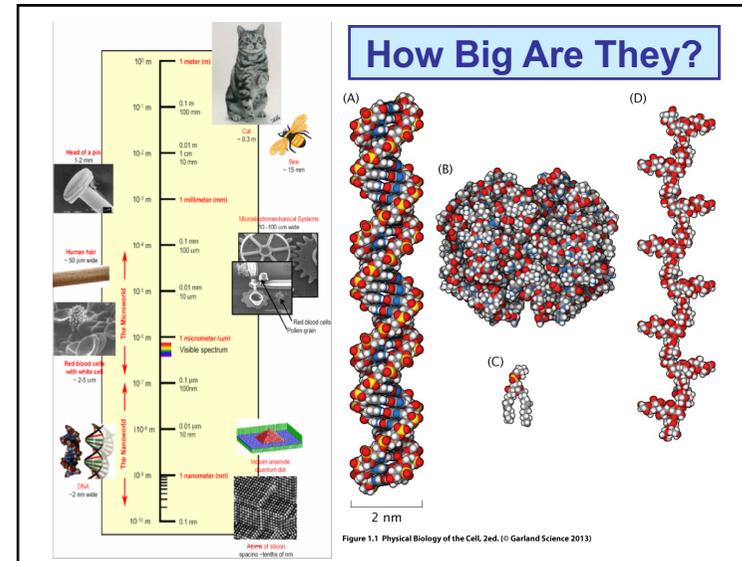


Formation of Biological Structures

Szabolcs Osváth

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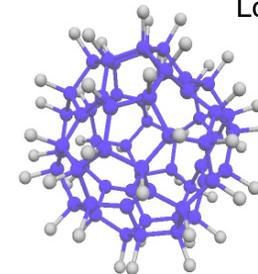
“Plenty of Room at the Bottom”

" The principles of physics, as far as I can see, do not speak against the possibility of maneuvering things atom by atom. It is not an attempt to violate any laws; it is something, in principle, that can be done; but in practice, it has not been done because we are too big."

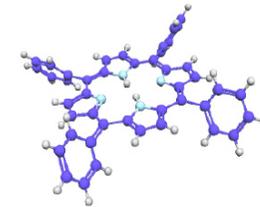
Richard Feynman, 1959

Wave – Particle Duality

Louis De Broglie: $\lambda = h/p$



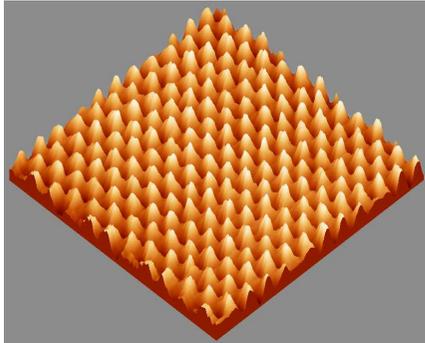
fluorofulleren
 $C_{60}F_{48}$
1632 Da



tetraphenilporfirin $C_{44}H_{30}N_4$

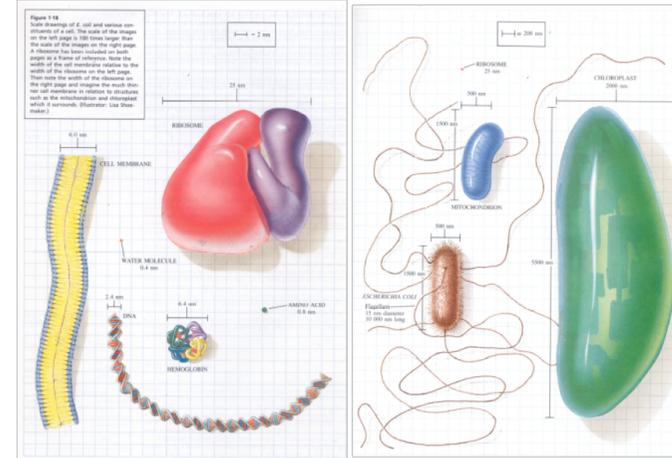
L Hackermuller, S Uttenthaler, K Homberger, E Reiger, B Brezger, A Zeilinger, M Arndt; Phys. Rev. Lett. 91 (2003) 90408

Wave – Particle Duality



Scanning Tunneling Microscope (STM) image

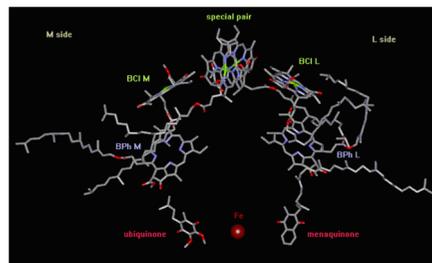
The Size Range of Life Processes



Structure – Function Relationship

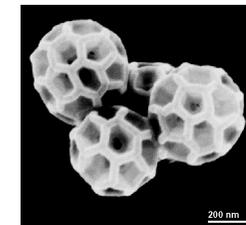
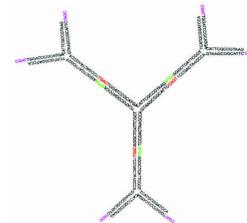
From the molecular level to ecosystems, there is a strong relationship between structure and function of biological systems.

Hartmut Michel, Johann Deisenhofer, Robert Huber
1982 – 3D structure of the bacterial photosynthetic reaction center
1988 – Nobel prize

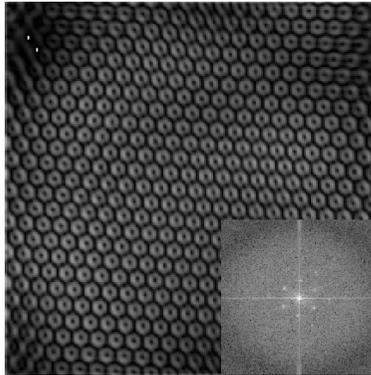


Self Assembly

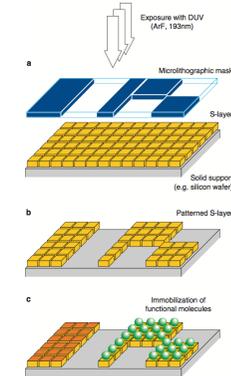
Molecular recognition
(e.g. self assembly of DNA molecules into „balls”).



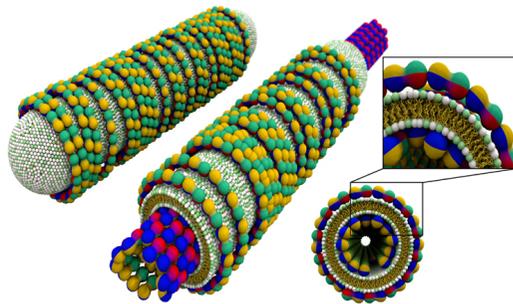
2D Order S-layer Protein (TEM Image)



Lithography of Nanostructures Using S-layer Protein

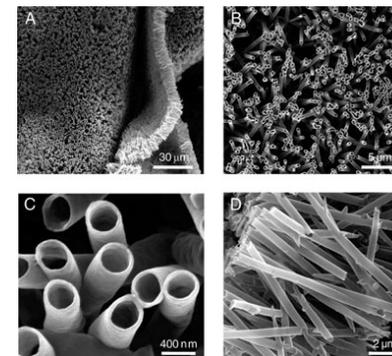


Nanotubes Built of Tubulin and Lipids as Drug Vehicles

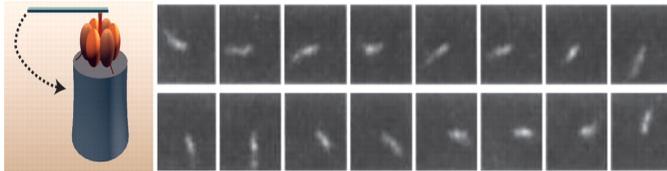


HSA Nanotubes

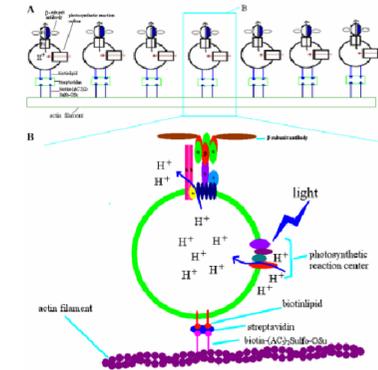
Nanotubes of Human Serum Albumin (HSA) grown on nanoporous polycarbonate membrane



F₁-ATPase Driven Nickel Propeller



Light - Driven Swimming Structure



Biopolymers

reaction	t _{1/2} @ 25 °C	t _{1/2} @ 100 °C	typical number of monomeric units in a polymer molecule	number of different monomers
DNA hydrolysis	140 000 years	22 years	3 · 10 ⁹ (human DNA)	4
RNA hydrolysis	4 years	9 days	few dozen (tRNA)	4
protein hydrolysis	400 years	5.5 weeks	few hundred	20

Biopolymers

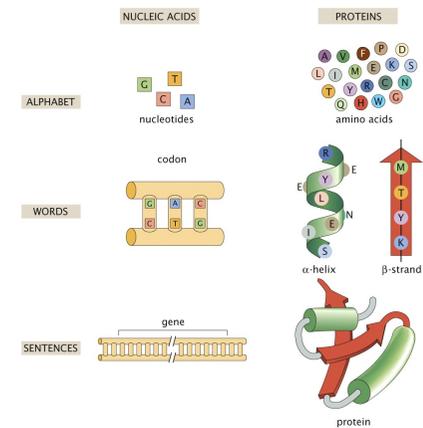
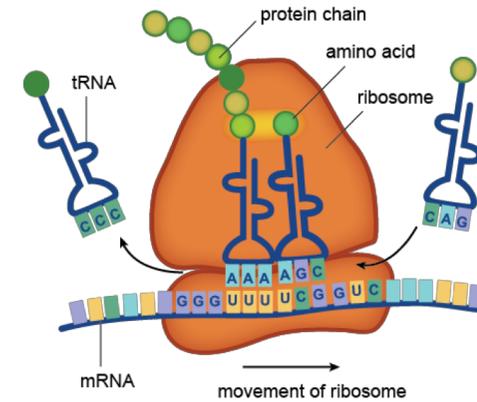


Figure 1.2 Physical Biology of the Cell, 2ed, (© Garland Science 2013)

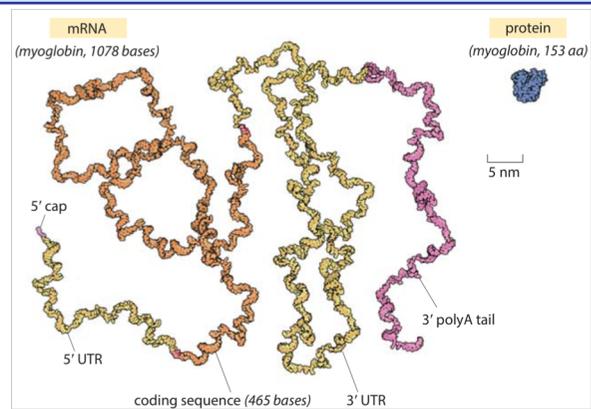
Role of RNA in Living Systems

- messenger (mRNS)
- ribosomal (rRNS)
- transfer (tRNS)
- regulator
- enzyme (ribozim)
- switch (riboswitch)
- virus gene RNS

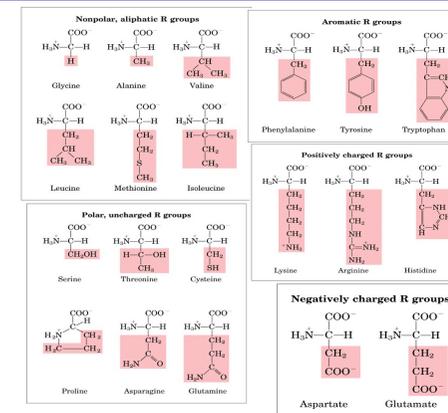
Ribosome Function



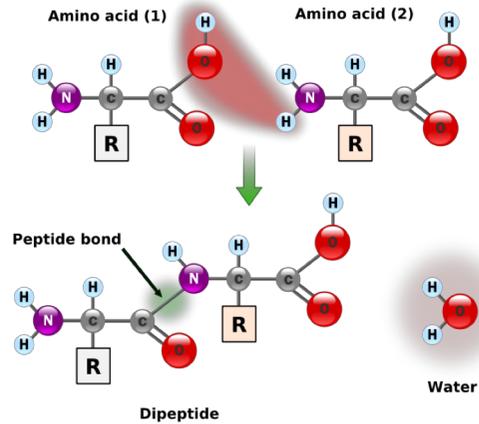
Relationship Between RNA Code and Protein Sequence



The Twenty Standard Amino Acids



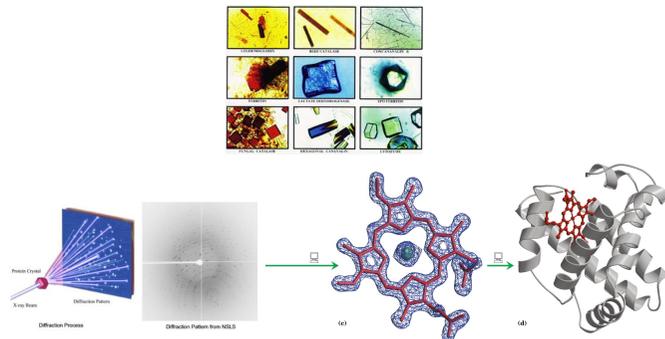
The Peptide Bond



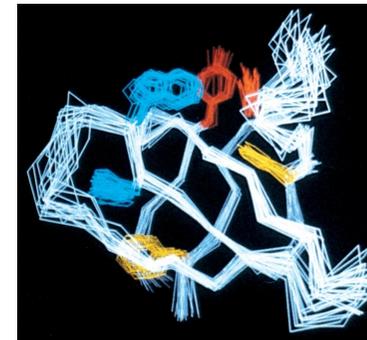
Role of Proteins in Living Systems

- chemical catalysis
- transport
- energy conversion and storage
- coordinated movement
- mechanical skeleton
- immune response
- molecular recognition
- passing information
- gene regulation
- growth and differentiation

X-ray Crystallography



NMR Structure Determination



NMR structure of the 64 amino acid SH3 domain of the Src protein

Interactions Stabilizing the Native State

short range repulsion

Van der Waals interaction

electrostatic interaction

hydrogen bonding

hydrophobic interaction

disulfide bridge

Short Range Repulsion

Due to the exchange (Pauli) interaction, at short distances there is a strong repulsion between electrons.

The potential energy of the repulsion increases quickly with decreasing distance ($\sim 1/r^{12}$).

Atoms can be considered hard spheres with a given radius (Van der Waals radius).

Van der Waals Interaction

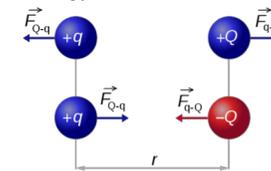
Occurs between any two atoms due to the interaction of induced dipole moments.

Dependence on the distance of the interaction energy: $\sim 1/r^6$

Electrostatic Interaction

Distance dependence of the interaction energy of the Coulomb force:

$$E = \frac{q \cdot Q}{4\pi \cdot \epsilon_0 \cdot \epsilon_r \cdot r}$$



The relative dielectric constant inside the protein is approx. 4, and 80 in water.

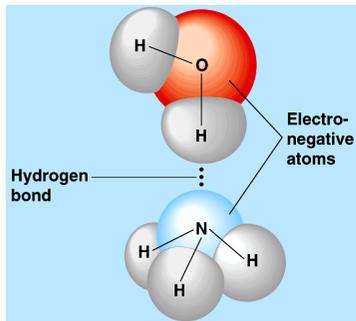
Salt bridges between ion pairs (Lys, Arg and Glu, Asp).

There is a large hydrate shell around charges in water.

Mobile ions can strongly shield charges.

Hydrogen Bonding

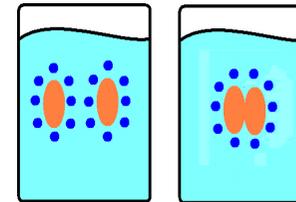
Attraction force between a H atom of a more electronegative atom or group (hydrogen bond donor) and another atom bearing a lone pair of electrons (hydrogen bond acceptor).



Hydrophobic Interaction

observed tendency of nonpolar surfaces to adhere in an aqueous solution and exclude water molecule

entropic effect originating from the disruption of hydrogen bonds of liquid water by the nonpolar solute



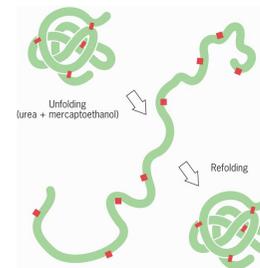
Disulfide Bridge

stabilizes the native structure
decreases the conformational entropy of the unfolded protein:

$$\Delta S = -2.1 \text{ J/K} - 1.5 \cdot R \cdot \ln n$$

n is the number of AAs between the two bonded AAs.

Anfinsen's Dogma



Refolding of Ribonuclease A

Christian B. Anfinsen

The information of the 3D protein structure is encoded in the 1 D AA sequence.

Importance of the Protein Folding Problem

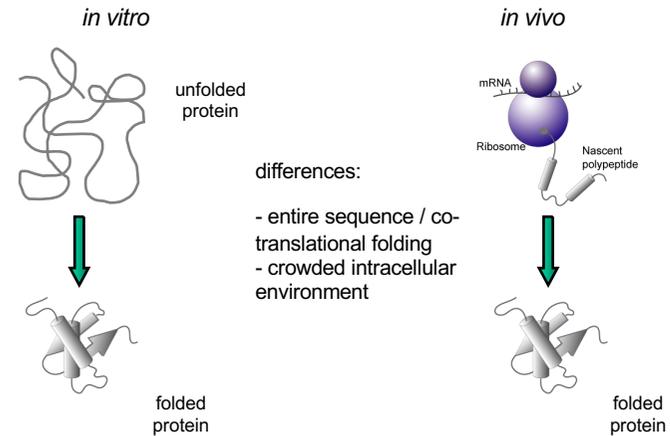
One of the most important questions of molecular biophysics.

We sequence genomes, we build databases, but we can't predict protein structure and function based on the genetic information.

There are roughly two dozen conformational diseases:

Misfolded proteins and deposition of amyloid plaques was observed in various diseases (pl. Creutzfeld-Jakob disease, Alzheimer disease, Parkinson disease).

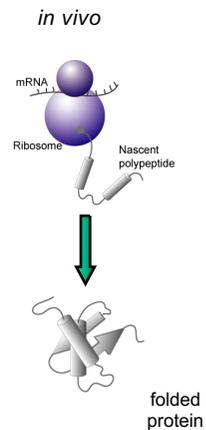
In vitro and *in vivo* Folding



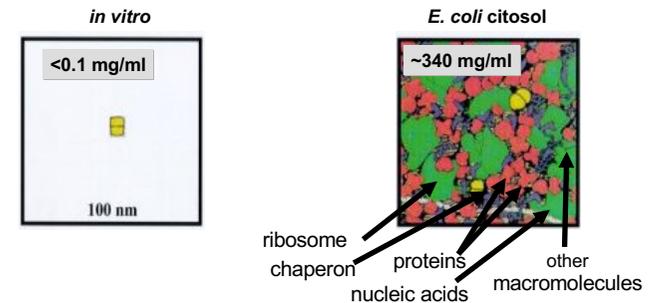
Co-Translational Folding

The N terminal of the nascent polypeptide chain starts to fold before completion of the translation.

20-30 AAs of the C terminal are protected within the ribosome.



Molecular Crowding



In vitro experiments

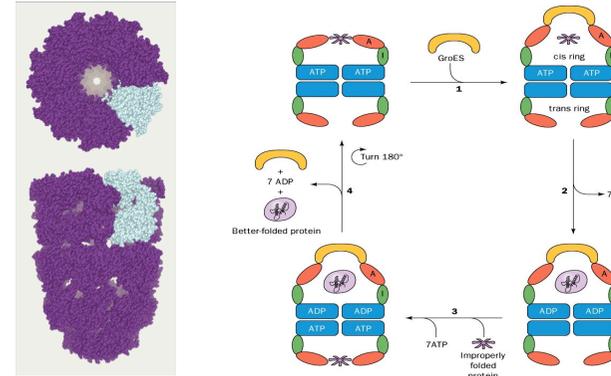
- lack of binding partner molecules
- lack of posttranslational modifications
- very different physico-chemical environment than in a cell

Effect of Molecular Crowding

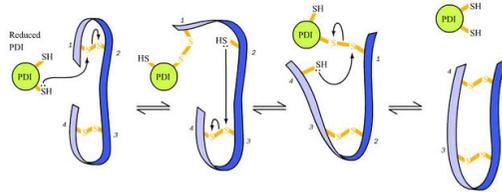
Molecular crowding:
a large fraction of the volume of the cytoplasm is filled by other molecules than water.

- dissociation constants decrease
- speed of protein-protein association increases
- association of fully or partially denatured proteins speeds up

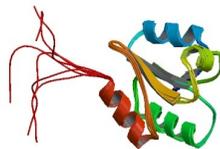
GroEL/ES Chaperon Cycle



Protein Disulfide Isomerase Function



structure of the human protein disulfide isomerase

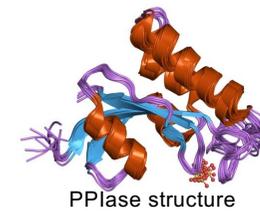
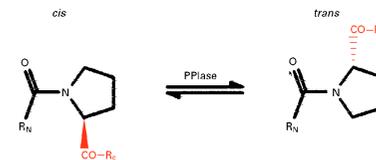


Proline Cis/Trans Isomerase

Due to the activation barrier between the cis and trans prolines, the presence of cis prolines in the native structure:

- speeds up early folding steps
- slows down the final formation of the native structure.

PPIase (peptidyl-prolyl isomerase)



Fate of the Protein in Eukaryotic Cells

cytosol protein synthesis and folding,
extracellular volume export of folded protein
mitochondrion limited protein synthesis
chloroplast limited protein synthesis
endoplasmic reticulum import of unfolded protein
peroxisome import of folded protein
nucleus import of folded protein
lysosome import of unfolded protein

Levinthal's Paradox - Calculation

Cyrus Levinthal

Consider a protein of 151 AAs. Assume all the 150 bonds connecting them have only two possible conformations. Assume that a reorientation of the bonds happens in 10^{-13} s.

A random search through the phase space would last:
 $2^{150} \cdot 10^{-13} \text{s} = 4.6 \cdot 10^{24} \text{years}$.

Age of Earth: $4.6 \cdot 10^9$ years

Age of the Universe: $13.7 \cdot 10^9$ years

Proteins typically fold on the ms to s timescale.

Levinthal's Paradox - Conclusion

The phase space of a protein is way too big to find the native structure by random search.

Cyrus Levinthal
1922 - 1990

