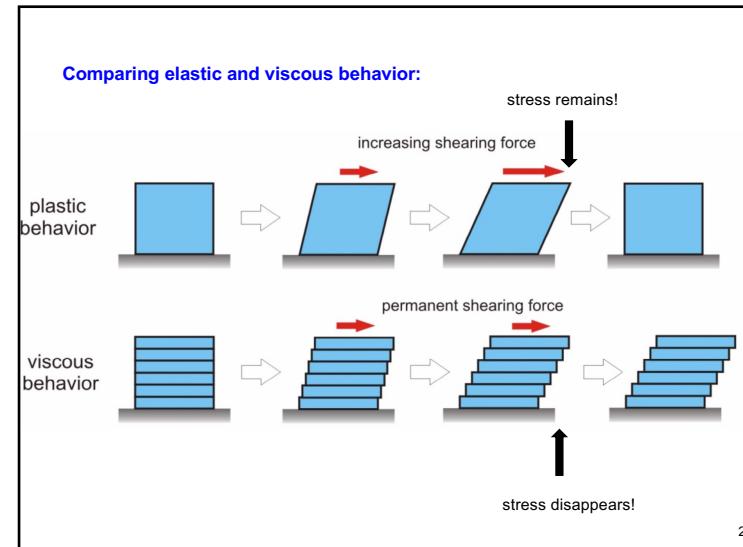
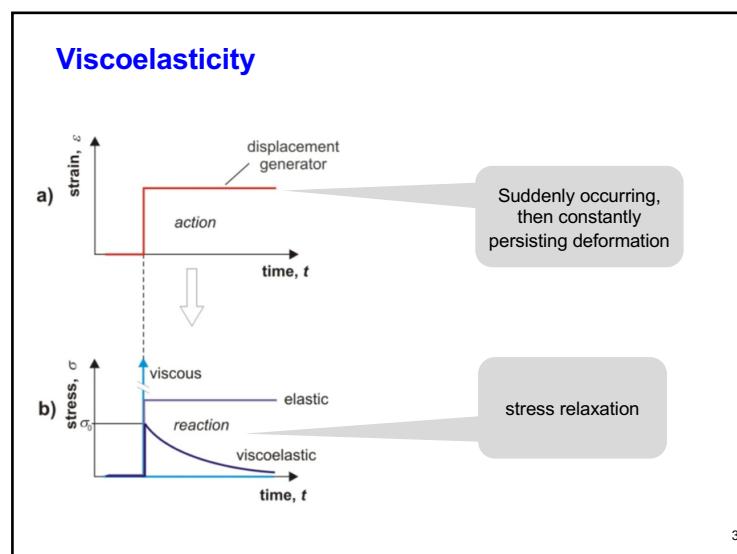


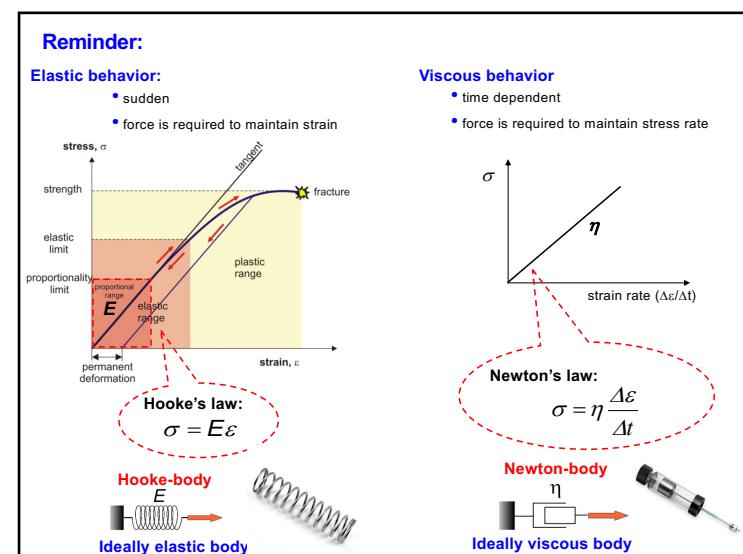
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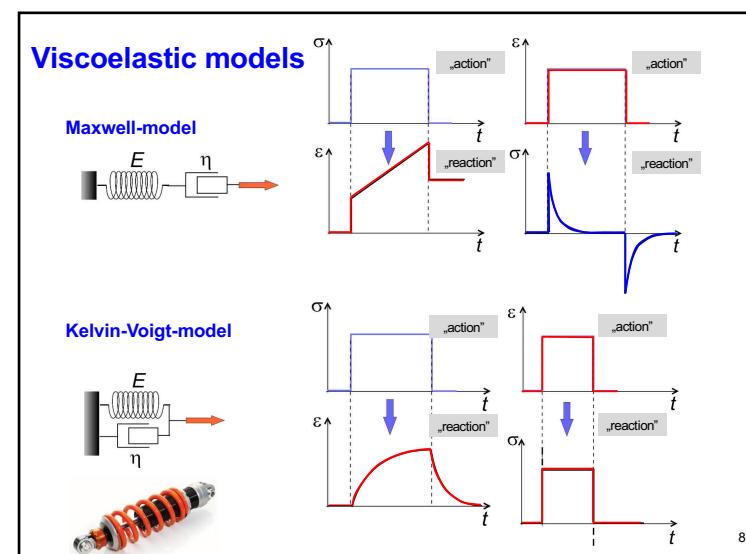
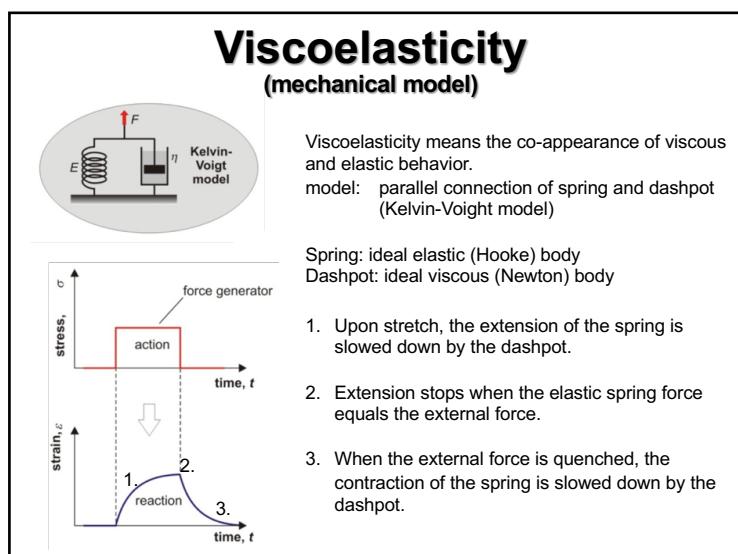
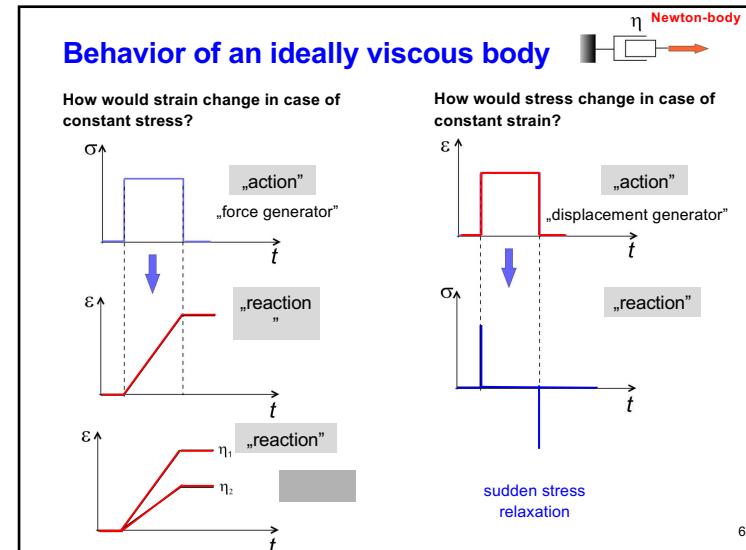
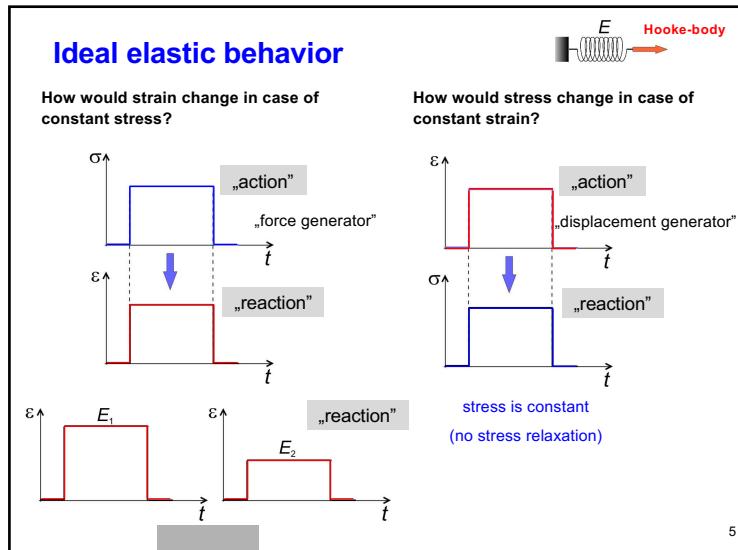
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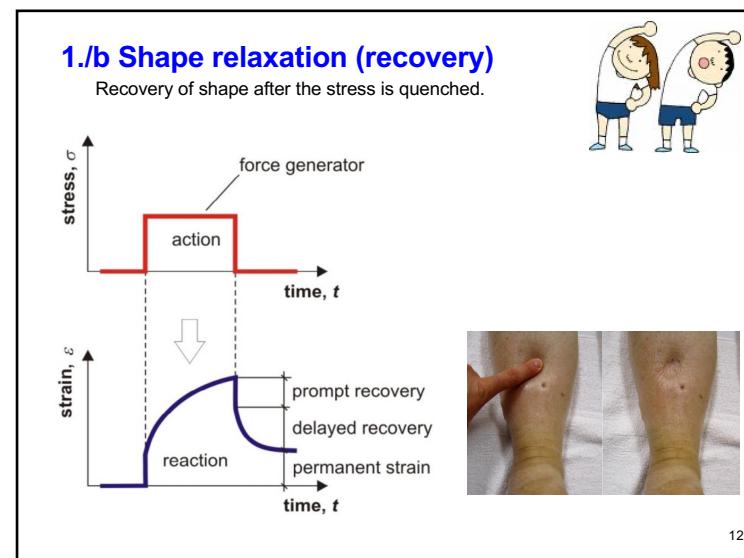
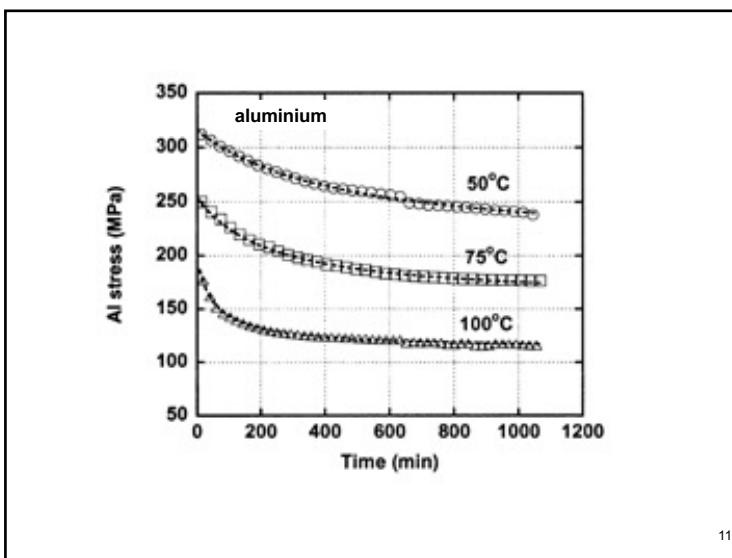
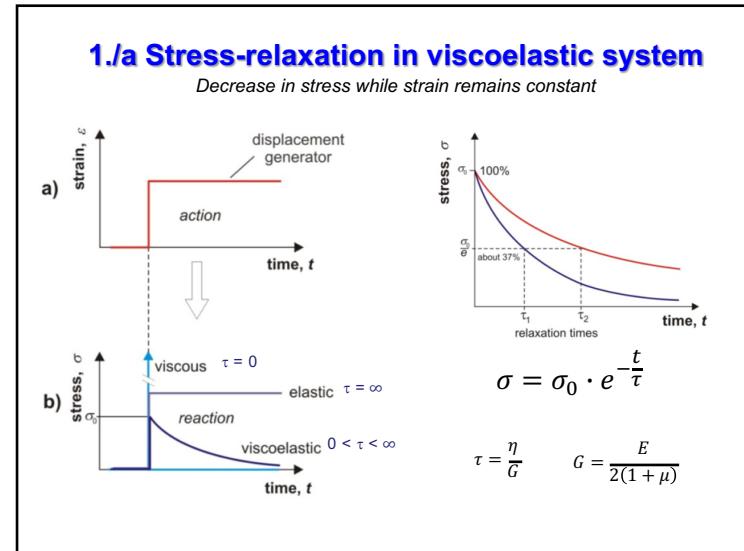
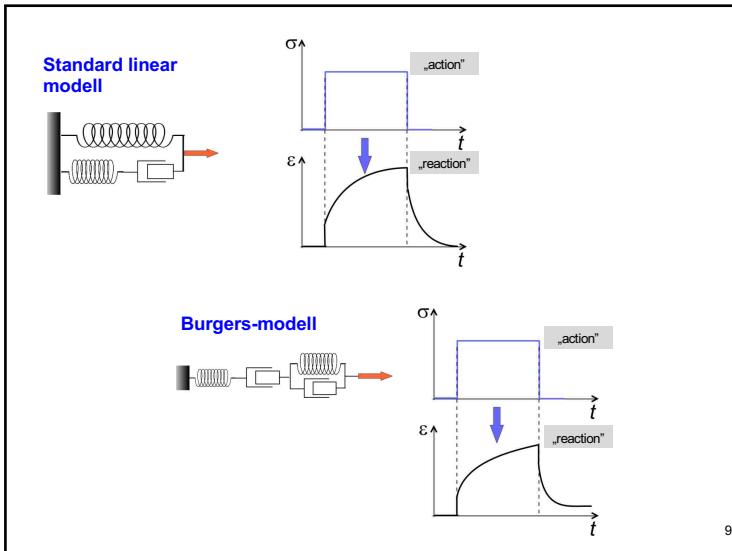


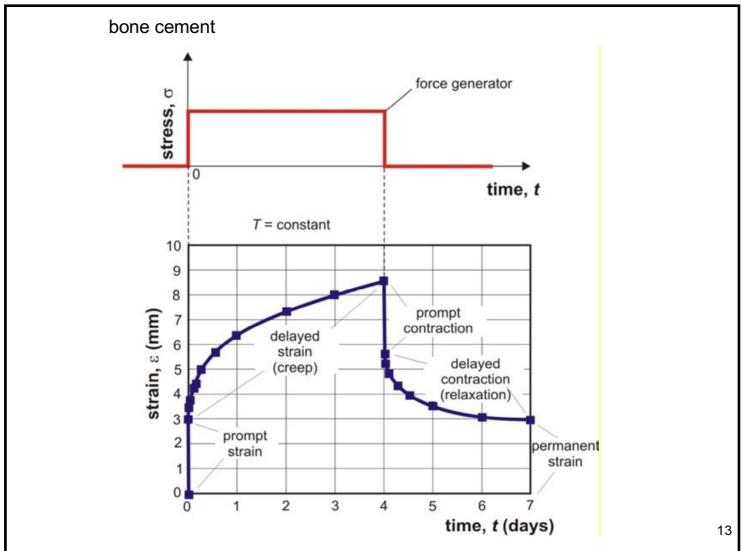
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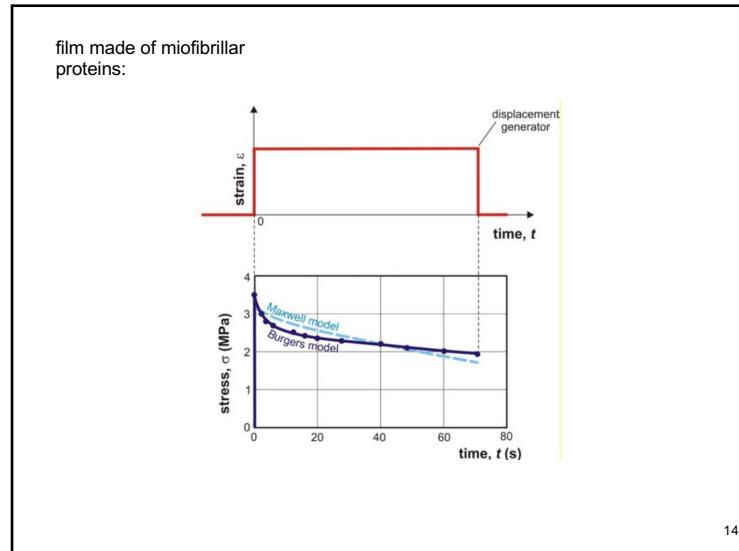
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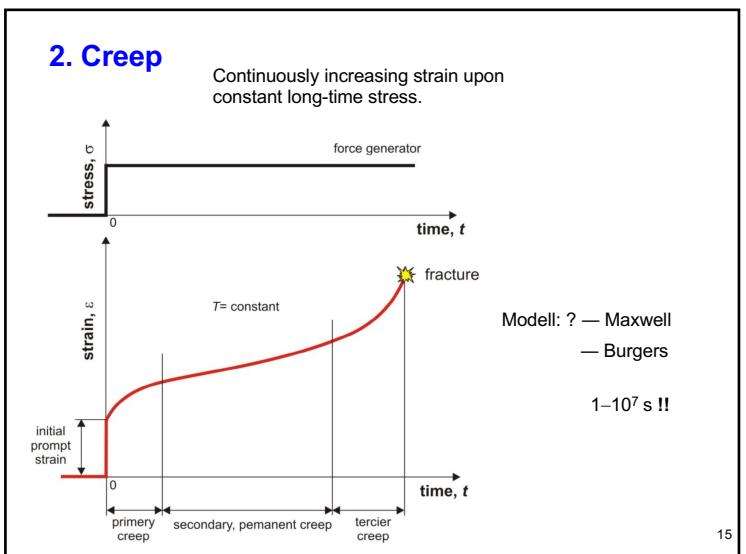




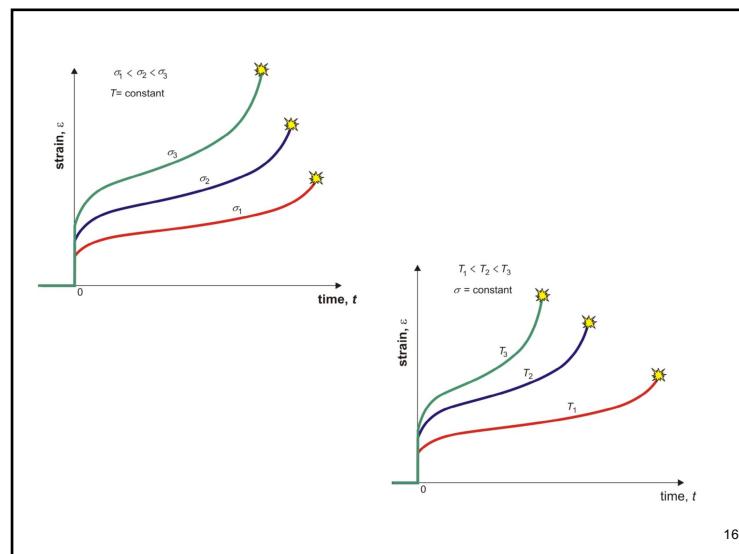
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Physical foundations of dental materials 10.

Thermal and electric properties

e-book chapter:
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Thermal properties

- temperature

- heat uptake/release



heat capacity (C):

$$C = \frac{\Delta Q}{\Delta T}$$

molar heat capacity (c_v):

$$c_v = \frac{C}{V}$$

specific heat capacity (c):

$$c = \frac{C}{m}$$

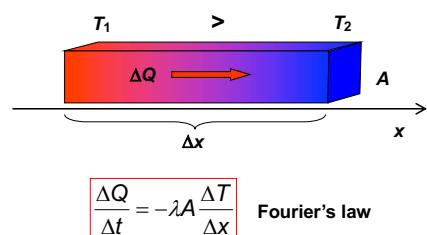

specific heat of dental materials:

material	c (J/kg·K)
enamel	750
dentine	1260
water	4190
amalgam	210
gold	126
porcelain	1100
glass	800
PMMA	1460
zinc-phosphate	500

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- heat conduction

- lattice vibrations
- free electrons



thermal conductivity of dental materials

material	λ (W/mK)
enamel	0,9
dentine	0,6
water	0,44
amalgam	23
gold	300
porcelain	1
water	0,6-1,4
acrylate	0,2
PMMA	0,2-0,3
zinc-phosphate	1,2

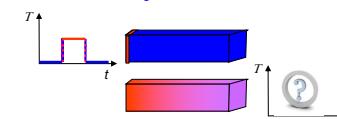
λ — thermal conductivity
 $J/(s \cdot m^2 \cdot K/m) = W/(m \cdot K)$

used in case of stationary conditions



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non-stationary conditions:

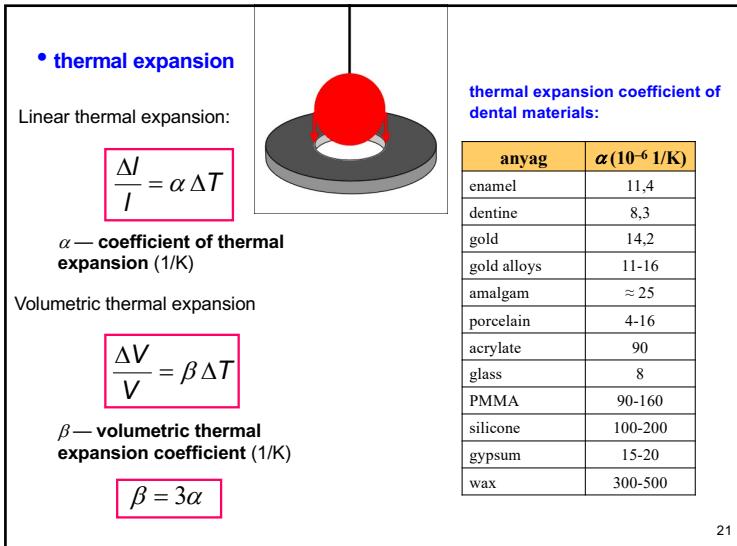


$$D = \frac{\lambda}{c \cdot \rho}$$

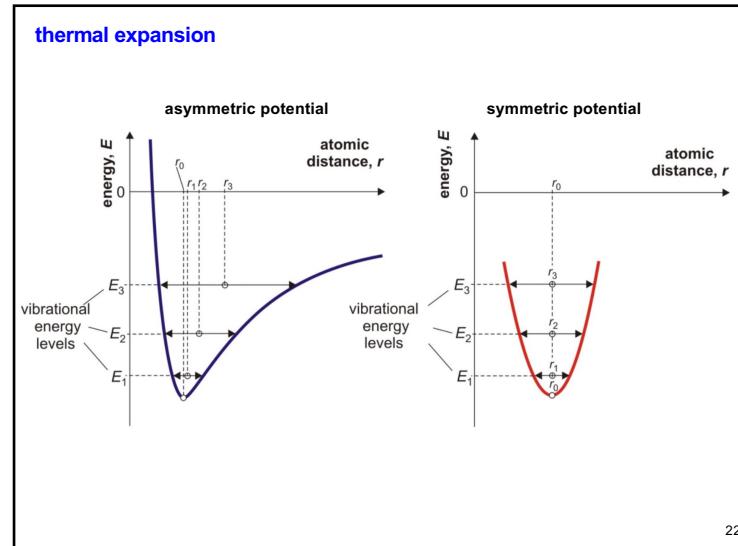
D — thermal diffusivity
(m^2/s)

material	λ (W/(mK))	D ($10^{-6} m^2/s$)
enamel	0,9	0,5
dentine	0,6	0,2
water	0,44	0,14
amalgam	23	9,6
gold	300	118
porcelain	1	0,4
water	0,6-1,4	0,3-0,7
acrylate	0,2	0,1
PMMA	0,2-0,3	0,12
zinc-phosphate	1,2	0,3

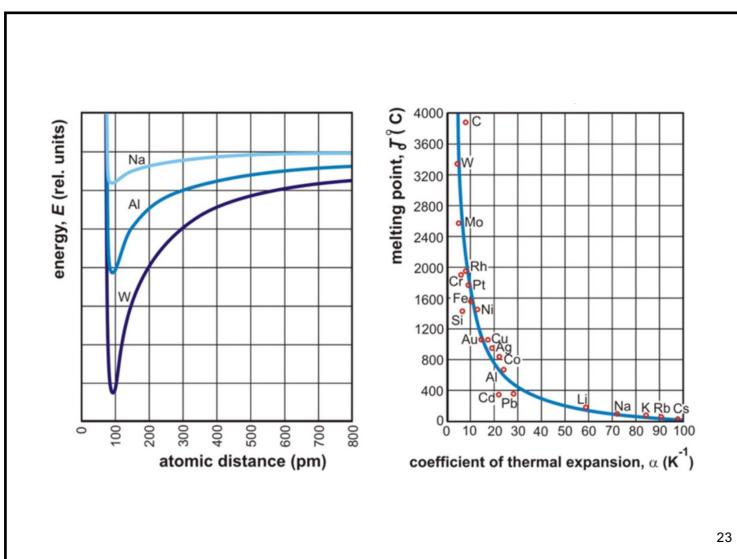
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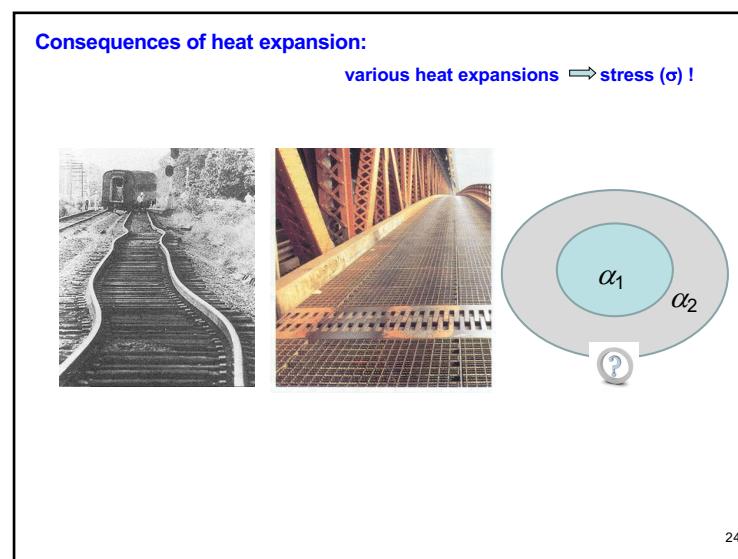
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Your feedback is highly appreciated!

Please prepare your phones to scan the QR code

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