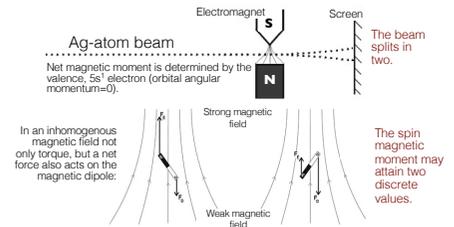


RADIO SPECTROSCOPIES, NMR, ESR, MRI

MIKLÓS KELLERMAYER

Atomic, molecular systems may behave as elementary magnets

Stern-Gerlach experiment (1922)



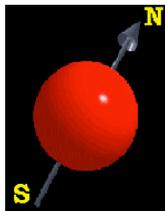
Nuclear magnetic resonance, (NMR) Nobel-prize, 1952



Magnetic resonance: resonance-absorption of electromagnetic energy by a material placed in magnetic field.

Systems with net spin: elementary magnets

- Elementary particles (p, n, e) have their own *spin*.
- Depending on the number of elementary particles and organizational principles (e.g., Pauli principle), *net spin* emerges within the system.
- Atomic nucleus: odd mass number - half nuclear spin (^1H , ^{13}C , ^{15}N , ^{19}F , ^{31}P); even mass number, odd atomic number - whole nuclear spin; even mass and atomic number - zero nuclear spin.
- Electron: net electron spin within a molecular system containing a stable unpaired electron (e.g., free radicals).
- Because of *charge* and *net spin*, *magnetic moment* emerges.



Nuclear magnetic moment:

$$M_N = \gamma_N L$$

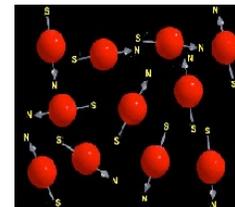
γ_N = gyromagnetic ratio (ratio of magnetic moment and angular momentum.)
 L = nuclear spin ($L = \sqrt{l(l+1)}\hbar$), l = spin quantum number.

Magnetic moment of the electron:

$$M_e = -g\mu_B \sqrt{S(S+1)}$$

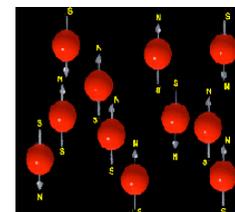
g = electron's g-factor (dimensionless number that describes the relationship between magnetic moment and gyromagnetic ratio)
 μ_B = Bohr's magneton (unit of the electron's magnetic moment)
 S = spin quantum number

In external magnetic field the elementary magnets orient

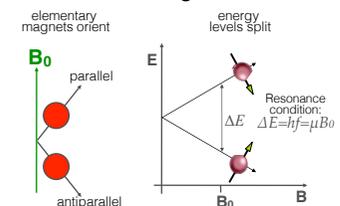


In absence of magnetic field:
random orientation of elementary magnets

Paramagnetism: magnetism emerging in external magnetic field (caused by the orientation of magnetic dipoles).



In magnetic field:

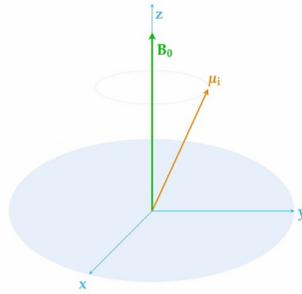


Edward Purcell, 1946

oriented elementary magnets do precessional mo



Precession of a top



Precession of an elementary magnetic moment (μ) in magnetic field (B_0) within a reference xyz space

Precession or Larmor frequency:

$$\omega_0 = \gamma B_0$$

$$f_{Larmor} = \frac{\gamma}{2\pi} B_0$$

Resonance condition:

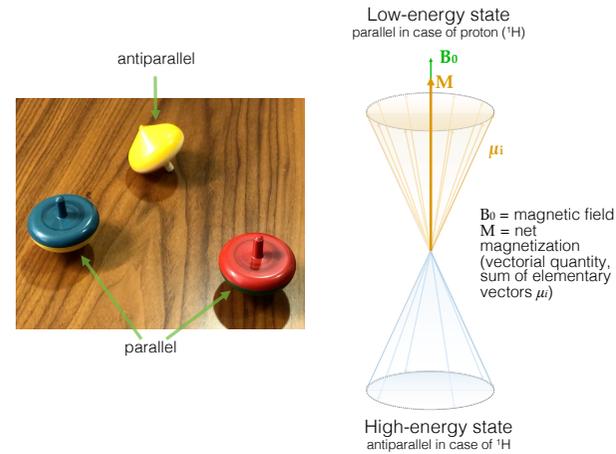
$$\Delta E = \frac{h\omega_0}{2\pi}$$



Felix Bloch, 1946

Net (macroscopic) magnetization

Due to spin access in different energy states



Ratio of low- and high-energy spin populations is determined by the Boltzmann distribution:

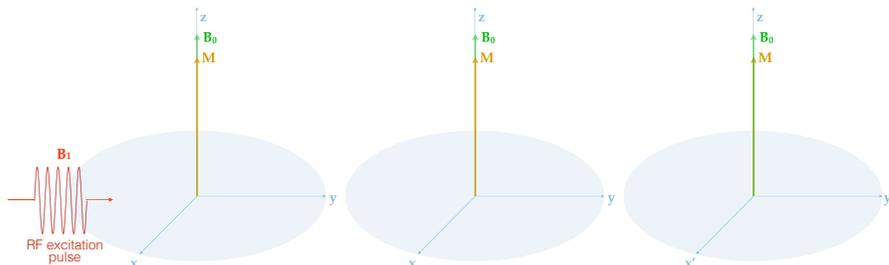
$$\frac{N_{antiparallel}}{N_{parallel}} = e^{-\frac{\Delta E}{k_B T}}$$

N.B.: magnetic field in MRI is 20-50-thousand times as strong as the earth's magnetic field.

Excitation

Resonance condition, Larmor frequency

Employed electromagnetic radiation: radiowaves (NMR, MRI), microwaves (ESR)

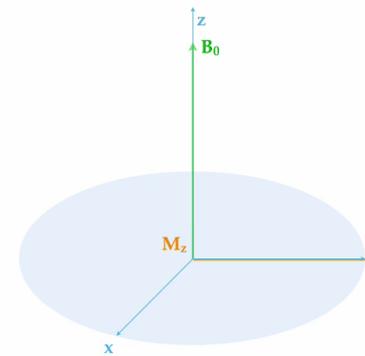


B_0 = magnetic field
 M = net magnetization
 B_1 = irradiated radio frequency wave

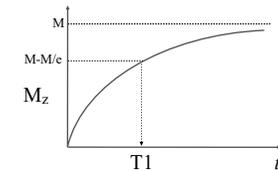
Spin-lattice relaxation

T1 or longitudinal relaxation

T1 relaxation process: return (relaxation) of the z-axis vectorial component of M (M_z) towards the direction of the external magnetic field



M_z : z-axis vectorial component of M

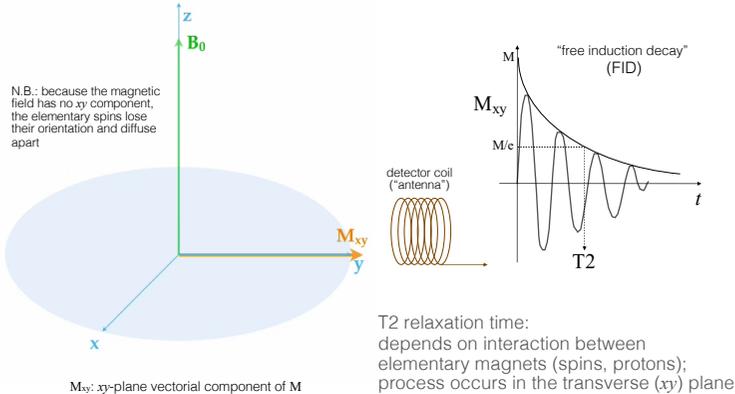


T1 relaxation time: depends on interaction between elementary magnet (proton) and its environment (lattice)

Spin-spin relaxation

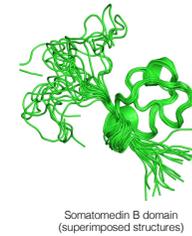
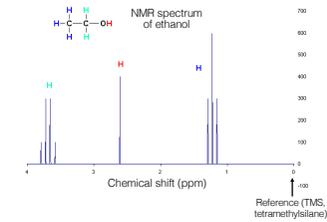
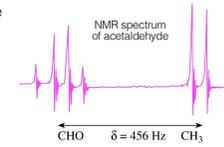
T2 or transverse relaxation

T2 relaxation process: diffusion (spreading) of the elementary magnetic moments (μ_i) resulting in the decay of the transverse (xy)-plane vectorial component of M (M_{xy})



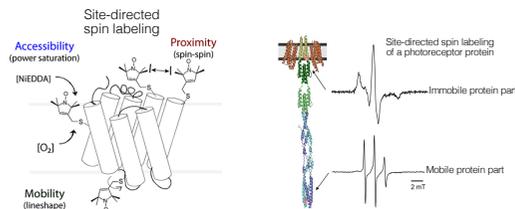
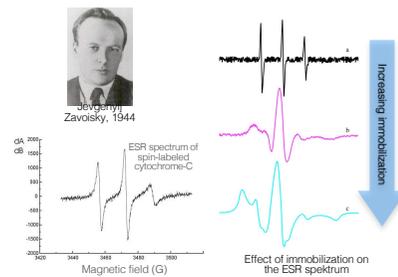
NMR spectroscopy

- NMR spectroscopy or Magnetic Resonance Spectroscopy (MRS)
- Spectroscopic method for measuring the local magnetic field around atomic nuclei. We measure the resonance frequencies of the nuclei positioned in the magnetic field.
- NMR spectrometer: superconducting magnet (cooled with liquid He), large magnetic field (spectral resolution is proportional to field strength).
- NMR spectrum: intensity of absorbed electromagnetic radiation as a function of frequency.
- The area under the "NMR-line" is proportional to the number of absorbing atomic nuclei.
- The electron cloud distorts the local magnetic field, therefore the frequency condition is shifted: "chemical shift". Chemical structure determination is possible.
- Protein NMR: possibility of measuring dynamics and the detection of disordered protein elements

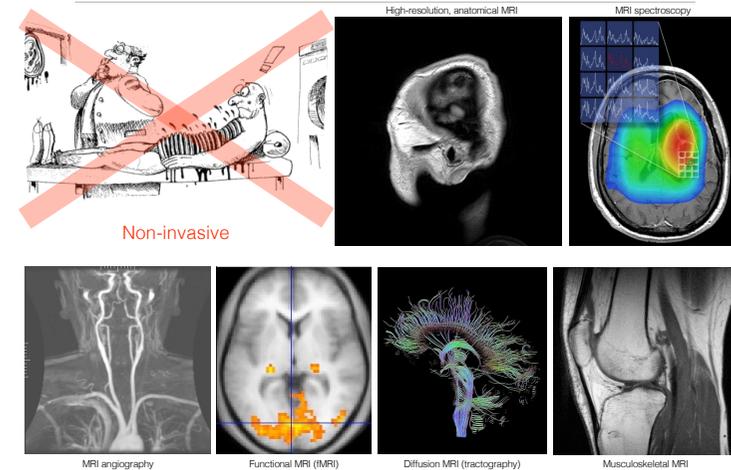


ESR spectroscopy

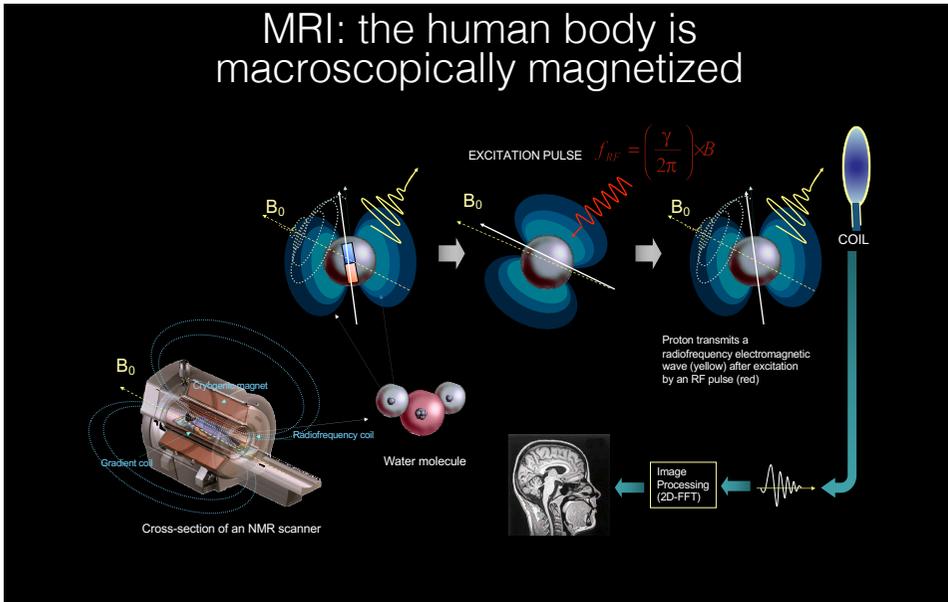
- Electron spin resonance (ESR) or electron paramagnetic resonance (EPR) spectroscopy.
- Spectroscopic method for investigating materials containing unpaired electron.
- EPR spectrum: intensity of electromagnetic radiation as a function of magnetic field.
- Magnetic field is lower, but radiation frequencies are greater (microwave) than in NMR.
- Spin-labeling: attachment of a chemical containing a stable unpaired electron. Site-directed spin labeling: spin labeling of reactive (mostly -SH) residues introduced into the targeted protein by molecular biological tools (point mutation).
- Dynamics of rotational motion can be measured up to the 10^{-4} - 10^{-2} s time range.



MRI: revolutionary device

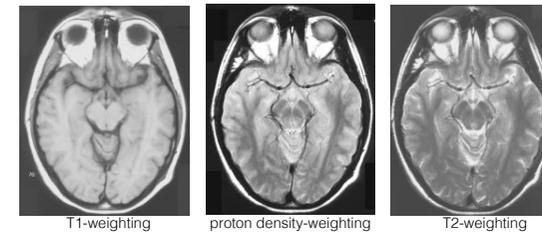
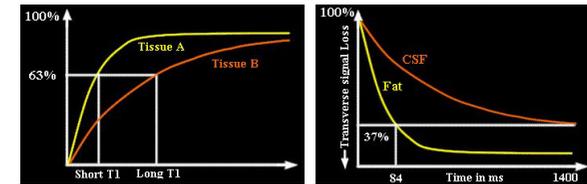


MRI: the human body is macroscopically magnetized



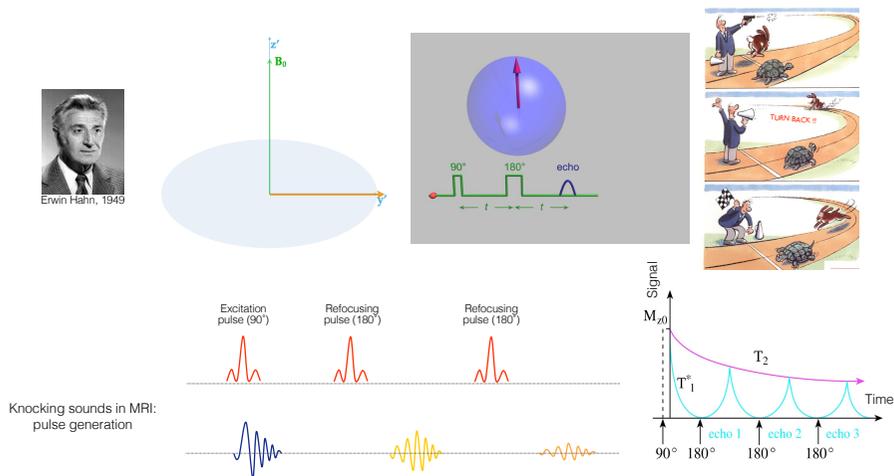
MRI 1: contrast

Color contrast based on spin density (proton density, PD) and relaxation times (T1, T2)

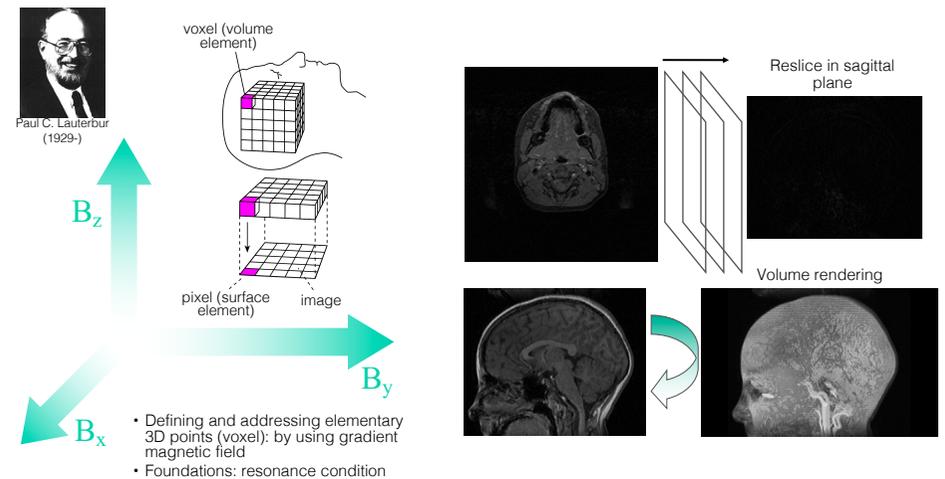


Measuring relaxation time: the spin-echo experiment

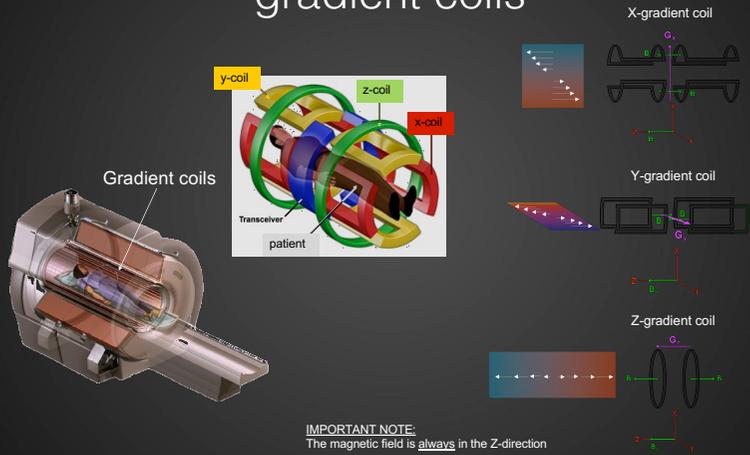
Repeating excitation, refocusing and "echo" pulses: spin-echo sequence



MRI 2: spatial encoding

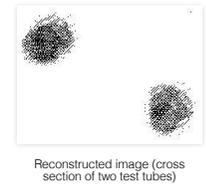
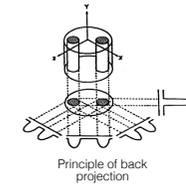


Generation of gradient fields: with gradient coils

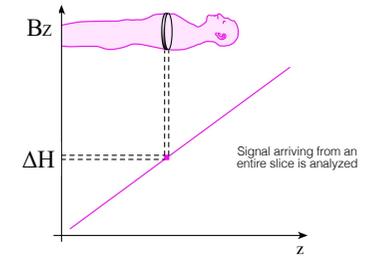


MRI 3: image reconstruction

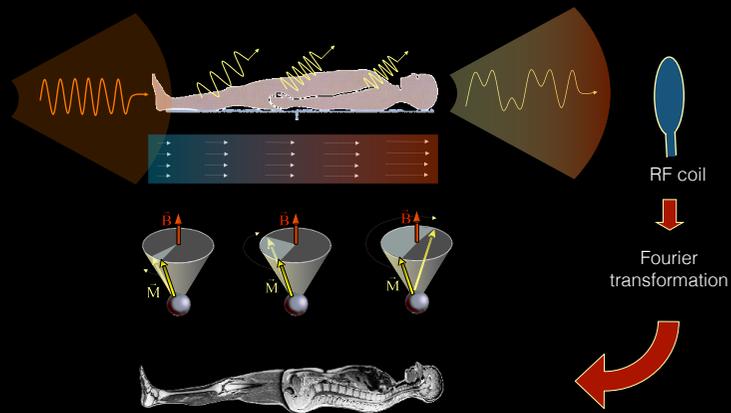
1. Back projection
 • as in CT scanning



2. 2D Fourier transformation
 • currently used method
 • „NMR Fourier Zeugmatography“

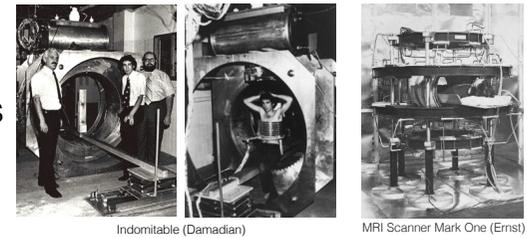


MRI: spatial coding and image reconstruction based on the resonance condition (B_0 -dependent ω)



MRI 4: scanners

Early times



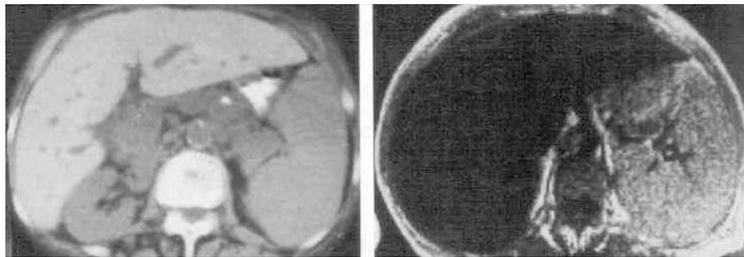
Present



MRI 5: contrast agents

Positive: paramagnetic elements (T1 contrast): Gd, Mn

Negative: superparamagnetic, ferromagnetic (T2 contrast): FeIII, MnII



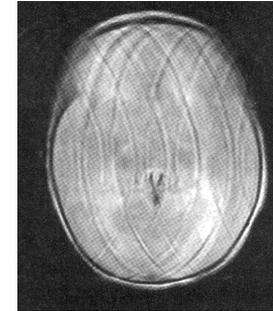
CT

MR T2

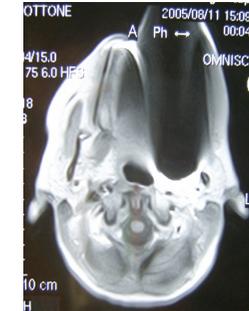
Haemochromatosis hepatis (iron accumulation in liver)

MRI 6: artefacts

- Motion
- Metals (implants, injury)



Motion artefact



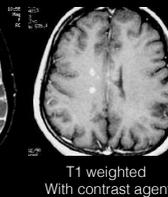
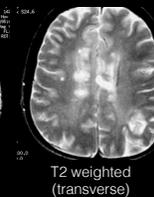
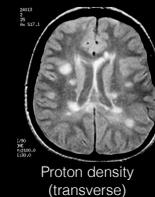
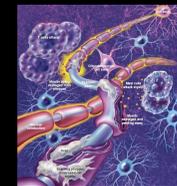
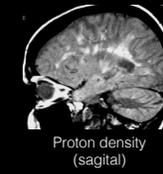
Metal in the orbit of the eye

MRI 7: dangers, contraindications

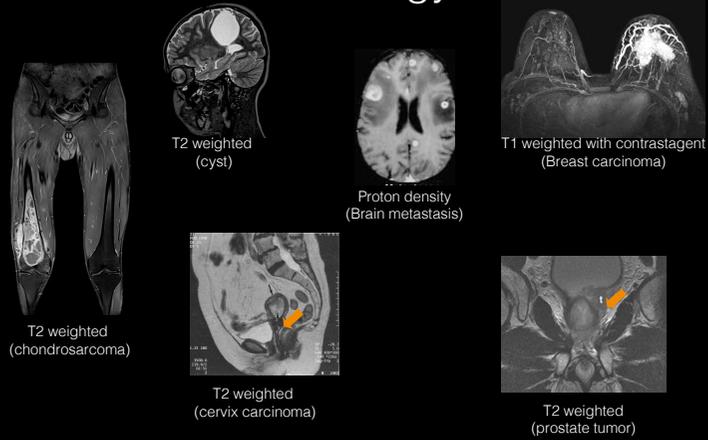
- Static magnetic field - metal objects
Contraindications: implanted devices (pacemaker, defibrillator, hearing aids, drug delivery devices), neurostimulators, brain aneurysm clamps, early cardiac valve implants
- Gradient field - induced current
- Radio frequency field - thermal effects (lens, testis)



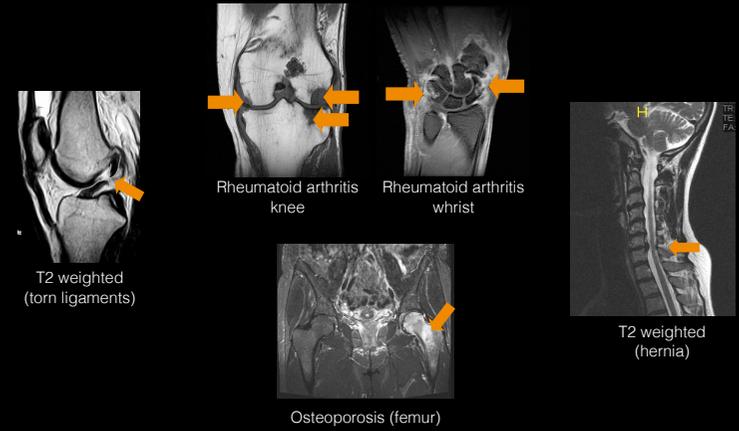
Applications: Anatomical imaging - Multiple sclerosis



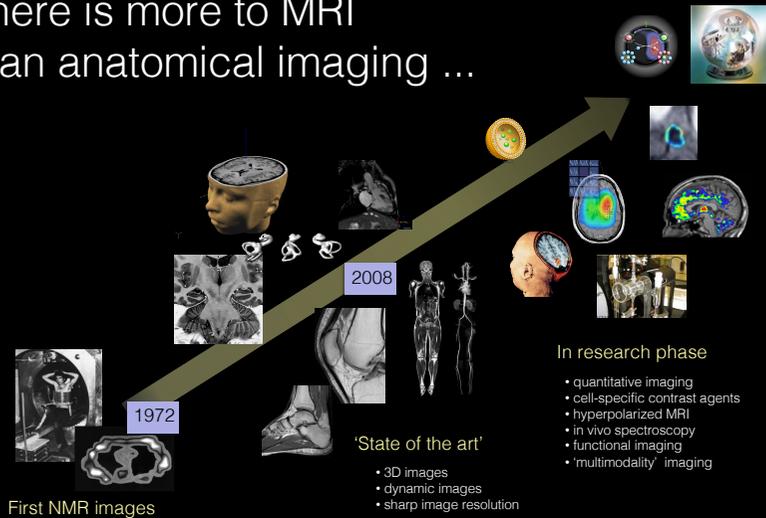
Anatomical imaging: Oncology



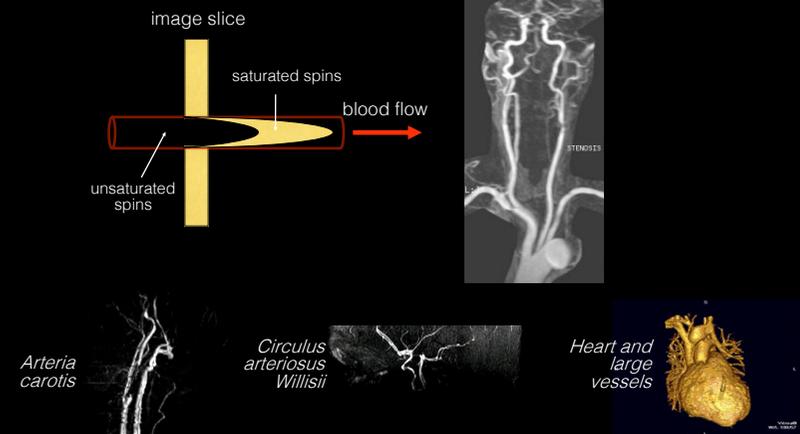
Anatomical imaging Bone and soft tissue



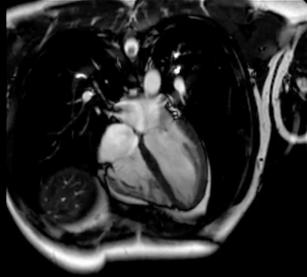
There is more to MRI than anatomical imaging ...



MRI: Non-invasive angiography



Time-resolved MRI (ECG-gating required)



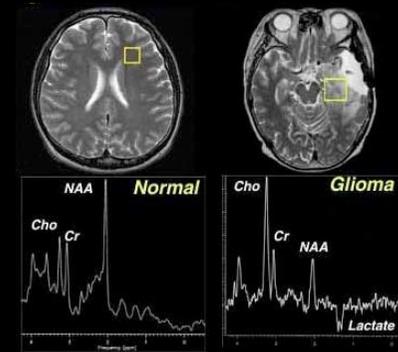
Blood flow across the cardiac chambers



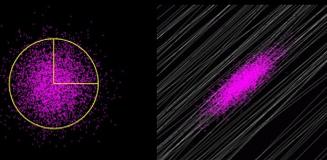
Opening and closing of aorta valve

MR Spectroscopy

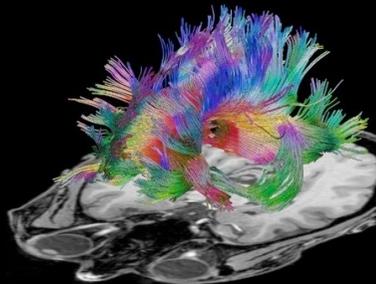
- Chemical shift
- Identification of metabolites
- Tumor diagnostics



Diffusion imaging



Anisotropic water diffusion:
contrast generation



Imaging neural tracts:
tractography

Corpus callosum

Functional MRI (fMRI)

High time resolution images recorded synchronously with
physiological processes



Activation in the
acoustic cortex



Effect of light pulses on
the visual cortex

Superposition of MRI and PET

Intracranial tumor



PET signal: cortical areas activated
during eye movement
Volume rendering

Feedback



<https://feedback.semmelweis.hu/feedback/pre-show-qr.php?type=feedback&qr=TQ3LPVV6ILE8T5PN>