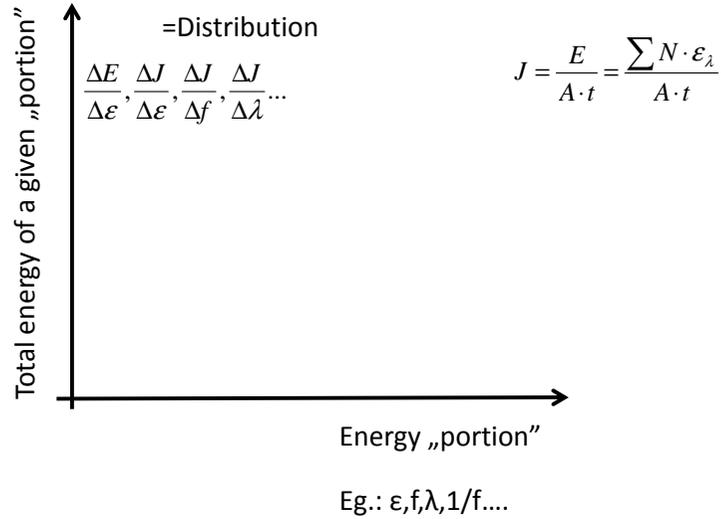
Conditions!!!

$\mu$ : absorption coefficient  
 $\epsilon$ : (decadic) molar extinction coefficient  
 c: concentration

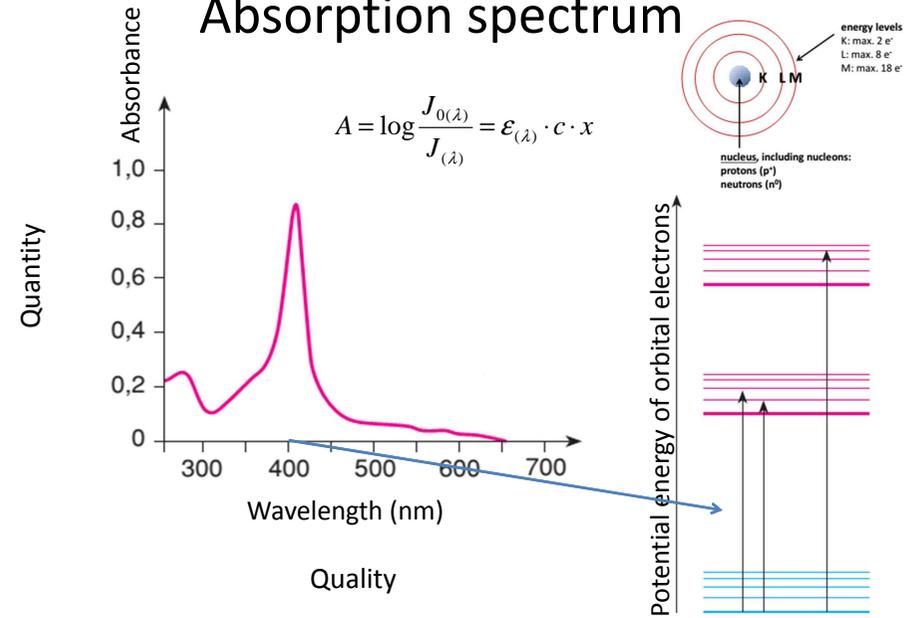
## Transparency



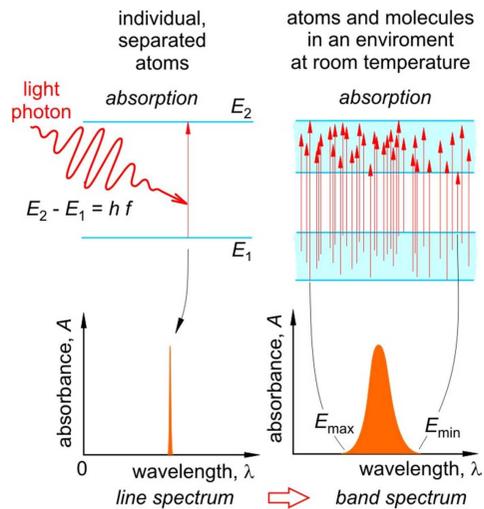
# Spectrum



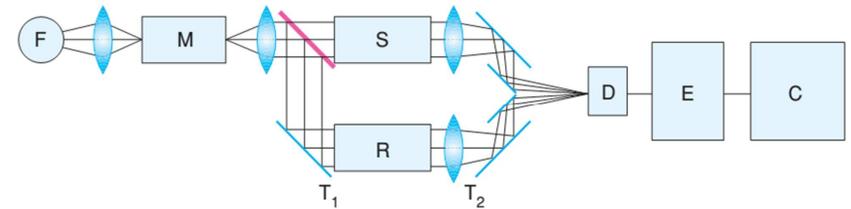
# Absorption spectrum



# Spectrum broadening



# Absorption spectrum measurement



## Monochromators:

Prism: refraction – advantage: total intensity

Optical grating: diffraction (interference image) – advantage: linear („ghosts”: harmonics)

# Scattering measurements

Concentration measurements:

Turbidimetry: (high intensity of scattering)  
measuring OD (condition: no absorption)

Nephelometry: (low intensity of scattering)  
collecting scattered light in a given angle

Particle size and geometry measurement:

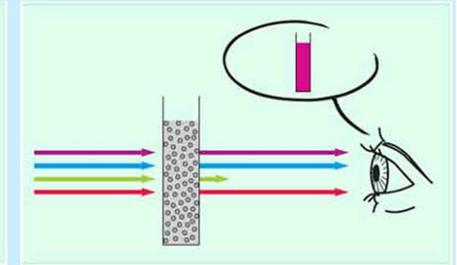
Dynamic light scattering: (size)  
collecting scattered light in a given angle  
scattered intensity changes in time

Static light scattering: (size, geometry)  
collecting scattered light in a given angle range (multiple angle)

# Color - absorption

Complementary colors

Wavelength [nm]	Absorbed color	Complementary color
650-780	red	blue-green
595-650	orange	greenish blue
560-595	yellow-green	purple
500-560	green	red-purple
490-500	bluish green	red
480-490	greenish blue	orange
435-480	blue	yellow
380-435	violet	yellow-green



The color of matter is related to its **absorptivity**, reflectivity and light scattering. The human eye sees the complementary color to that which is absorbed

# Colors

