

Physical Biology of the Living Cell

Szabolcs Osváth

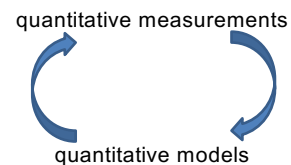
Semmelweis University

Topics – Semester I (<http://biofiz.semmelweis.hu/>)

Qualitative and quantitative modelling in biology (Dr. Szabolcs Osváth)	Feb. 11.
Structural hierarchy of proteins (Dr. László Smeller)	Feb. 18.
Stability of biological structures (Dr. László Smeller)	Feb. 25.
Formation of biological structures (Dr. Szabolcs Osváth)	Mar. 3.
Experimental methods to study biological structures - I (Dr. László Smeller)	Mar.10.
Experimental methods to study biological structures - II (Dr. Miklós Kellermayer)	Mar.17.
Microscopy studies of intracellular structures (Dr. Miklós Kellermayer)	Mar.24.
Super-resolution microscopy (Dr. Szabolcs Osváth)	Mar. 31.
Dynamic intracellular protein structures (Dr. Miklós Kellermayer)	Apr. 21.
Single molecule biological activity (Dr. Miklós Kellermayer)	Apr. 28.
Visit to the research laboratories of the Dept. of Biophysics and Radiation biology of the Semmelweis University (Dr. Szabolcs Osváth)	May 5.
Problem solving and consultation (Dr. Szabolcs Osváth)	May 12.

Science

science is humanity's
endeavor to know and
understand the world



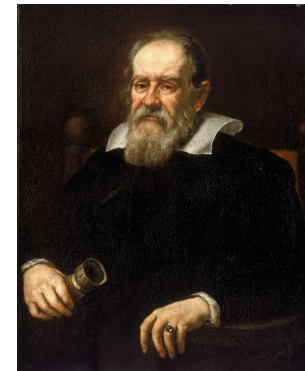
the seven hundred-year-old linden
from Szőkedencs

Mathematics

„ ... “Mathematics is the
language with which God has
written the universe.”

Advantages of applying math:

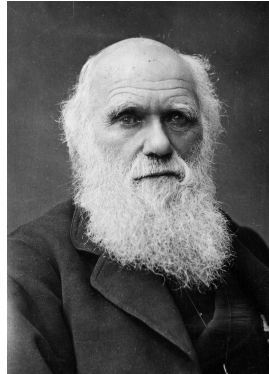
- accurate predictions
(comparison with more
sophisticated measurements)
- strict derivations
- abstract thinking



Galileo Galilei (1564 – 1642)

The role of mathematics

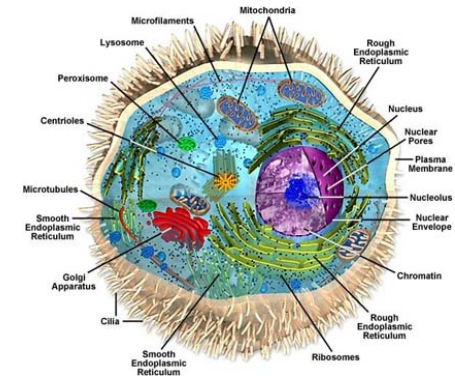
„... in after years I have deeply regretted that I did not proceed far enough at least to understand something of the great leading principles of mathematics, for men thus endowed seem to have an extra sense.”



Charles Darwin (1809 – 1882)

The role of mathematics

cell biology achieved great success without the use of mathematics



The role of mathematics

accumulated large amounts of observations of stars and planets



Tycho Brahe (1546 – 1601)

The role of mathematics

He recognized the laws of planetary motion - he created a mathematical model.



Johannes Kepler (1571 – 1630)

Premises of model building

Is the problem interesting or important?

Certainty level of "facts"

- Facts observable by anyone
(e.g., the cell contains proteins)
- Facts accepted after extensive experimental testing
(e.g., proteins are synthesized on the ribosome)
- Speculative statements
(e.g., mitochondria are descendants of bacteria)

A fundamental idea: Biological entities must not violate the laws of physics and chemistry.

Why is life alive?

Central properties

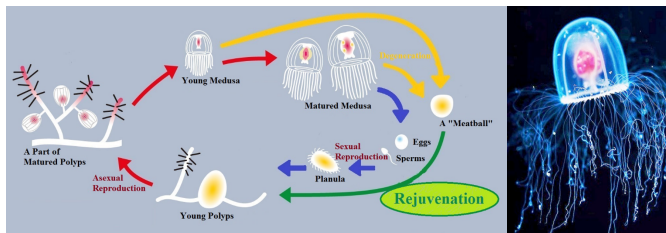
- growth, energy utilization, reproduction, death

The living cell is built of surprisingly few elements

The cell contains structurally and functionally specialized macromolecules

- proteins, nucleic acids, carbohydrates, lipids
- macromolecules - combinatorial assembly of units

Turritopsis dohrnii



The oldest living creatures



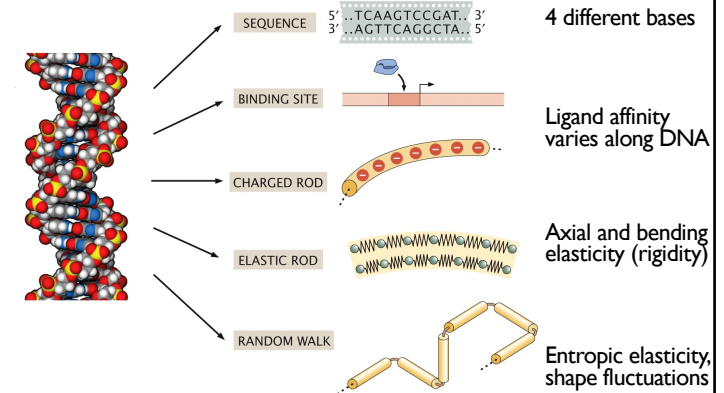
"Methuselah", 4,851-year-old
Bristlecone Pine
(*Pinus longaeva*)

Pando 80,000-year-old
Populus tremuloides
(quaking aspen) forest

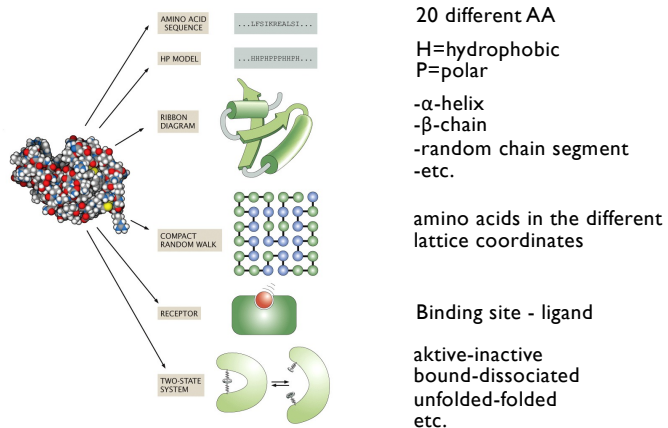
Biological model building

- Abstraction
- Simplification
- We cannot attain a complete atomic description of the macromolecules
- Projections are made, which reflect a certain property of the macromolecule
- Idealization

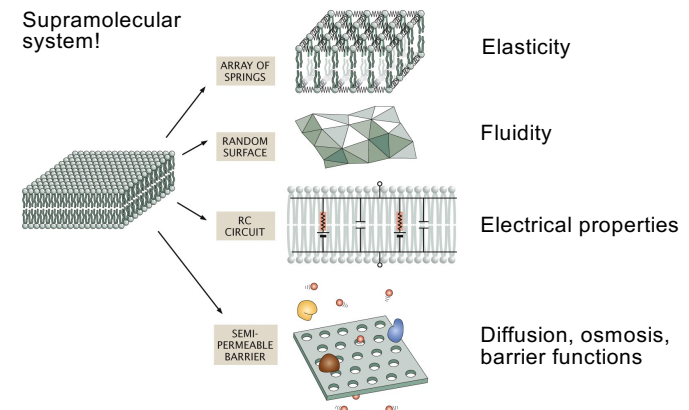
Idealization of the DNA molecule

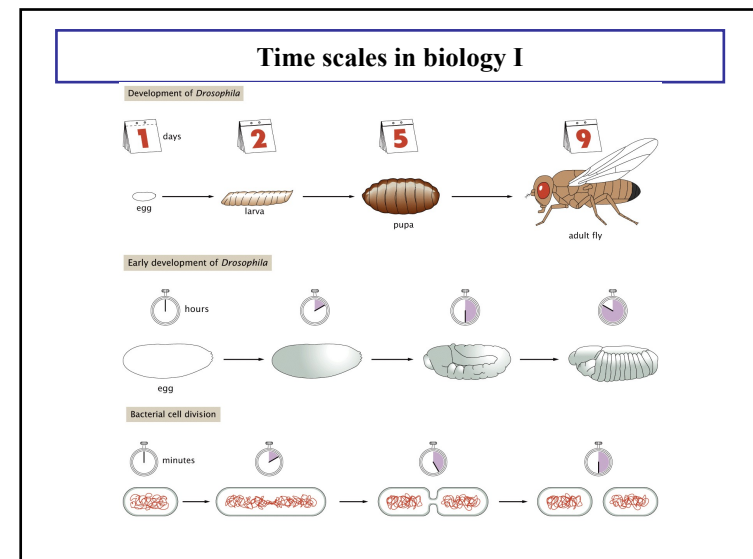
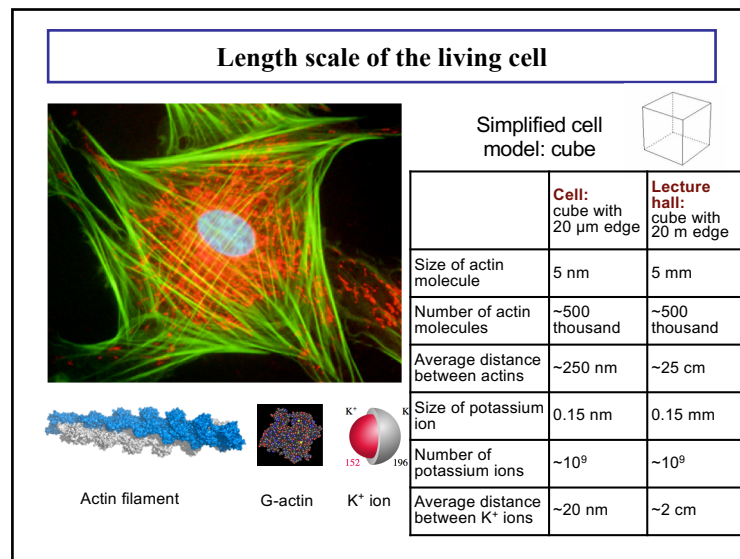
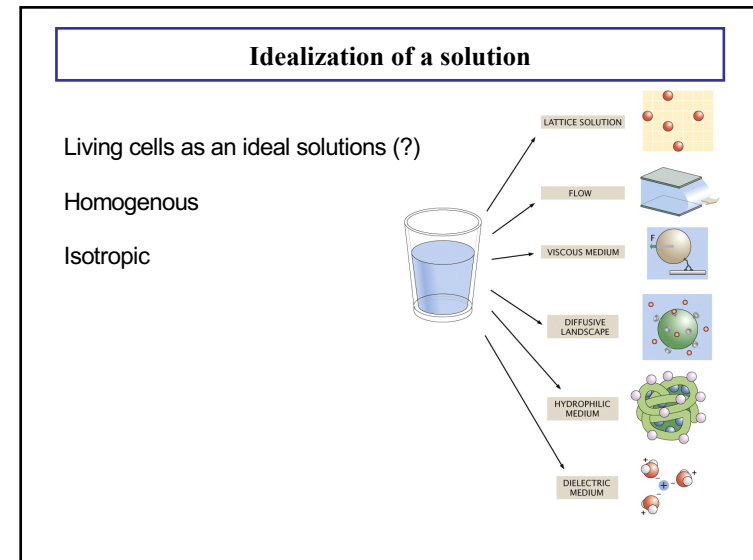
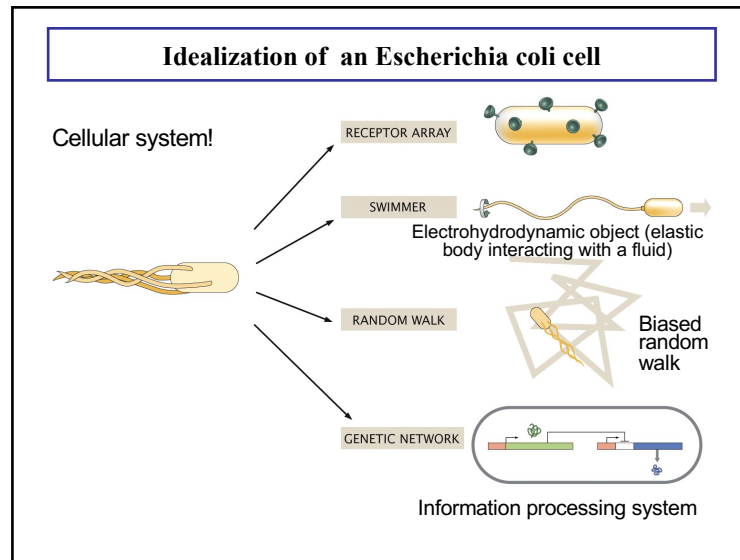


Idealization of a protein molecule

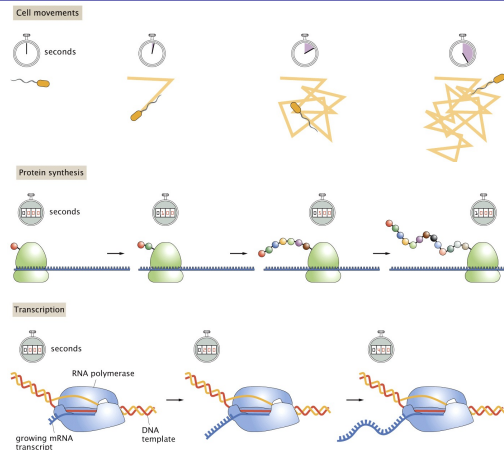


Idealization of lipids and membranes

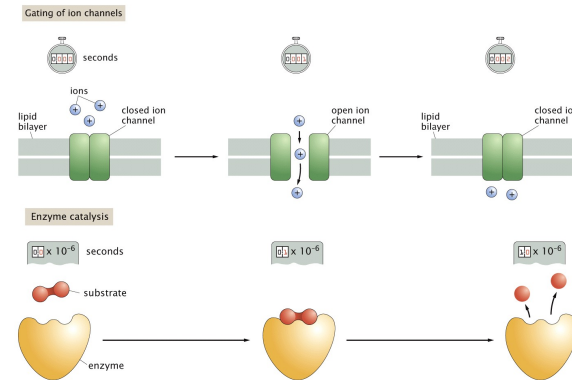




Time scales in biology II

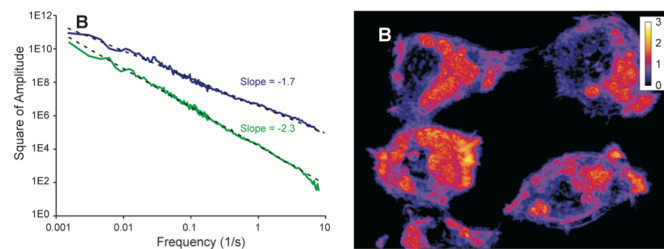


Time scales in biology III



Furthermore, light absorption 10^{-15} s!

Time scales of intracellular movements

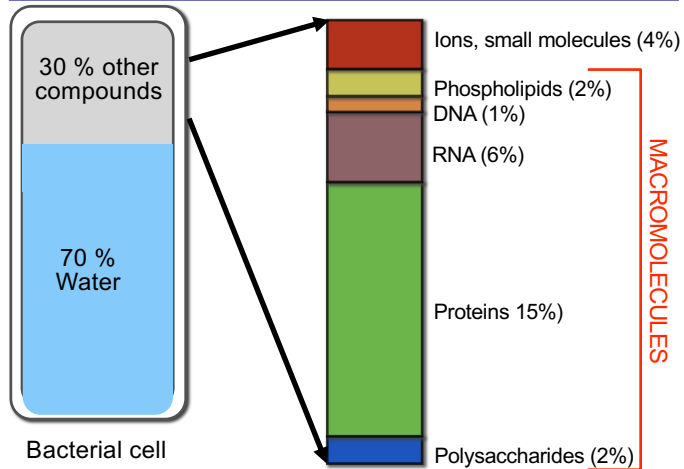


Intracellular movements occur on all time scales

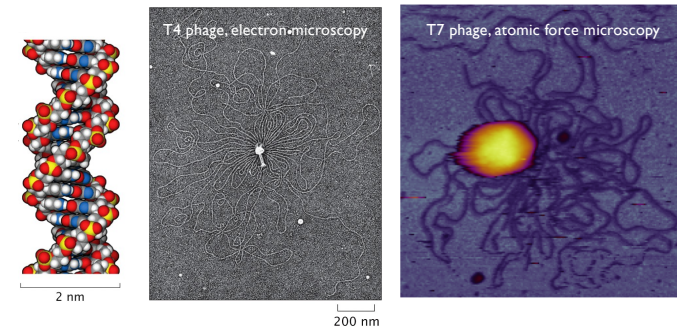
Some rules of thumb in quantitative biology

- 1 dalton (Da) = $1\text{g/mol} \approx 1.6 \times 10^{-24}\text{g}$
- 1 nM \approx 1 molecule/bacterium $\approx 10^3$ - 10^4 molecule/eukaryotic cell
- 1 M \approx $1/\text{nm}^3$
- Cellular protein concentration \approx 2-4 million/ μm^3
- 1 mg of 1 kb DNA fragment \approx 1 pmol $\approx 10^{12}$ molecules
- Mean distance between molecules at 1 M concentration \approx 1 nm
- Molecular mass of a typical amino acid \approx 100 Da
- Water concentration/density \approx 55 M \approx 1000 kg/ m^3
- Volume of a water molecule \approx 0.03nm^3
- Length of a base pair (along DNA) \approx 0.3 nm
- Volume of a base pair \approx 1nm^3

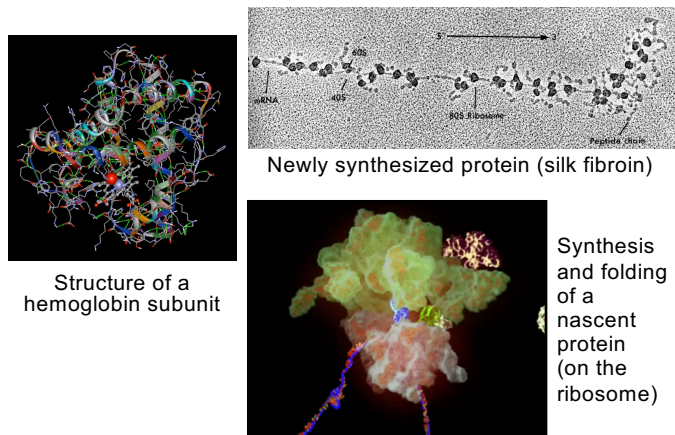
Proportion of macromolecules in the cell by mass is large



Biological macromolecules are giant molecules



Biological macromolecules are exciting molecules



Experiments can be done at single molecule level

